

Low Cost Bipolar Plates for Large Scale PEM Electrolyzers

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Knowledge for Tomorrow

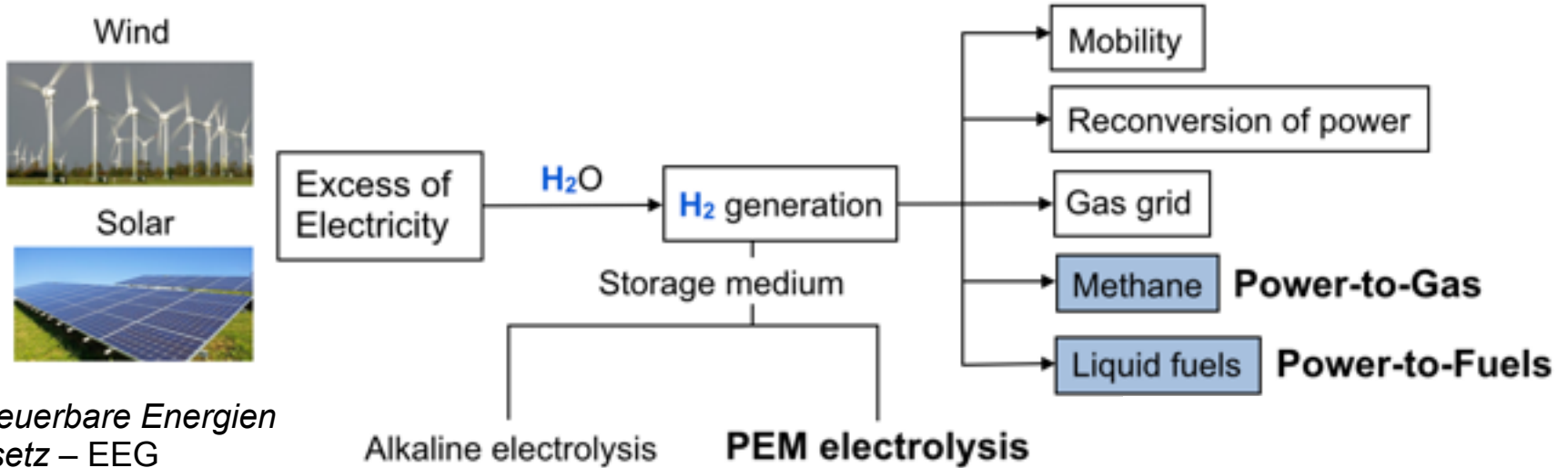


Outline

- Introduction: H₂ as energy vector
- Cost issue in PEM-electrolysis systems
- Strategy for producing the coatings for low cost bipolar plates
- Physical and electrochemical characterization of **Au/Ti** and **Pt/Ti** coatings
- Conclusions



Hydrogen as energy vector



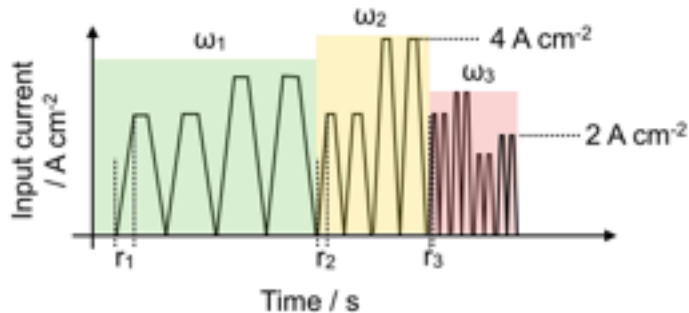
Erneuerbare Energien
Gesetz – EEG

40% - 45% by 2025

55% - 60% by 2035

80% by 2050

AST protocol based on
wind profile



- Uses only DI water
- No additional compressor is needed
- Current density > 6 A cm⁻²
- Rapid response (dynamic operation)
- Expensive

