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Highlights

Experimental investigation of the emissions and performance of a micro gas turbine setup with enhanced EGR

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- EGR is limited by the incompleteness of the combustion due to oxygen depletion
- An mGT is modified to control the EGR rate and operated until flame-out
- A CO₂ concentration of 7.9% is reached and 70% EGR minimizes CO emission
- EGR can increase the combustion chamber residence time promoting NO_x production
- Questions at high pressure combustion require future work at pressurized conditions

Experimental investigation of the emissions and performance of a micro gas turbine setup with enhanced EGR

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Abstract

Exhaust gas recirculation (EGR) is investigated to reduce the amine-based carbon capture penalty of combined cycle gas turbines by the reduction in mass flow rate and the increase in CO₂ concentration permitted by the semi-closed cycle. Furthermore, EGR is one of the best pathways to reduce NO_x emissions. While many numerical investigations have been performed in literature, there is a clear lack of full-scale experimental investigations on a real gas turbine. To answer that need, an MTT EnerTwin® micro gas turbine has been modified and equipped with an external EGR loop allowing to apply recirculation rates up to flameout. Within the wide spectrum of EGR fraction, the composition of the exhaust gases and the combined heat

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and power production has been measured. While EGR effectively allows to reach a dry CO₂ concentration up to 7.9%, decrease NO_x emissions and slightly improve the combined thermal and electrical production (due to the higher specific heat capacity of the working fluid, when applying EGR, and low recirculation temperature), CO emissions are the main limiting factor before flameout is reached. However, the results observed at low pressure (3-5 bar and TIT of 950 °C for mGT) cannot be directly transposed at high pressure (15-20 bar and TIT of 1300-1400 °C for industrial GT) due to the sensitivity of NO_x formation chemistry to pressure and temperature levels. The significant differences between mGTs and industrial GTs make complex any comparisons and emulations between those two scales. Extrapolating the results from mGTs to industrial GTs thus present some limitations and further investigations need to be done to understand the impact of pressure on the NO_x and CO production. Nevertheless, EGR has been characterized experimentally on a mGT and identified as a clear potential pathway to carbon neutrality by improving post combustion capture efficiency owing to the gain in CO₂ concentration.

Keywords: micro gas turbine, exhaust gas recirculation, NO_x reduction, CO emission, CO₂ concentration

Nomenclature

Acronyms	CAPEX	capital expenditure	
	CCGT	combined cycle gas turbine	
	EGR	exhaust gas recirculation	
	HHV	higher heating value	kJ/kg
	HRSG	heat recovery steam generator	
	LHV	lower heating value	kJ/kg
	mGT	micro gas turbine	
	PCC	post-combustion capture	
	TIT	turbine inlet temperature	°C
	TOT	turbine outlet temperature	°C
Greek letters	η	energy efficiency	%
	λ	air-fuel equivalence ratio	
	μ	statistical mean	
	ϕ	fuel-air equivalence ratio	
	σ	statistical standard deviation	
Roman letters	c_p	specific heat capacity	J/kg °C
	\dot{m}	mass flow rate	kg/s
	P	power	kW
	Q	heat duty	kW
	W	mechanical work	kW
	x	molar fraction	% mol.
Subscripts	amb.	ambient	
	e	electrical	
	th	thermal	

1. Introduction

1 Gas turbines have already been identified as one of the pathways to car-
2 bon neutrality. On the road to zero-carbon economy, the intermittency and
3 fluctuations of renewables need to be compensated and gas turbines come
4 up as a flexible and efficient solution. Alongside their responsiveness to load
5 demand, their use in combination with post-combustion capture (PCC) en-
6 sures the compliance with carbon emission levels while still valorizing the
7 energetic and economic potential of fossil fuels.

8
9 While exhaust gas recirculation (EGR) is a well-known technique used to
10 cut down on NO_x emissions [1], Gülen et al. [2] and De Paepe et al. [3] have
11 also highlighted its four advantages on carbon capture as a way of:

- 12 1. increasing the CO_2 concentration;
- 13 2. decreasing the O_2 concentration (chemical stability of the amines);
- 14 3. decreasing the NO_x emissions (chemical stability of the amines);
- 15 4. decreasing the mass flow of exhaust gases to treat.

16 As a lower flow of exhaust gases with a higher concentration of CO_2 sig-
17 nificantly decreases the size and the economical investments related to the
18 carbon capture unit [4], the reduction in O_2 and NO_x concentrations also
19 results in a benefit to the chemical stability of the amines used for after-
20 treatment of the exhaust gases [5].

21

22 Achieving the highest level of EGR therefore seems to be a promising way
23 of mitigating the inevitable energy penalty and CAPEX increase of post-
24 combustion capture (PCC) by getting closer to stoichiometric conditions [4].
25 However, it has been shown by Ali et al. [6], Tanaka et al. [7] and El Kady
26 et al. [1] that going below 16% of dry O₂ at the combustion chamber inlet
27 leads to unacceptable levels of unburned hydrocarbons and CO emissions.
28 Moreover, the implementation of EGR requires an external blower that can
29 decrease the net power of the cycle [3].

30
31 Camaretti et al. [8] have experimented with EGR on an mGT for natural
32 gas and biogas, and have noticed a depletion in NO_x and O₂. However, they
33 have not applied EGR below 16% O₂ at the combustion chamber inlet. On
34 the industrial side, Syed et al. [9] have identified the N₂O and Zeldovich's
35 routes as the dominating mechanisms of NO_x production, however their study
36 remains numerical. Ali et al. [6] have validated a model of mGT and have
37 applied EGR up to 3.7% mol CO₂ in the flue gases. Nevertheless, this value
38 remains far from the stoichiometric limit. On the numerical side, De Santis et
39 al. [10] have investigated the impact of CO₂ on flame stabilization and flame
40 speed in the combustion chamber of an mGT. They have shown that the
41 increased CO₂ concentration reduces the flame speed and thus the combus-
42 tion stability, nevertheless, no experimental validation has been performed.
43 At DLR, Kutne et al. [11] have studied the flame stabilization at different
44 pressure levels and with different EGR rates. They have highlighted that in-

45 creasing the pressure allows to stabilize the flame at higher EGR rates. Even
46 though Cameretti et al. [8], De Santis et al. [10] and Ali et al. [6] have either
47 simulated or emulated EGR on micro gas turbines to monitor the evolution
48 of CO, CO₂ and NO_x emissions, none of them have tried to experimentally
49 push to the flameout limit. Furthermore, their studies have not covered the
50 broader aspects such as the power generation or the combined production of
51 heat. The impact of EGR on CO, CO₂, O₂ and NO_x is thus documented in
52 literature but only for moderate recirculation rate and mostly without ex-
53 perimental validation.

