

**THE SEARCH FOR HYDROGEN PEROXIDE ON ENCELADUS.** S.F. Newman<sup>1</sup>, B.J. Buratti<sup>1</sup>, R.H. Brown<sup>2</sup>, R. Jaumann<sup>3</sup>, J. Bauer<sup>1</sup>, T. Momary<sup>1</sup>, <sup>1</sup>Jet Propulsion Laboratory, California Institute of Technology, 4800 Oak Grove Drive, Mail Stop 183-501, Pasadena, CA 91109. <sup>2</sup>Lunar and Planetary Laboratory, University of Arizona, Tucson, USA. <sup>3</sup>German Aerospace Center (DLR), Berlin, Germany. [Sarah.F.Newman@jpl.nasa.gov]

**Introduction:** The discovery of a dynamic, active plume emanating from the south pole of Enceladus has left scientists with many more questions than it has answered. What is causing the hot spot underneath the south pole? How are the water vapor particles formed? How does Saturn's magnetosphere interact with these particles? What is the nature of the surface and subsurface environments of the south pole? One way to explore these questions is to study the surface ice with photometric and spectral data. Using observations from the Visual and Infrared Mapping Spectrometer (VIMS) aboard the spacecraft Cassini, we have searched for the presence of H<sub>2</sub>O<sub>2</sub> on Enceladus. Our preliminary results indicate a tentative detection of H<sub>2</sub>O<sub>2</sub> in a condensed form. The presence or absence of H<sub>2</sub>O<sub>2</sub> is indicative of the radiolytic environment of a surface or the surrounding atmosphere. In addition the depth and position of H<sub>2</sub>O<sub>2</sub> bands are useful for determining grain size, porosity and thermal history [1].

**Production of H<sub>2</sub>O<sub>2</sub>:** High energy radiation, such as UV photons or H<sup>+</sup> ions, has been shown to convert water ice to H<sub>2</sub>O<sub>2</sub> [1,2]. Energetic plasma dissociates H<sub>2</sub>O into H + OH, and H<sub>2</sub>O<sub>2</sub> either forms from the combination of two OH radicals or by reactive scattering of H and OH by water [3]. In irradiation experiments, less H<sub>2</sub>O<sub>2</sub> is detected after radiolysis at higher temperatures, however, the addition of O<sub>2</sub> or CO<sub>2</sub> to water ice can counter this effect [1]. CO<sub>2</sub> acts as an electron scavenger, removing free electrons that can break up H<sub>2</sub>O<sub>2</sub>, and O<sub>2</sub> provides an additional pathway for the formation of H<sub>2</sub>O<sub>2</sub> through H-atom addition. [1] detected H<sub>2</sub>O<sub>2</sub> after irradiation of pure water ice films with 0.8 MeV H<sup>+</sup> ions at 16 K, but not at 80 K. When O<sub>2</sub> or CO<sub>2</sub> was deposited with the water ice prior to irradiation, H<sub>2</sub>O<sub>2</sub> was detected at 80 K.

**Spectral Signature:** H<sub>2</sub>O<sub>2</sub> can be identified in near- to mid-infrared ice spectra. Only three H<sub>2</sub>O<sub>2</sub> absorption bands in this region can be distinguished from water ice bands: a combination band around 3.5 μm, a symmetric bending mode at 6.86 μm, and an asymmetric bending mode at 7.2 μm [2]. Of these, the 3.5 μm band is the strongest, but its position is somewhat variable. [4] examined the spectra of thin films of crystalline and amorphous H<sub>2</sub>O<sub>2</sub> at 4 and 80 K. In general, the crystalline H<sub>2</sub>O<sub>2</sub> had more defined peaks, whereas the amorphous bands were broader and rounder, due to hydrogen bonding and disorder. For condensed amorphous H<sub>2</sub>O<sub>2</sub>, the combination band was found at 3.559 μm, and for crystalline H<sub>2</sub>O<sub>2</sub>, it moved to 3.52 μm. In addition, deposition of H<sub>2</sub>O and

H<sub>2</sub>O<sub>2</sub> together results in an upward shift in the wavelength of absorption bands due to intermolecular frequencies [4].