Alkaline electrolyzer = 1,394 USD/kW

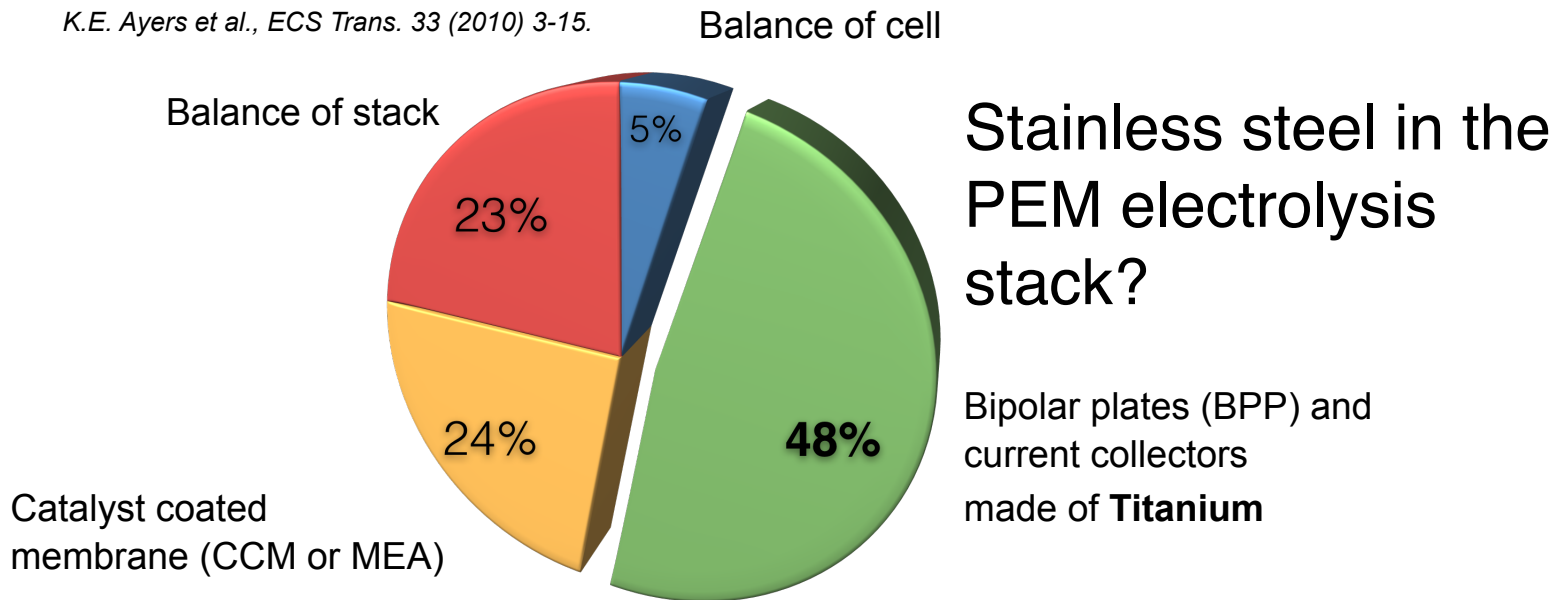
PEM electrolyzer = 2,648 USD/kW

"Study on development of water electrolysis in the EU"
Fuel Cells and Hydrogen Joint Undertaking



Cost of a PEM electrolysis stack

K.E. Ayers *et al.*, *ECS Trans.* 33 (2010) 3-15.