54

55 No experiment has currently been run on a full scale gas turbine, and
56 some works have been realized on mGT but EGR has never been applied up
57 to the point of flameout. As a result, there is a clear need to experimentally
58 achieve higher EGR levels and understand:

- 59 • how the power production of the machine is impacted by the EGR;
- 60 • how the quality of the heat recoverable in the economizer is impacted
61 by the change in composition of the flue gases;
- 62 • how CO and NO_x emissions evolve while getting closer to stoichiometry;
- 63 • what is the maximal CO₂ concentration in the exhaust gases before
64 flameout.

65 Those understandings are crucial to assess the impact of EGR on the power
66 production, the combined heat recovery potential and the composition of the

67 exhaust gases.

68

69 On the side of large-scale GTs, Tanaka et al. [7] have clearly identified
70 the dilemma between a high combustion temperature, synonym of greater
71 efficiency, and thermally boosted NO_x production. They have described how
72 EGR positively decreases the emission of NO_x by cutting down on the local
73 high flame temperature zone and enhancing the homogeneity of the air/fuel
74 mixture. Furthermore, they have also highlighted that presence of NO_x in
75 the air inlet, induced by EGR, has no effect on their production during the
76 combustion. In a previous work, El Kady et al. [1] have mimicked EGR in a
77 premixed combustor up to 1 MPa. They have shown that EGR narrows the
78 flame stability limits but allows to reach more than 8% CO_2 on a dry basis
79 in the flue gases. Furthermore, it has been confirmed that EGR contributes
80 to NO_x reduction and that CO emissions are the limiting factor of EGR.

81

82 This paper thus addresses the lack of experimental studies by presenting
83 a modified mGT where EGR rates up to flameout can be applied. This mod-
84 ified setup allows to investigate the impact of EGR on the exhaust mass flow
85 rate, net power production and emissions of O_2 , CO_2 , NO_x and CO. In large-
86 scale combined cycle power plant, the application of exhaust gas recirculation
87 changes the composition of the gases flowing through the heat recovery steam
88 generator, potentially impacting the performances of the bottoming cycle.
89 The presence of an external economizer in the mGT setup for the production

90 of hot water thus enables to get some insights on the heat recovery potential
91 in the exhaust gases. While it does not represent correctly the HRSG of a
92 steam cycle, some trends can be extracted on how EGR impacts the quality
93 of the heat at the turbine outlet.

94

95 By applying higher EGR rates, the experimental setup presented in this
96 paper innovatively allows to study and understand the impacts and limits of
97 EGR on an mGT. The heat and power production as well as the emission
98 levels in the exhaust gases will therefore be monitored for different recircula-
99 tion rate up to flameout. Given that EGR is one of the potential contributors
100 envisaged to decarbonize CCGT, this modified setup can also be innovatively
101 assimilated as a small-scale combined cycle gas turbine. This experimental
102 work does not only verify the trends highlighted by the numerical simulations
103 carried out in literature, but can also be used, under a few hypotheses, to
104 emulate the behaviour of large-scale CCGT. Despite the significant difference
105 between mGTs and industrial GTs, making complex any comparisons and
106 emulations between those two scales, the approach consisting on extrapolat-
107 ing the results from mGT to industrial GT has been done successfully by
108 Reboli et al. [12]. This method has however some limitations and further
109 investigations thus need to be done to understand the impact of pressure on
110 the NO_x and CO production.

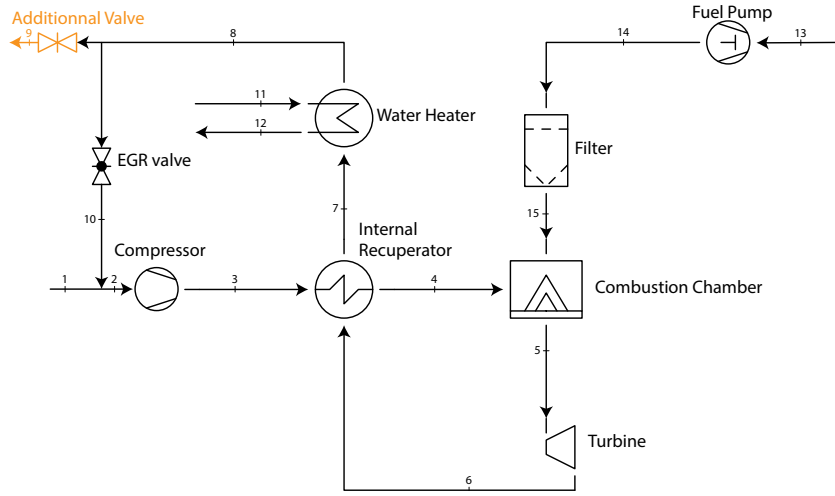


Figure 1: The MTT EnerTwin[®] micro gas turbine is equipped with an internal recuperator allowing to recover heat at the turbine outlet in order to preheat the air entering the combustion chamber. The experimental setup also contains a water heater for the production of warm water that can mimic an HRSG. In addition to the original EGR valve, an additional valve has been installed at the exhaust to restrict flow and force the recirculation.

111 2. Methodology

112 To fill the gaps on experimental work where EGR is applied up to flame-
 113 out, a commercial mGT has been modified and equipped with an external
 114 EGR loop as represented in Figure 1. Exhaust gas recirculation thus takes
 115 a fraction of the flue gases in the exhaust pipe (8) and recirculates them to
 116 the compressor inlet (2). By doing so, the fresh air flow rate (1), specifically
 117 the oxygen flow rate, is decreased leading to combustion conditions closer to
 118 stoichiometry.

119

120 The exhaust gas recirculation rate is defined as the ratio between the
 121 flow recirculated to the compressor (10) and the flow leaving the combus-

122 tion chamber (5). While many studies are currently identifying EGR rate
123 limitations at 40% [4] or 45% [13] leading to an air-fuel equivalence ratio
124 λ around 1.5 (fuel-air equivalence ratio ϕ of 0.67 as shown in Figure 2) on
125 large-scale gas turbines, the highly diluted combustion encountered in mGTs
126 requires much larger EGR rates to reach the same equivalence ratio. Fig-
127 ure 3 compares the air-fuel equivalence ratio for the GE 9HA.02 and the
128 MTT EnerTwin[®] for the same EGR rate. As represented in Figure 3, the
129 air-fuel equivalence ratio of the mGT starts at 9.3 (fuel-air equivalence ratio
130 ϕ of 0.1) when no EGR is applied. At 45% EGR, λ is still at 5.1 opposed
131 to the 1.5 that would have been expected on an industrial GT fed with pure
132 methane. To reach a λ of 1.5, 84% EGR needs to be applied to the mGT
133 while the stoichiometric limit is situated at 89%. For two different turbines,
134 the recirculation rate cannot be used to compare the operating conditions of
135 the combustion chamber as it does not lead to the same meaning in terms of
136 air-fuel equivalence ratio.

