

IMCS25_332 - Flexible and thin Top-Bottom-Electrode Hydrogen Sensors for Aerospace and Battery Applications

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Summary

Hydrogen (H₂) is undoubtedly the most popular future green energy carrier due to its harmless by-product, water. Thus, hydrogen becomes the most attractive energy carrier for light-weight and high safety requiring applications such as aerospace and power storage. However, as far as storage and transport of hydrogen in mobile applications are concerned, still the precautions must be taken, because H₂ is colourless and odourless but flammable in air with very low ignition energy and combusts in air [1]. The so-called Top-Bottom-Electrodes (TBE) combined with TiO₂ sensing layer [Pt/TiO₂/Pt] utilising Memristor sensor principle. Pt results in reduced Pt/TiO₂ Schottky interface [2].

Content

By Memristor sensors, the sensing mechanism is primarily justified with the variations in Schottky Barrier Height at the Pt/TiO₂ and TiO₂/Pt interfaces. Moreover, it is proven that in terms of hydrogen sensing, Pt acts as a catalyst to dissociate hydrogen molecules that can migrate quickly to the TiO₂ surface and further diffuse into the bulk. Relying on this fact, the Cr-doped TiO₂ sensing layer is fabricated in two different stoichiometric relations by means of reactive sputtering as sandwiched between the top and bottom Pt-electrodes. The bottom side of the sensing layers deposited by inducing 5 sccm O₂ while top-side with 7 sccm O₂, each for 4h (Fig. 1). Thus, the Schottky barrier height at TBE interfaces differs. To enable flexibility, the sensor fabrication is carried on polyimide (PI) foils.

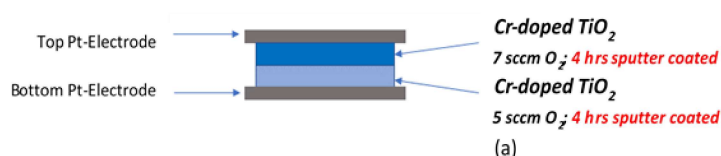


Figure 1: Schematic of Top-Bottom-Electrode sensor configuration fabricated Cr-doped TiO₂ sensing layers

The gas sensing tests of Pt/TiO₂/Pt structured sensors were performed with a computer-controlled experimental setup (sensor and catalyst characterization unit - SESAM) at DLR Cologne. The sensing measurement tests towards 600-2000 ppm H₂ are carried in synthetic air and in argon carrier gas at temperatures varying RT, 40°C and 60°C under the bottom electrode bias conditions. At RT and in Argon, the sensor shows a moderate response displayed with significant drop in resistance on hydrogen release whereas remains very poor in synthetic air. At 40° C, under both carrier gas environment, the sensitivity increases due to enhanced interaction between hydrogen molecule and the sensing surface and baseline stability improves. Significant baseline resistance drift has been eliminated by heat-treatment of the whole sensor at 100°C yielding better sensor property. The baseline resistance of the sensor is unstable at 60° C however sensitivity improves further.

References

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