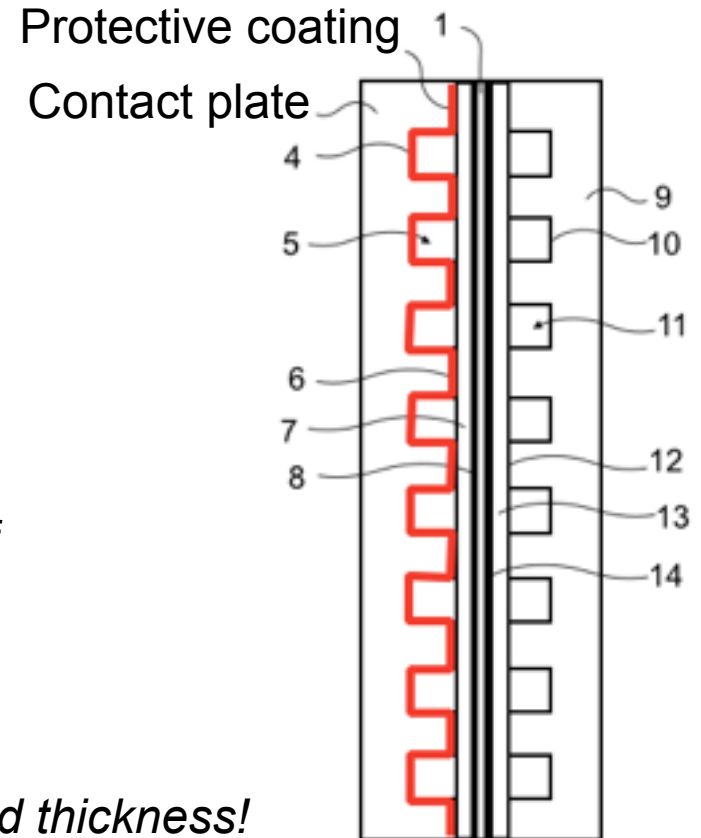
CAPEX vs. OPEX



Requirements of the protective coating

1. Resistant to corrosion
2. Low electrical resistivity / High electrical conductivity
3. Strong adherence to the substrate
4. Low cost material, facile and scalable deposition technique
5. Withstand mechanical loads during operation
6. Minimal differences in the coefficient of thermal expansion
7. Resistant to H₂ embrittlement (cathode side)

No limitations in weight and thickness!



Strategy for coating stainless steel

Coatings for PEM fuel cells:



Coating 1

Coating 2

Corrosion resistant and conductive coating on stainless steel

Pitting corrosion in PVD coatings



The released Fe^{2+} poison the CCM of the electrolyzer

S. Sun et al., J. Power Sources 267 (2014) 515.

Coatings for PEM electrolyzers:



Coating 1: Titanium coating by vacuum plasma spraying (VPS)

Coating 2: Surface modification of the Ti coating by electrodeposition or PVD magnetron sputtering

Cost reduction?

PEMFC: PVD coatings alone meet the DOE requirements.

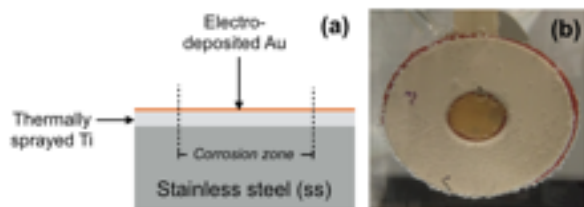
PEM electrolyzers: current Ti bipolar plates have to be coated to reduce the passivation.

Dramatic cost reduction for **large area** bipolar plates \longrightarrow **Megawatt scale**

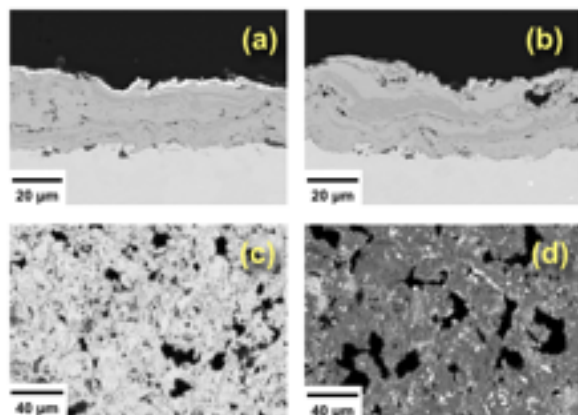


Physical characterization of Au/Ti coating

Optical and SEM images



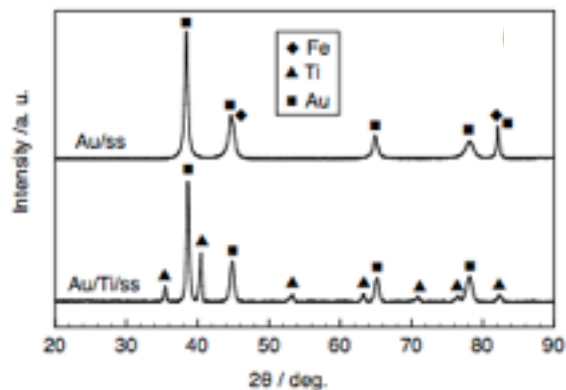
Thermally sprayed Ti coatings with a plasma enthalpy of 21.3 MJ kg⁻¹ electroplated with Au afterwards