137

138 Starting from the operating conditions without EGR, the exhaust gases
139 are progressively recirculated while keeping constant the mass flow in the
140 combustion chamber. A simple input-output modelled is used assuming com-
141 plete combustion of fuel. The dry and wet compositions of the gases before
142 and after the combustion chamber are then computed. It can be observed in
143 Figure 4 that 71% EGR are required to move the dry O₂ fraction from 21%
144 (air related with no EGR) towards the 16% dry limit announced by El-Kady

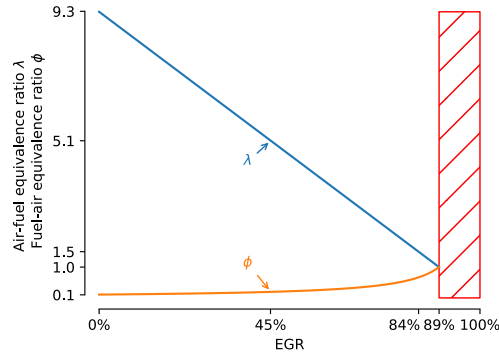


Figure 2: Starting at 9.3, the air-fuel equivalence ratio λ decreases linearly with the EGR level to reach 1 at stoichiometry when 89% of EGR is applied. Due to the high dilution rate, 84% of EGR has to be applied to reach a λ of 1.5, equivalent to the value encountered with 45% EGR on a large-scale industrial gas turbine.

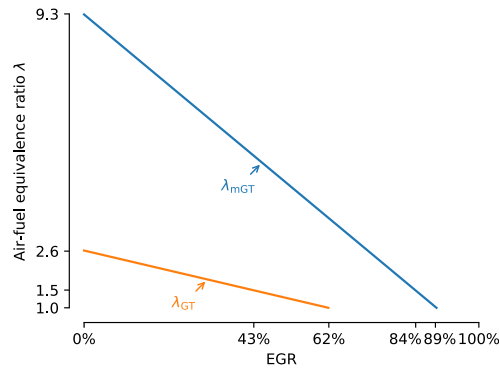


Figure 3: Compared to the GE 9HA.02, the higher dilution encountered in a micro gas turbine allows to apply a bigger recirculation rate for a same λ . For instance, the stoichiometry is reached with a recirculation rate of 62% for an industrial GT, this is only obtained with 89% of recirculation rate for the mGT.

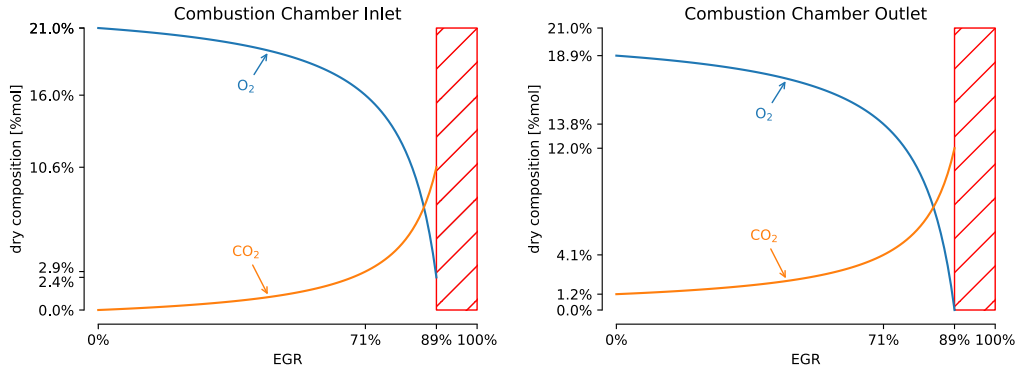


Figure 4: The application of EGR allows to increase the CO₂ concentration at the combustion chamber inlet and outlet as well as decreasing the O₂ molar fraction. The maximum EGR level that can be applied to keep analytically a complete combustion is 89%. At 71% of EGR, the O₂-concentration at the combustor inlet reaches 16%.

145 et al. [1]. At that level, the exhaust gas contains 4.1% of CO₂ and 13.8% of
 146 O₂ on a dry molar basis. Nevertheless, Figure 4 shows that higher recircu-
 147 lation rates can be theoretically expected to reach the stoichiometric limit
 148 at 89% EGR. That limit therefore maximises the CO₂ outlet concentration
 149 (12%) but also leads to the most severe combustion condition with only 2.4%
 150 of O₂ at the combustion chamber inlet. Those conditions are difficult to sta-
 151 bilize in real combustion chambers and the emissions of CO and unburned
 152 are often limiting factors, lowering the achievable EGR rate.

153

154 While the analytical equilibrium-based laws allow to predict the dry mo-
 155 lar composition of the exhaust gases, at both the inlet and outlet of the
 156 combustion chamber, some questions are remaining with respect to:

- 157 • kinetic mechanisms related to the production of NO and NO₂;

- 158 • incomplete combustion leading to the production of CO;
- 159 • flexibility of the combustor to come closer to the stoichiometric limit.

160 An experimental setup where higher EGR rates can be applied is therefore
161 required to observe the evolution of:

- 162 • the concentration of CO₂;
- 163 • the concentration of O₂;
- 164 • the concentration of CO;
- 165 • the concentration of NO and NO₂;
- 166 • the net electrical power produced;
- 167 • the heat recovered from the exhaust gases for cogeneration via hot
168 water production.

169 *2.1. Initial unmodified setup*

170 The initial setup is an MTT EnerTwin® micro gas turbine which can
171 produce between 1.0 and 3.2 kW_e combined with a thermal production rang-
172 ing from 6.0 to 15.6 kW_{th}. Alongside electricity generation, this mGT has
173 the option to produce domestic hot water (up to 80 °C). While the net grid
174 output efficiency varies between 10 and 16%, the combination with the heat-
175 ing system allows to achieve a total efficiency exceeding 94% of the LHV.

176

177 As represented in Figure 1, the MTT EnerTwin® mGT exploits the
178 recuperated Brayton cycle. The inlet air enters the variable speed radial
179 compressor (2) at atmospheric conditions and is compressed up to 2.4 bar.
180 After the compression (3), the air is preheated in the internal recuperator
181 up to 720 °C (4), where it encounters a pressure drop, before being mixed
182 with natural gas in the combustion chamber. The flow (5) is then expanded
183 through the turbine to 1 bar (6) and a fraction of the remaining energy is
184 transferred to the compressed air in the internal recuperator by cooling the
185 exhaust gases from 790 °C down to 230 °C (7). The flow finally enters the
186 water heater for the combined heat purpose where in the setup tap water
187 is employed as cooling agent. This uncommon practice allows the exhaust
188 gases to be cooled as low as 14 °C (8).