H<sub>2</sub>O<sub>2</sub> produced by irradiation of H<sub>2</sub>O displays the combination band at 3.501 μm or 3.511 μm with FWHM ranging from 0.02 to 0.12 μm [1,2]. Addition of O<sub>2</sub> to H<sub>2</sub>O before irradiation results in a shift of the absorption band from 3.501 μm to 3.52 μm, and a CO<sub>2</sub>/H<sub>2</sub>O mixture produces a band at 3.508 μm [1]. Other factors affect the position of this band, including the concentration of H<sub>2</sub>O<sub>2</sub> (a larger concentration shifts the band to higher wavelengths), grain size, porosity, and thermal history of the ice [1]. There is also a H<sub>2</sub>O<sub>2</sub> absorption feature in the UV from around 300 nm to shorter wavelengths [3].

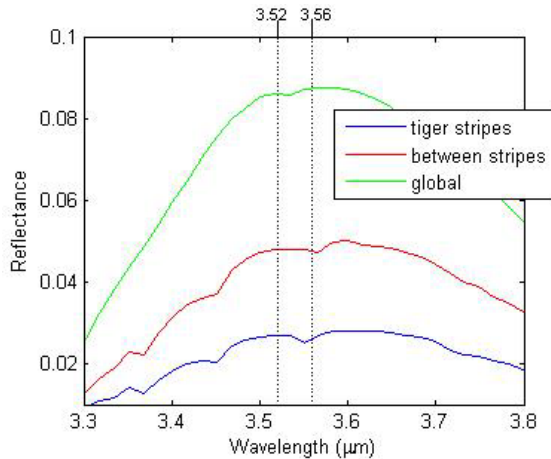
**H<sub>2</sub>O<sub>2</sub> on Europa:** H<sub>2</sub>O<sub>2</sub> has previously been detected on the surface of Europa with a concentration of 0.13% [3]. This detection includes analysis of the 3.5 μm band, in addition to absorption in the UV. Scientists believe that the H<sub>2</sub>O<sub>2</sub> is produced by the bombardment of high energy particles from the Jovian magnetosphere [3]. A smaller H<sub>2</sub>O<sub>2</sub> signal has also been detected on Ganymede and Callisto, where the strength of the radiative bombardment is weaker [5].

**Enceladus Results:** We have tentatively identified H<sub>2</sub>O<sub>2</sub> on Enceladus with data from the VIMS, using the 3.5 μm absorption band as an indicator. This band appears in slightly different positions in the spectra of different regions of Enceladus (Fig. 1). For the spectrum of the south polar "tiger stripes", the band is located at 3.547 μm. For the region between the stripes, the band is at 3.563 μm. A global spectrum of Enceladus (minus the signal from the south pole) shows the band at 3.530 microns.

In general, the 3.5 μm band is stronger and shifted to higher wavelengths in the south polar region of Enceladus. The wavelength shift could be due to a higher concentration of H<sub>2</sub>O<sub>2</sub> in this region, or a higher degree of amorphous as opposed to crystalline H<sub>2</sub>O<sub>2</sub> ice. [6] found a higher concentration of amorphous water ice at the south pole of Enceladus relative to the rest of the satellite, thus a higher concentration of amorphous H<sub>2</sub>O<sub>2</sub> ice in this region would be in agreement with this result.

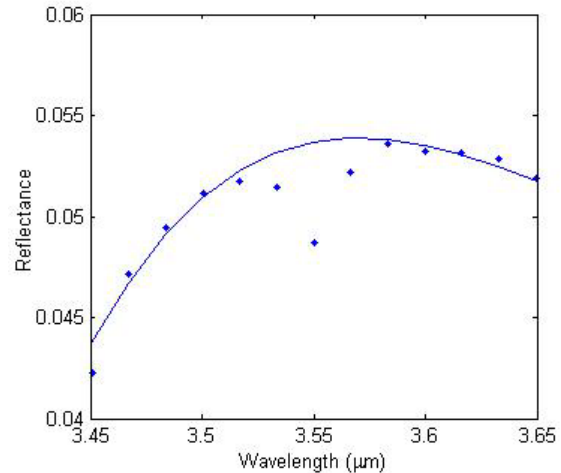
Although the south pole as a whole contains water ice of a more amorphous nature, the tiger stripes have a higher degree of crystallinity than the region in between the stripes [6]. In our work, both the tiger stripes and the region between them display the H<sub>2</sub>O<sub>2</sub> band very close to the amorphous position at 3.56 μm, although the position for the tiger stripe spectrum is