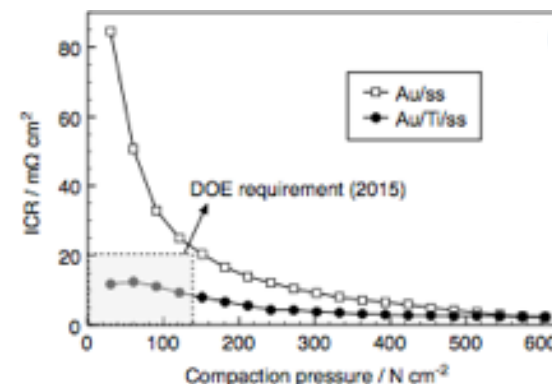
Before

After

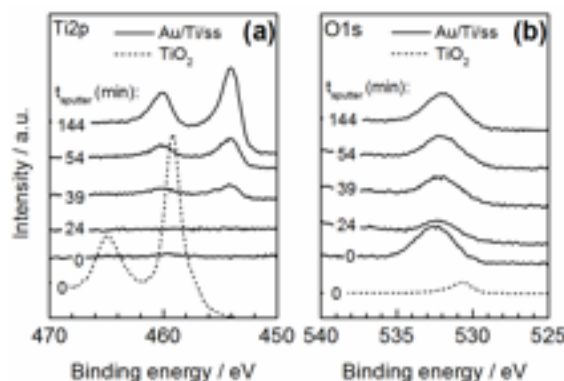
XRD



ICR



XPS



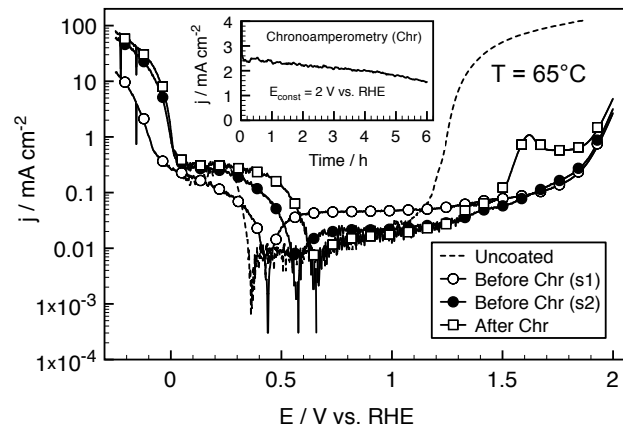
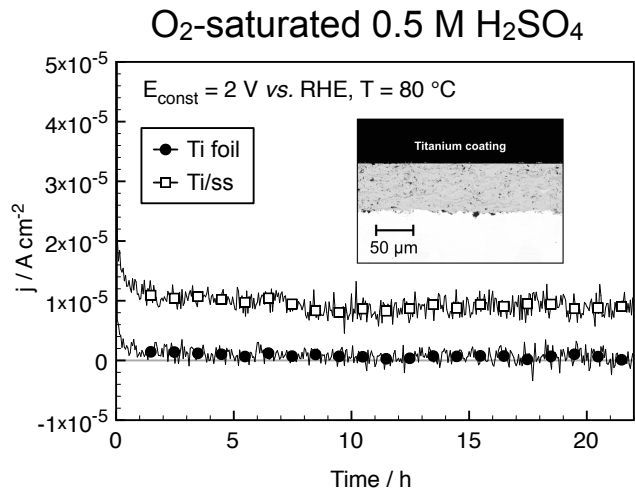
Au/Ti coating

- Deposited by successive steps of VPS and electrodeposition
- It has a roughness factor of 0.62. This property allows the adherence of Au on Ti.
- The presence of TiO₂ between Ti and Au is below the detection limit.
- Meets the DOE requirements of IRC.