189

190 The manufacturer has already foreseen an EGR valve in the mGT al-
191 lowing a fraction of the exhaust gases to be recirculated to the compressor
192 inlet (10). While working with higher exhaust gas temperatures (related to a
193 production of hot water when a buffer is connected to the mGT), the control
194 logic monitors the EGR valve opening to decrease the electric power produc-
195 tion when the grid inverter current limit is reached. Alongside the electric
196 current limit, the inner controller allows to enter different set-points in order
197 to regulate the thermal power. The control algorithm will then monitor the
198 fuel pump and the rotational speed so that the turbine outlet temperature (6)
199 remains constant.

200

201 The mGT has been equipped with different sensors allowing to measure
202 the temperature, pressure and flow rate at different locations. The compo-
203 sition of the flue gases is measured with a Testo® 350 analysis box that
204 can survey the evolution of O₂, NO, NO₂, CO and CO₂ with an infrared
205 cell. The uncertainties related to the measurement of the exhaust gases are
206 presented in Table 1. The pressure levels are measured with Huba Control®
207 sensors (absolute error ± 0.012 bar) and temperatures either with Tasseron®
208 NTC 10K3% sensors (absolute error ± 0.9 °C) or with K-type thermocouples
209 (absolute error ± 2.2 °C). The inlet air mass flow is measured with the vortex
210 flowmeter from Yokogawa® and the gas flow with the Elster® BK-G4M16
211 (absolute error $\pm 2\%$ of read value).

212

213 *2.2. Manual control of the EGR valve*

214 Although the EGR valve was already controlled by the inner controller of
215 the mGT, the EGR level needs to be changed independently from the control
216 system. The connection lines to the valve have therefore been cut and linked
217 to an external control system allowing to manually decide the closing of the
218 valve. The designed PID system enhances to vary the closing ratio from 0
219 to 100% at the proper moment imposed by the testing campaign. The PID
220 has been implemented on an Arduino Uno® board and gives a signal to the
221 L298 full-bridge driver, controlling the DC motor, based on the closing angle

component	measuring range	accuracy	resolution	reaction time t_{90}
O ₂	0 - 25 % vol.	$\pm 0.8\%$ of fsv	0.01 % vol.	20 s
CO	0 - 199 ppm	± 10 ppm	1 ppm	40 s
	200 - 2 000 ppm	$\pm 5\%$ of mv	1 ppm	40 s
	2 001 - 10 000 ppm	$\pm 10\%$ of mv	1 ppm	40 s
NO	0 - 99 ppm	± 5 ppm	1 ppm	30 s
	100 - 1 999 ppm	$\pm 5\%$ of mv	1 ppm	30 s
	2 000 - 4 000 ppm	$\pm 10\%$ of mv	1 ppm	30 s
NO ₂	0 - 99.9 ppm	± 5 ppm	0.1 ppm	40 s
	100 - 500 ppm	$\pm 5\%$	0.1 ppm	40 s
CO ₂	0 - 25% vol.	$\pm 0.3\%$ vol. + 1% of mv	0.01% vol.	10 s
	25 - 50% vol.	$\pm 0.5\%$ vol. + 1.5% of mv	0.1% vol.	10 s

Table 1: Testo® refers for each cell the measuring range, the accuracy, the resolution as well as the reaction time t_{90} . (mv: measured value, fsv: full-scale value)

222 of the valve measured with a KMA210 angle sensor.

223

224 When fully open, the EGR valve only allows to recirculate up to 30% of
225 the flue gases. As indicated in Figure 2, this is too low to reach flameout.
226 On large-scale CCGTs, EGR levels can reach 45% in an adapted combustor
227 before combustion issues are faced (e.g. incomplete combustion) due the
228 air-fuel ratio close to stoichiometry. However, the highly diluted combustion
229 encountered in mGTs ($\lambda \simeq 9.4$) requires much larger EGR levels to reach
230 the same richness as the one in the industrial combustors. In order to exceed
231 the limit of 30% EGR, still leading to a highly diluted combustion ($\lambda \simeq 6.6$),
232 an additional valve has been placed in the exhaust gas duct as represented
233 in orange in Figure 1. This valve increases the pressure drop in the exhaust

234 pipe (9) and forces a higher fraction of the flow to be recirculated (10) (throt-
235 tling effect). This last modification leads to recirculation ratios reaching 84%
236 and combustion conditions closer to stoichiometry. As the combustion issues
237 are faced for a λ around 1.5 in industrial GTs, it is interesting to investigate
238 the stability of the combustor as well as the emission levels for this richness.
239 Moreover, a given closing angle of the exhaust valve leads to the flame be-
240 ing extinguished, and therefore makes it possible to test all the EGR levels
241 physically achievable on this installation. As explained by Pappa et al. [14],
242 the flameout is related to the staggering of the combustion chamber where
243 the primary zone encounters smaller equivalence ratio λ (between 0.29 and
244 0.44), while the overall equivalence ratio is equal to 1.5.

245 *2.3. Experimenting with natural gas*

246 The fuel price can be a real constraint restricting the number of experi-
247 ments as well as their duration. While working with pure methane reduces
248 the uncertainties related to the fuel gas composition and thus to the compu-
249 tation of the exhaust gas composition, its price per kWh can be 30 up to 60
250 times more expensive than working with natural gas. However, natural gas
251 has the main drawback of an unknown mean composition that is changing
252 daily.

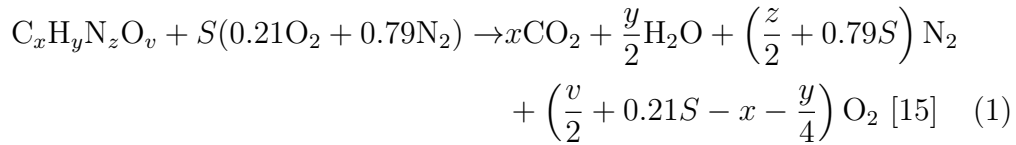
253

254 According to the data delivered by the gas provider, the natural gas de-
255 livered in the UMONS lab is composed of: CH_4 , C_2H_6 , C_3H_8 , C_4H_{10} , C_5H_{12} ,

256 He, N₂, CO₂ as well as higher refined gases (C₆H₁₄ and more). For simplicity
 257 reasons in the chemical equations, natural gas is written under its summa-
 258 rized formula: C_xH_yN_zO_v.