shifted to a slightly lower wavelength. This inconsistency, between the forms of water ice and  $\text{H}_2\text{O}_2$ , could be reconciled by the fact that water ice is crystallized at 140 K, while  $\text{H}_2\text{O}_2/\text{H}_2\text{O}$  mixtures crystallize between 140-200 K [4]. Thus the tiger stripes may be hot enough to crystallize pure water ice, but cool enough such that condensed hydrogen peroxide could remain amorphous.



**Figure 1.** Enceladus spectra from several different areas of the satellite in the region of the  $3.5 \mu\text{m}$   $\text{H}_2\text{O}_2$  absorption band. The blue line is from the “tiger stripes”, the red line is from the region in between the stripes and the green line is a global spectrum. The vertical dotted lines highlight  $3.52$  and  $3.56$  microns, the positions of the band for crystalline and amorphous condensed  $\text{H}_2\text{O}_2$  ice, respectively.

To examine the  $3.5 \mu\text{m}$  band in more detail, we plot the spectrum of the south polar region of Enceladus and the expected continuum without this band (Fig. 2). The position of this band is  $3.547 \mu\text{m}$ , with a FWHM of  $0.04 \mu\text{m}$ . The position of the  $\text{H}_2\text{O}_2$  combination mode band for all of the Enceladus spectra corresponds best to the lab spectra of condensed amorphous ice, condensed crystalline ice, or a mixture. In addition, it also matches somewhat well with the spectrum of an irradiated mixture of  $\text{H}_2\text{O}$  and  $\text{O}_2$  [1].



**Figure 2.** Spectrum of the south pole of Enceladus near  $3.5 \mu\text{m}$  and the interpolated continuum without the absorption band. The Enceladus spectrum is shown in dots and the continuum is shown as a line.

**Conclusions:** We believe we have found evidence of  $\text{H}_2\text{O}_2$  on Enceladus from initial analysis. It is most likely in a condensed form. The position of the band in our data suggests that the south polar region contains  $\text{H}_2\text{O}_2$  ice of a more amorphous nature, whereas the rest of the surface of Enceladus contains more crystalline  $\text{H}_2\text{O}_2$ . We propose that the  $\text{H}_2\text{O}_2$  has been transformed from  $\text{H}_2\text{O}$  through bombardment by high energy particles coming from Saturn’s magnetosphere. Furthermore, the  $\text{H}_2\text{O}_2$  band is stronger in the polar spectra, suggesting that the radiation at the poles is stronger, perhaps due to a magnetic field at Enceladus diverting the high-energy particles polewards. This explanation could also account for the distribution of more amorphous  $\text{H}_2\text{O}_2$  at the poles and more crystalline  $\text{H}_2\text{O}_2$  elsewhere, as radiative bombardment could disrupt the crystalline structure of  $\text{H}_2\text{O}_2$  as it does with  $\text{H}_2\text{O}$  [7].

We would like to confirm our detection of  $\text{H}_2\text{O}_2$  by looking for the absorption feature in the UV. We would also like to look for  $\text{H}_2\text{O}_2$  on the other icy satellites of Saturn.

**References:** [1] Moore, M.H. and Hudson R.L. (2000) *Icarus* 145, 282-288. [2] Zheng, W., et al. (2006) *Astrophys. J.* 639, 534-548. [3] Carlson, R.W., et al. (1999) *Science* 283, 2062-2064. [4] Giguere, P.A. and Harvey, K.B. (1959) *J. Mol. Spectr.* 3, 36-45. [5] Hendrix, A.R. et al. (1999) *LPSC XXX*, 2043. [6] Newman, S.F. et al. (2007) *submitted to Icarus*. [7] Moore, M.H. and Hudson, R.L. (1992) *Astrophys. J.* 401, 353-360.