Electrochemical characterization of Au/Ti coating

Corrosion tests (half-cell)

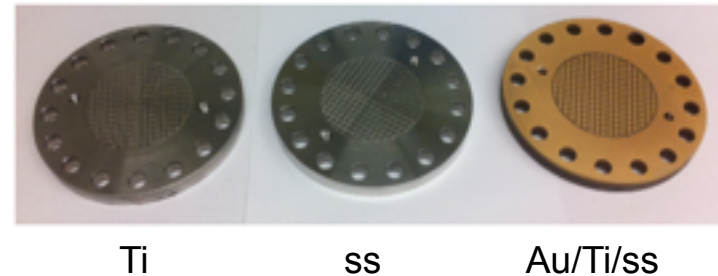


Electrochemical parameters

$$R_p = \frac{\beta_a \beta_c}{2.3 j_{\text{corr}} (\beta_a + \beta_c)}$$

Corrosion potential (E_{corr}); corrosion current (i_{corr}); anode (β_a) and cathode (β_c) Tafel slopes and polarization resistance (R_p) of the coated stainless steel before (scan 1 and 2) and after the chronoamperometric (chr) test.

Au/Ti/ss	$E_{\text{corr}} / \text{V vs. RHE}$	$j_{\text{corr}} / \mu\text{A cm}^{-2}$	$\beta_a / \text{mV dec}^{-1}$	$\beta_c / \text{mV dec}^{-1}$	$R_p / \times 10^3 \Omega$	$[\text{Fe}^{2+}] / \text{ppm}$
Before chr (s1)	0,44	4,76	63	91	3,40	–
Before chr (s2)	0,57	1,52	36	35	5,08	–
After chr	0,66	3,82	148	61	4,92	0



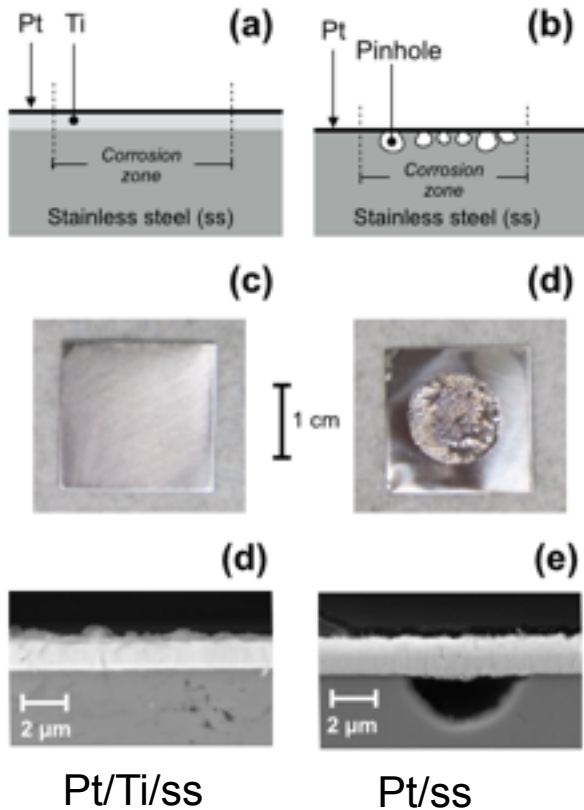
Au/Ti coating

- The i_{corr} of the coated sample is 270 times lower than the uncoated one.
- It improves the contact between GDL / CC and the BPP.
- The Au modification detached after 6 h of being constantly polarized at 2 V vs. RHE.

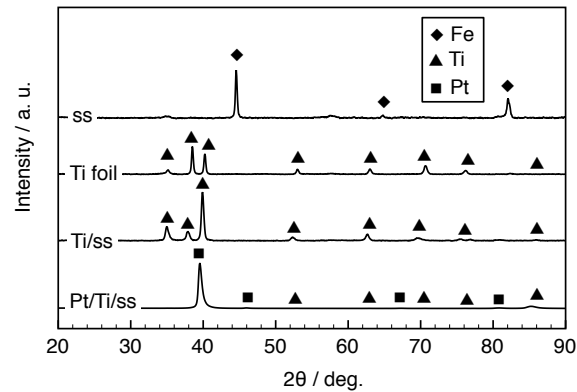


Physical characterization of Pt/Ti coating (< 4 wt% PMG)