259

260 For each fuel, the maximal amount of CO₂ in the exhaust gases is obtained
 261 at stoichiometry as presented in Equation 1 for the molar ratio of air S^* com-
 262 puted in Equation 2. The dry maximal molar concentration CO_{2, max}, that
 263 will be used afterwards, can be directly deduced from the fuel composition
 264 as expressed in Equation 3. Based on the data provided by the gas provider,
 265 the CO_{2, max} is computed and represented in Figure 5. While methane has a
 266 CO_{2, max} of 11.7%, the presence of gases with longer carbon chains in natural
 267 gas leads to an expected value slightly higher than 12.05%. Besides its lower
 268 cost, experimenting with natural gas allows thus to reach higher CO₂ content
 269 in exhaust gases and facilitates to simulate situations with a lower level of
 270 oxygen at the combustion chamber inlet.



$$\frac{v}{2} + 0.21S^* - x - \frac{y}{4} = 0 \leftrightarrow S^* = \frac{x + \frac{y}{4} - \frac{v}{2}}{0.21} \quad [15] \quad (2)$$

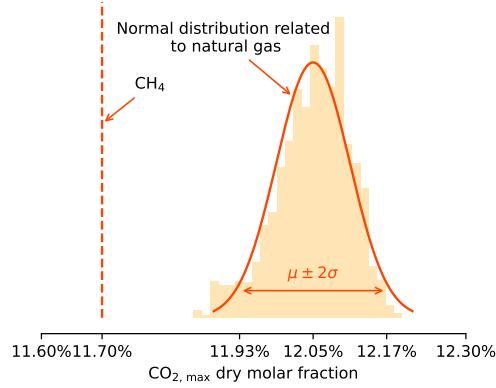


Figure 5: 95% of the maximal CO_2 dry molar fraction, resulting from the stoichiometric combustion of natural gas, is located between 11.93% and 12.17%. Due to the presence of atoms with a higher carbon content in natural gas, the $\text{CO}_{2, \max}$ of methane (11.7%) is below the one of natural gas (12.05%).

271

$$\text{CO}_{2, \max} = \frac{x}{x + \left(\frac{z}{2} + 0.79S^*\right)} = \frac{x}{x + \left(\frac{z}{2} + 0.79 \frac{x + \frac{y}{4} - \frac{v}{2}}{0.21}\right)} \quad [15] \quad (3)$$

272

273 2.4. Organization of the testing campaign

274 To carry out the tests, the mGT was first preheated for two hours at a set
 275 point of 70%. The preheating phase heats up the components (in particular
 276 the heat exchangers and combustion chamber) so that their temperatures are
 277 constant. Ten different EGR rates are then applied, waiting up to 30 minutes
 278 between each to ensure that stability is achieved. During the post-treatment
 279 phase, the data is averaged over the last 5 minutes of the applied EGR rate.

280 3. Results and discussion

281 The experimental setup previously presented is run on a range of EGR
282 ratios starting from 0% up to flameout at 84%. The power produced and the
283 heat transmitted to the water are then measured and the composition of the
284 exhaust gas is analysed in terms of O₂, CO₂, NO_x and CO.

285 3.1. Global cycle performance

286 The impact of EGR on the net power and the heat duty have been com-
287 puted and are represented in Figure 6. Measurements at the generator termi-
288 nals show that EGR positively affects the electrical production. Indeed, the
289 net power produced increases by 2.1 W_e per additional EGR percentage and
290 applying 80% EGR thus leads to a relative increase of 8%. This tendency is
291 explained by:

- 292 • the control strategy of the mGT keeping a constant TOT;
- 293 • the below-ambient temperature of the recirculated gases (due to indi-
294 rect tap water cooling).

295 At the outlet of the water heater (point 8 in Figure 1), the exhaust gases
296 are cooled down to around 14 °C. This temperature is 5 °C lower than the
297 ambient temperature and applying EGR therefore decreases the compressor
298 inlet temperature (2) proportionally. While a colder flow (2) at the compres-
299 sor inlet is beneficial to the cycle, this also means that the compressor outlet
300 temperature (3) is decreased and subsequently the combustion chamber inlet

301 temperature (4) too. Although this phenomenon is partially counteracted by
302 the recuperator, more fuel is consumed to maintain a constant TOT, allowing
303 more heat to be recovered from the water heater, as can be seen in Figure 6.
304 Per percentage of EGR applied, an additional $24 W_{th}$ is recovered in the wa-
305 ter heater and this represents a 14% increase at 80% EGR. As opposed to the
306 simulations presented by De Paepe et al. [3], the absence of auxiliaries and
307 the lower air inlet temperature of this setup benefit to the cycle. The mGT
308 installed in the UMONS lab does not suffer from high back pressure when
309 advanced recirculation rates are applied. This result is obtained thanks to
310 the added external recirculation loop which have a diameter high enough to
311 allow recirculation without significant friction loss. It is the combined effect
312 of the large diameter with the relative small flow rate. An external fan is
313 therefore not needed and the back pressure only increases by 600 Pa between
314 0% EGR and 84% EGR.

315

316 Furthermore, the last author of this paper has raised in previous work [3]
317 the question of the impact of the change in composition on the cycle perfor-
318 mance. To get rid of the impact of the recirculated gases colder temperature,
319 new experiments have been realized when the ambient temperature (1) is
320 equal to the recirculation temperature at the outlet of the water heater (8).
321 When the compressor inlet (2) temperature is maintained at 16 °C, one can
322 see that EGR still increases the net power and the heat duty of the cycle
323 (Figure 8). However, the slopes of the dashed regression lines from Figure 6

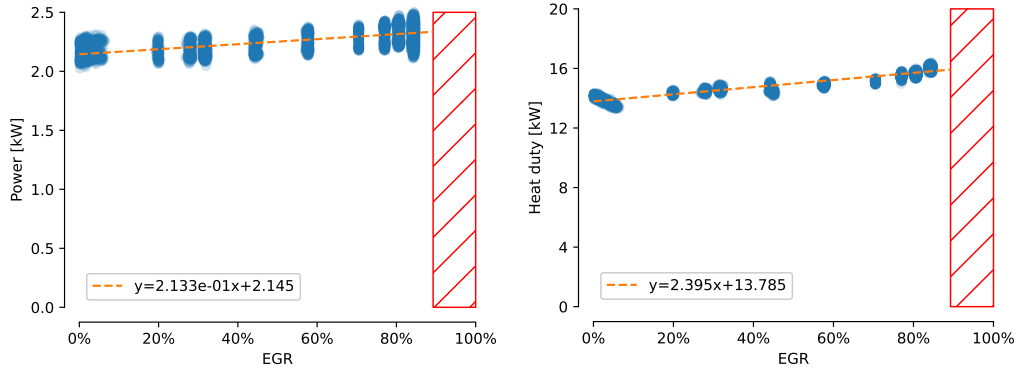


Figure 6: The application of EGR increases the net power production as well as the heat recovery potential in the conditions where the temperature of the recirculated gases (14 °C) is lower than the ambient temperature (16 °C).

324 are reduced in Figure 8 respectively with 36% and 33%, leading to more
 325 constant heat and power productions. This phenomenon is explained by the
 326 change in composition of the flue gases. Indeed, EGR increases the content
 327 of water in the flow leading to a higher c_p . From 1030 J/kg °C when no
 328 EGR is applied, the c_p reaches 1141 J/kg °C at the turbine inlet when 84%
 329 of the exhaust gases are recirculated (Figure 7). At the compressor side, the
 330 c_p changes from 1003 J/kg °C to 1119 J/kg °C. In order to keep a constant
 331 TOT, more fuel is thus required and the combined heat and power produc-
 332 tions are increased.

