Analytical Investigation of Fuel Cells by Using In-situ and Ex-situ Diagnostic Methods

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Outline

✓ Introduction

- ✓ Spatially Resolved Measuring Technique for PEFC/DMFC
- ✓ Investigation of PEM Fuel Cells With Combined In-situ and Ex-situ Methods
- ✓ Spatially Resolved Measuring Technique for SOFC
- → Conclusion





Motivation for Using Diagnostic Tools

- > To understand the processes in fuel cells in great detail
 - Influence of the operating conditions on the conditions inside the fuel cell
 - Understanding of degradation processes
 - Inhibition of degradation processes
- > To support the development and improvement of fuel cells
 - Detection of problems of components
 - Targeted development
- ► To control fuel cell systems
 - Optimization of operating conditions
 - Increase of lifetime
 - Early error detection, increased reliability



Investigation of Degradation Processes

Physical methods (ex-situ diagnostics)

- Scanning Electron Microscopy (SEM/EDX)
- X-Ray Photoelectron Spectroscopy (XPS)
- X-Ray Diffraction (XRD)
- Porosimetry with mercury intrusion & nitrogen adsorption
- Temperature Programmed Desorption / Reduction / Oxidation (TPD / TPR / TPO)

Electrochemical methods (in-situ diagnostics)

- ➢ I-V characteristics
- Electrochemical Impedance Spectroscopy (EIS)
- Locally-resolved measurements (e.g. current density distribution)





Segmented Cells for Locally Resolved Diagnostics

Current density, Temperature, Impedance Spectroscopy







Current Density Distribution at Reduced Humidification



Resolution: 1 colour step = 10.0 % of average current density (445mA/cm²)



Integration of Segmented Bipolar Plates in a Shortstack



Sensor - Control System





PEFC Long-Term Experiment





Change of Cell Voltage During Constant Load at 500 mA cm⁻²



Elapsed time / h





Degradation in PEFC

X-ray photoelectron spectroscopy



- Partial decomposition of PTFE identified by X-ray photoelectron spectroscopy
- PTFE decomposition mainly on the anode
- \rightarrow Decrease of hydrophobicity
- \rightarrow Changed water balance
- → Reversible loss of electrochemical performance



Investigation of Degradation Processes Platinum catalyst



Reaction layers in MEAs, prepared by the DLR dry spraying technique

Top: new electrode

Middle: **anode** reaction layer of an used MEA, normal operation

Bottom: **cathode** reaction layer of an used MEA, normal operation

Left side: 20.000 fold magnification Right side: 100.000 fold magnification



Degradation in PEFC

Quantification of voltage losses during degradation by EIS





Motivation

- ✓ Strong local variation of gas composition, temperature, current density
- Distribution of electrical and chemical potential dependent on local concentrations of reactants and products
 - ✓ Reduced efficiency
 - ✓ Temperature gradients
 - ✓ Thermo mechanical stress
 - ✓ Degradation of electrodes







Measurement Setup for Segmented Cells





- Local and global i-V characteristics
- ✓ Local and global impedance measurements
- ✓ Local temperature measurements
- Local fuel concentrations
- Flexible design: substrate-, anode-, and electrolyte-supported cells
- ✓ Co- and counter-flow



Segmented Cells

>Anode supported cells with segmented cathode (H.C.Starck/InDEC)

Electrolyte supported cells: segmented cathode and anode









OCV Voltage Measurement for Determination of Humidity

- Voltage distribution at standard flow rates:
- 48.5% H_2 , 48.5% N_2 + 3% H_2O , 0.08 SlpM/cm² air





Nernst equation:

$$U_{rev} = U_{rev}^{0} - \frac{RT}{zF} \ln \left(\frac{p_{H2O}}{\sqrt{p_{O2}} p_{H2}} \right)$$

Produced water: S4: 0.61%, S8: 0.72%, S12: 0.78%, S16: 3.30%



Locally Resolved I-V Characteristics at High Fuel Utilisation

Global I-V characteristics:



50% H₂, 50% H₂O; fu_{max}= 60% 50% O₂, 50% N₂

Local I-V characteristics:







Variation of Load - Reformate





Reformate: Changes of the Gas Composition at 0 mA/cm²





Alteration of the gas composition at 435 mA/cm²





Combined Experimental and Modeling Approach

Objectives of the study:

- → Better understanding of the local variations

 - → Optimisation of cell components



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Electrochemical model of local distributions



Potential for Optical Spectroscopies

a) In situ microscopy b) In situ Raman laser diagnostics **Digital CCD camera** Imaging Distance microscope spectrograph, (resolution1 µm) Heat & radiation shield Lenses/filter Quarz window Transparent 15 cm flow field H SOFC Pulsed Nd:YAG laser Open tube (532 nm, 10 ns) (5 mm)

- ✓ Raman spectroscopy
- Laser Doppler Anemometry (LDA)
- → Particle Image Velocimetry (PIV)
- → Fast-Fourier Infrared (FTIR)
- Coherent Anti-Stokes Raman Spectroscopy (CARS)
 - Electronic Speckle Pattern Interferometry (ESPI)



X-Ray Tomography (CT) Facility at DLR



X-Ray CT Facility v|tome|x L450 at DLR Stuttgart

3 dimensional non intrusive imaging of SOFC cassette



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Summary

- ✓ The measuring techniques and exemplary results proved the possibility of
 - identifying different degradation mechanisms in fuel cells
 - optimizing operating conditions in fuel cells
 - establishing control strategies for reliable operation of fuel cells
- Combined in-situ and ex-situ diagnostic methods can provide additional important information:
 - Spatially resolved measurements to obtain local distribution of cell properties (current, voltage, impedance, gas composition, temperature)
 - Application of combined in-situ and ex-situ analytical methods
 - Combined analytical and modeling approach
- In order to overcome the remaining challenges in fuel cell development, i.e. performance, durability, reliability and costs, it is of paramount importance to improve the understanding of fuel cell operating mechanisms by applying sophisticated analytical methodology