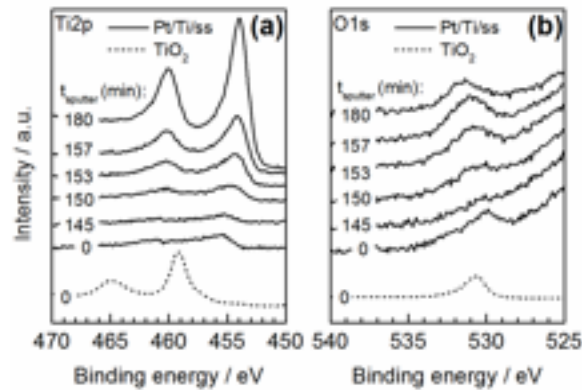
Optical and SEM images after corrosion measurements



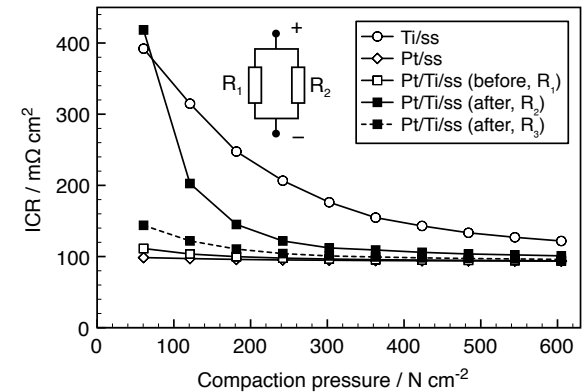
XRD



XPS



ICR



Pt/Ti coating

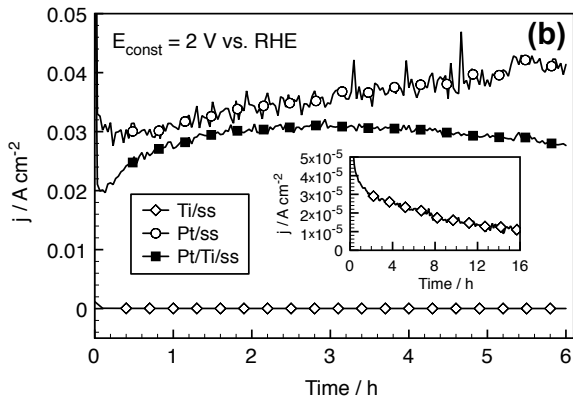
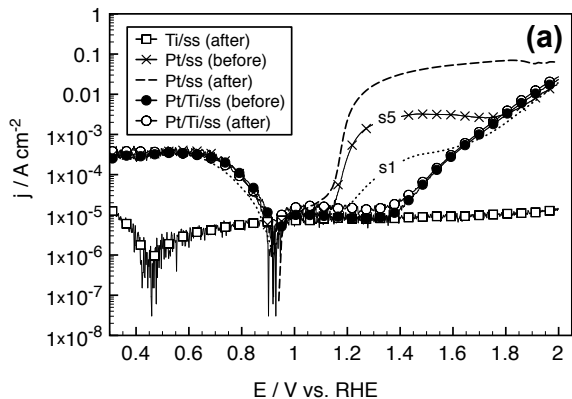
- Deposited by successive steps of VPS and PVD magnetron sputtering techniques.
- The presence of TiO₂ between Ti and Pt could not be detected
- The IRC @ 240 N cm⁻² of Pt/Ti coating increased 6% after corrosion measurements.



Electrochemical characterization of Pt/Ti coating

Corrosion tests (half-cell)

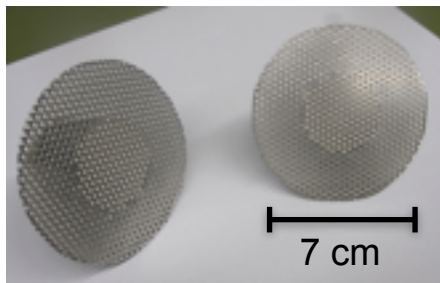
O₂-saturated 0.5 M H₂SO₄, T = 65°C



Electrochemical parameters

Sample	E _{corr} / V vs. RHE		i _{corr} / μA cm ⁻²		β _a / mV dec ⁻¹		β _c / mV dec ⁻¹		R _p / x 10 ³ Ω		[Fe ²⁺] / ppm
	Before	After	Before	After	Before	After	Before	After	Before	After	After
Stainless steel (ss)	0.36	0.54	1,7	0,6	50	56	20	34	3,7	15,3	100.8
Ti/ss	0.22	0.46	0,71	0,44	99	102	92	73	29,2	42,0	0
Pt/ss	0.91	0.94	1,22	0,66	53	17	19	21	5,0	6,2	27.73
Pt/Ti/ss	0.93	0.92	0,73	0,56	29	13	22	15	7,5	5,4	0