333

334 The evolution of the net work W_{net} is described by Equation 4 where the
 335 fluid properties show to have a significant impact (γ and c_p). As previously
 336 explained, the EGR rate changes the composition of the gases flowing through
 337 the turbomachinery Figure 7. Based on the variation in c_p and γ depicted

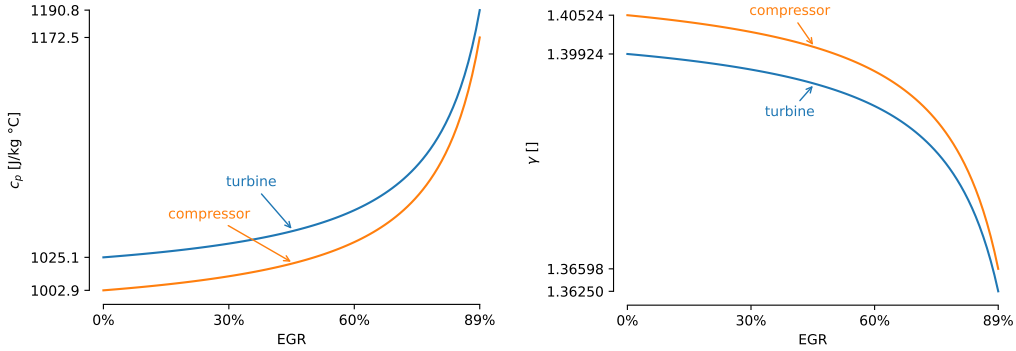


Figure 7: Due to the composition change and the higher H₂O content, the c_p increases with the application of EGR, however the heat capacity ratio γ decreases.

338 in Figure 7, the relative increase in net work $W_{\text{net, x\% EGR}}/W_{\text{net, 0\% EGR}}$ can
 339 be analytically computed as depicted in Figure 9. While the experimental
 340 work highlighted a relative increase of 1.13 Figure 8 at the maximal allowable
 341 level of EGR, regarding the stability of combustion, the following analytical
 342 expression predicts a relative increase of 1.11 Figure 9. The small differ-
 343 ence between those two values comes from the simplified model presented in
 344 Equation 4 in terms of the evolution of turbomachinery efficiency and fluid
 345 properties. As demonstrated in Figure 8, the back pressure increase does not
 346 have a strong impact on the net power production. The 600 Pa maximal
 347 raise in the back pressure is indeed not sufficient to invert the trend in the
 348 enhancement of the power production due to change in fluid properties.

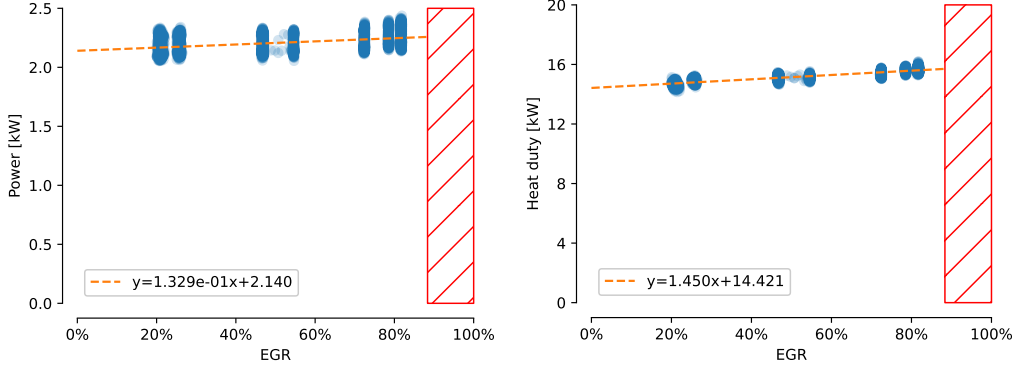


Figure 8: The application of EGR, when the compressor inlet temperature is maintained constant increases more slightly the net power production and the heat recovery potential.

$$W_{\text{net}} = -(W_c + W_t) = -\dot{m} \left(\frac{c_{p,2}}{\eta_c} T_2 \left(\left(\frac{p_3}{p_2} \right)^{\frac{\gamma_2 - 1}{\gamma_2}} - 1 \right) + c_{p,5} \eta_t T_5 \left(\left(\frac{p_6}{p_5} \right)^{\frac{\gamma_5 - 1}{\gamma_5}} - 1 \right) \right) \quad [16] \quad (4)$$

349 For this latter experiment, the efficiency Equation 5 is computed based on
350 the higher heating value (HHV) to take into account the partial condensation
351 occurring in the water heater (7-8) Figure 1. Figure 10 shows that EGR
352 slightly increases the efficiency η_{HHV} . This evolution is due to the increase
353 in W_e and Q_{th} related to the higher specific heat capacity encountered when
354 partial recirculation is applied. Regarding to the net work W_e , the evolution
355 of the composition during the different EGR levels increases the net power as
356 described by Equation 4 represented in Figure 9. Regarding the heat flux Q_{th} ,
357 the increase of the mass heat capacity has a positive impact on the external
358 economizer. Indeed, the control strategy ensuring a constant TOT makes

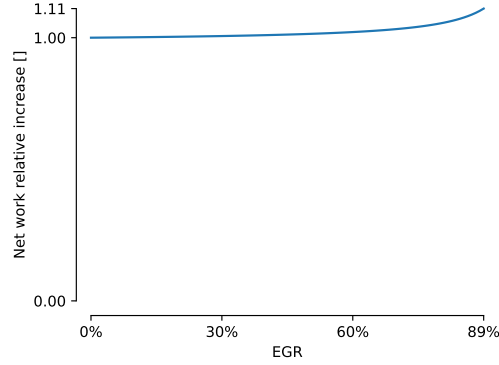


Figure 9: The evolution of the relative work when the variations in c_p and γ are taken into account, as calculated in the analytical Equation 4, shows that a relative increase of 11% is reached when 89% of EGR is applied.

359 the economizer hot inlet temperature constant, while the high efficiency of
 360 the economizer also enables a constant hot outlet temperature among the
 361 different EGR levels. As the temperature difference remains constant, the
 362 increase in c_p directly leads to a higher heat recovery.

$$\eta_{\text{HHV}} = \frac{P_e + Q_{\text{th}}}{\dot{m}_{\text{fuel}} \text{HHV}} \quad [16] \quad (5)$$

363 3.2. O_2 , CO_2 , NO_x and CO content

364 Figure 11 shows that the O_2 dry fraction follows the analytical predictions
 365 presented in Figure 4. Starting at 18.9%, when no EGR is applied due to the
 366 highly diluted conditions, the concentration of O_2 in the exhaust gas reaches
 367 a minimum at 6.6% with 84% of EGR.

368 Figure 12 shows that the CO_2 dry fraction, measured with the infrared
 369 cell, follows the analytical predictions presented in Figure 4. Starting at 1.2%

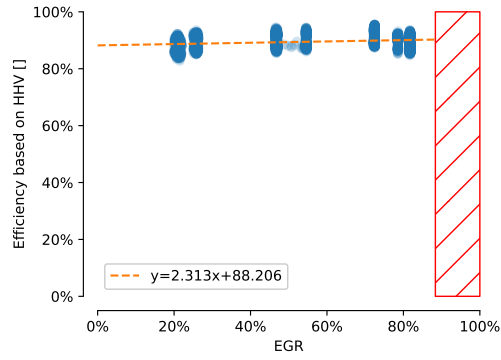


Figure 10: The efficiency slightly increases with the EGR rate has highlighted by the positive slope of the regression line (0.02% per additional EGR percentage).

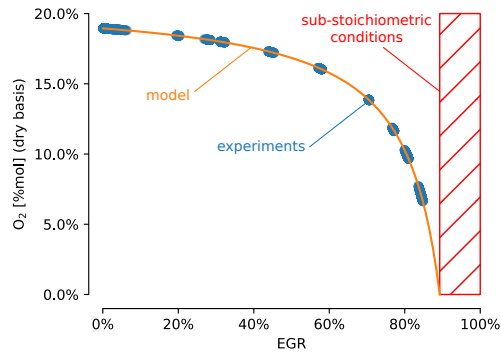


Figure 11: The dry molar composition of O₂ at the combustion chamber outlet follows the theoretical curves when EGR is applied.

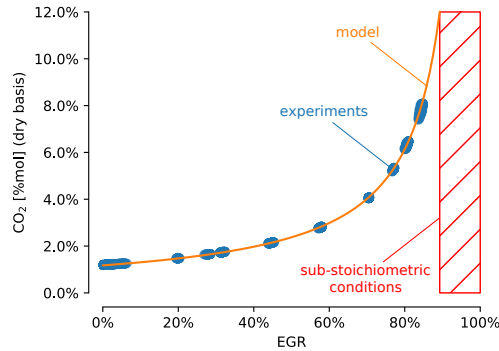


Figure 12: With the EGR level, the combustion conditions come closer to stoichiometry and the dry molar composition of CO_2 increases. The measurements in the exhaust gases (blue dots) follows the theoretical predictions (orange line). However, the recirculation rate cannot exceed 84%, as the mixture would be too rich, leading to flameout.

370 CO_2 content when no EGR is applied, the exhaust gases composition evolves
 371 to a maximum of 7.9% with 84% of EGR. Afterwards, additional recircula-
 372 tion leads to a flameout making higher CO_2 concentrations unachievable.

373

374 Due to the combustor architecture [17], a part of the incoming air is sep-
 375 arated from the main air and passes through dilution holes. A fraction of the
 376 O_2 entering the combustion chamber is thus not usable for the combustion
 377 explaining the limitation to a λ of 1.5. The ratio between the air entering the
 378 primary zone and the total incoming air has been estimated between 20%
 379 and 30%. The equivalence ratio λ in the primary zone is therefore between
 380 0.29 and 0.44, when the overall equivalence ratio is equal to 1.5.

381

382 The concentration of NO and NO_2 are measured in the exhaust duct by
 383 means of the Testo® 350 gas analyser. It can be observed in Figure 13

384 that EGR impacts the NO_x concentrations as well as their emissions by also
385 reducing the mass flow rate of exhaust gases. Due to the combustion cham-
386 ber temperature (more than 1000 °C), thermal NO is produced according to
387 Zeldovich’s mechanism, however the low residence time encountered does not
388 favour the conversion of NO into NO_2 . As expected, NO_2 is less present in
389 the exhaust gases (approx. 2 ppm) than NO (between 10 and 20 ppm).

390

391 Up to 50% EGR, the concentration of NO is increasing due to the fictive
392 additional residence time provided by EGR. Whereas the flow rate through
393 the combustion chamber remains constant, EGR decreases the flow rate of
394 fresh air by applying internal recirculation. In addition to the recirculated
395 NO_x , additional NO_x is added by the Zeldovich’s mechanism during combus-
396 tion increasing their concentration in the exhaust gases. Beyond 50% EGR,
397 the depletion in oxygen and nitrogen decreases NO_x formation.

398

399 When the reduction of exhaust flow is taken into account, EGR clearly
400 appears as an advantageous way of cutting NO_x emissions as represented
401 in Figure 13. While the concentration of NO_x slightly increases up to 50%
402 EGR, the decrease in mass flow rate leads to lower total emissions.

403

404 Moreover, the reduction of oxygen in the flue gases benefits to the level
405 of $\text{NO}_x|_{15\% \text{ O}_2}$. Indeed, as presented in Equation 6, the presence of the molar
406 fraction of O_2 (x_{O_2}) at the denominator further decreases the NO_x concen-

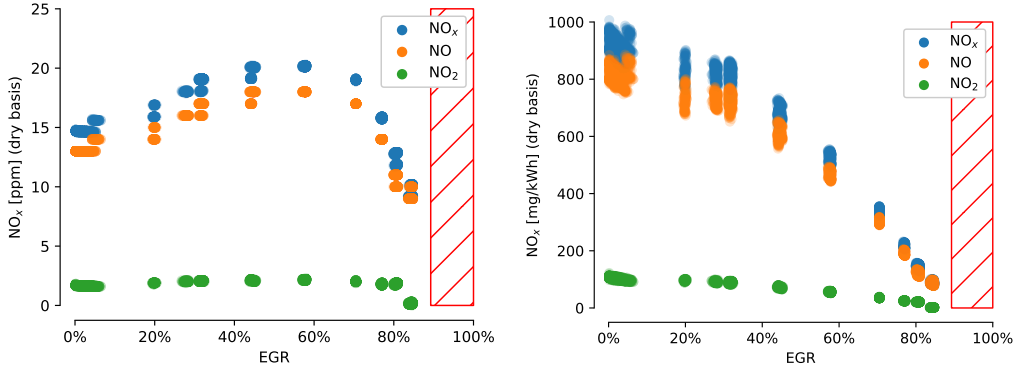


Figure 13: Due to the low residence time, NO_2 is less present in the exhaust gases than NO , however a maximum in NO concentration is encountered at around 50% EGR. Nevertheless, the decrease in mass flow rate leads to lower total emissions.