Contact plates



Long term tests in real conditions are currently being carried out by the industry

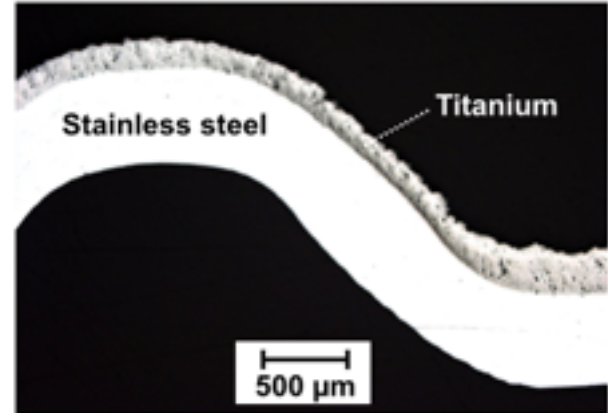
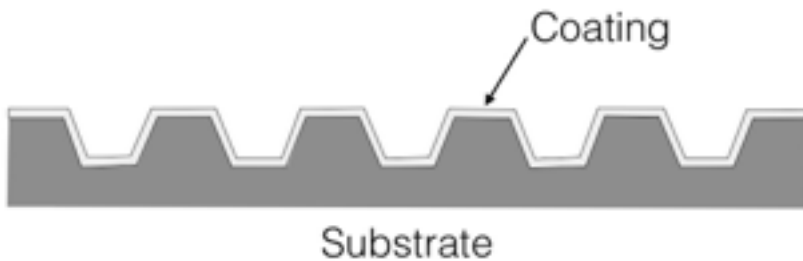
Pt/Ti coating

- It successfully protected the stainless steel substrate from corrosion and remained highly conductive at the end of the electrochemical experiments.
- In contrast the Pt coating on ss failed dramatically and the Fe²⁺ and Cr³⁺ heavily polluted the electrolyte as result of corrosion of the substrate.



Cost of the thermally sprayed Ti coating

Cost of coating BPPs for 1 MW PEM electrolyzer:
(2 V @ 2 A cm⁻²)



250 x



1000 cm²

= **2.9 USD** / bipolar plate
(considering only consumables, feedstock powder and working hours)

Cost of possible stack base materials and their respective electrical conductivities

Metal	Cost (30 x 30 x 0,3 cm ³) / USD ¹	Electrical conductivity (S/m) at 20 °C
Titanium (99.6+)	614	2.38 × 10 ⁶
Fe/Cr18/Ni10 (ss 304)	268	1.45 × 10 ⁶
Copper (99.9%)	242	5.96 × 10 ⁷
Aluminium (99%)	161	3.5 × 10 ⁷

¹ <http://www.goodfellow.com/catalogue/GFCatalogue.php?Language=E>



Conclusions

- There is an urgent need to reduce the cost of PEM electrolyzers for large scale storage of surplus electricity. The bipolar plates are very expensive.
- Dense Ti coatings were produced by vacuum plasma spraying (with high enthalpy) on stainless steel substrates.
- The coatings were evaluated in simulated conditions: O₂-saturated 0.5 M H₂SO₄, 65 - 80 °C, $E_{\text{const}} = 2 \text{ V}$.
- Further surface modification resulted in low ICR and full protection of the substrate over extended periods of time.
- The cost of coating stainless steel bipolar plates of 1000 cm² with Ti, for the Megawatt range, is estimated in 2.9 USD per piece.
- Cheaper materials as Cu and Al can be used as substrate. The developed coatings are currently being tested in commercial electrolyzers.



Acknowledgments

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