407 tration restored at 15% O_2 as presented in Figure 14.

$$\text{NO}_x|_{15\% \text{ O}_2} = \text{NO}_x|_{\text{dry}} \left(\frac{\text{O}_2, \text{ amb.}\% - 15\%}{\text{O}_2, \text{ amb.}\% - x_{\text{O}_2}\%} \right) \quad (6)$$

408 Due to mixing issues, a lack of O_2 is partially encountered in the chamber
 409 leading to incomplete combustion and production of CO . Up to 70% EGR,
 410 the CO concentration in the exhaust gases remains under 300 ppm. On
 411 that range, it can be observed in Figure 15 that the concentration increases
 412 slightly by reaching a plateau level at 70% EGR (± 32 ppm) roughly two
 413 times the value measured when no EGR is applied (± 16 ppm). From an
 414 EGR rate of 70%, the CO concentration evolves exponentially with higher
 415 EGR rate, rising up to 2700 ppm at 84%. The combustion architecture is
 416 composed of a primary zone in which enters roughly 30% of the incoming air.
 417 While the overall air equivalence ratio λ is still around 3.3 when 64% EGR

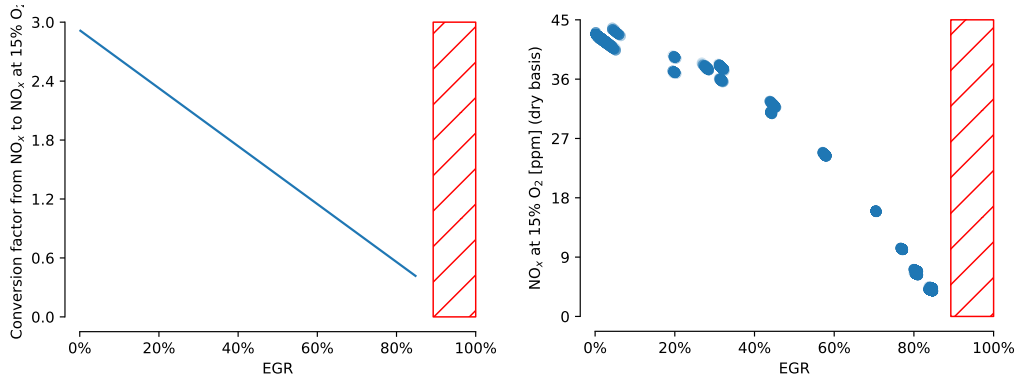
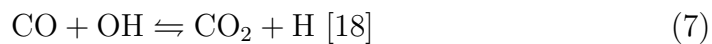


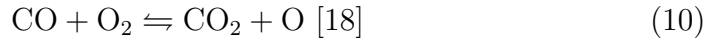
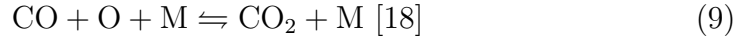
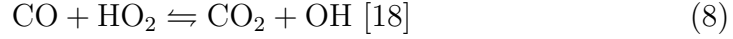
Figure 14: Environmental legislations often refer to NO_x emission referenced to 15% dry O_2 . Above 64% EGR, the conversion factor from NO_x to $\text{NO}_x|_{15\% \text{ O}_2}$ is lower than one, the application of EGR thus benefits to the NO_x reduction from a legislative point of view.

418 are applied, the stoichiometry is reached in the primary zone. The applica-
 419 tion of higher EGR rates therefore results in increased CO production, which
 420 is only partially burnt in the secondary zone due to the low temperature of
 421 the incoming air.

422

423 In addition to the lack of oxygen, the main cause of CO production, the
 424 dissociation of CO_2 has also been identified as a contributing factor to CO
 425 emissions [1]. The CO_2 present in the inlet composition thus also reacts
 426 through different mechanisms described by Masri et al. [18] in Equation 7,
 427 Equation 8, Equation 9 and Equation 10 to form CO. In Equation 9, M refers
 428 to a molecular third-body enhancing the efficiency of the reaction ($\text{M} = \text{H}_2\text{O}$,
 429 CO_2 , H_2 , CO , O_2 or N_2).





430 However, the application of EGR decreases the exhaust mass flow rate
431 and positively impacts the specific CO production (mg/kWh). As shown
432 in Figure 15, the amount of specific CO emitted decreases with EGR and
433 reaches a minimum at 70%, corresponding to 16% O₂ at the combustion
434 chamber inlet. This first tendency is explained by the slight increase of the
435 concentration in the exhaust gases counteracted by the exhaust flow decrease.
436 After 70% EGR, the specific CO emission increases due to the exponential
437 growth of its concentration. CO thus appears as a limiting factor for EGR.

438

439 **4. Future work**

440 The realized experiments have shown that EGR has the potential to in-
441 crease the CO₂ concentration and decrease the mass flow of flue gases while
442 keeping good performance. However, particular attention has to be made on
443 the CO emissions to ensure that the legal limits are respected. While the
444 results obtained already answer open questions from literature by experimen-

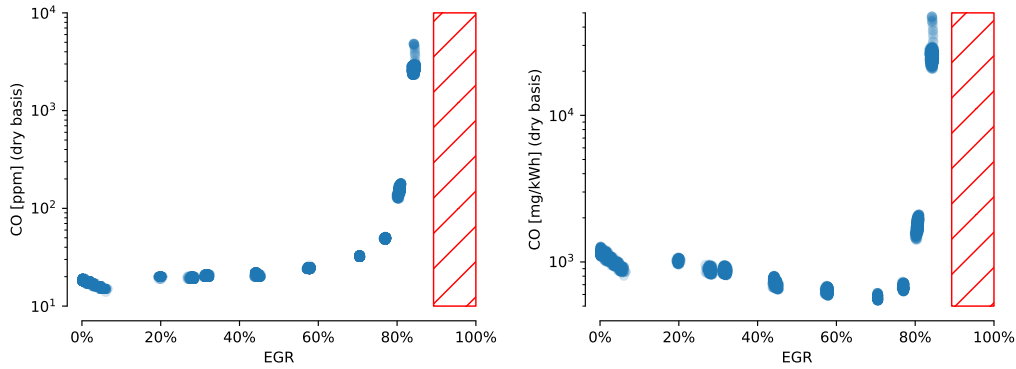


Figure 15: While EGR increases the specific CO concentration in the exhaust gases, a minimum in the CO emission is encountered at 70%, corresponding to the 16% dry O₂ at the combustor inlet.

445 tal evidence, further work should be planned to better understand the impact
 446 and limitations of exhaust gas recirculation.

447

448 Regarding the pressure levels, the combustor pressure is limited at 2.4 bar
 449 which is far from the range (15-30 bar) reached on an industrial gas turbine.
 450 Experiments at high pressure will therefore allow to better understand the
 451 influence of the pressure on CO and NO_x emissions when different rates of
 452 EGR are applied.

453

454 Tanaka et al. [7] have shown that the NO_x contained in the air have no
 455 effect on additional formation during the combustion. Nevertheless, EGR
 456 significantly increases the amount of NO_x at the combustion chamber inlet.
 457 Deeper investigations should be done to observe how NO_x are subject to
 458 the additional fictive residence time of EGR, to the O₂ depletion and CO₂

459 increase as well as by the reburning phenomenon [19]. In this phenomenon,
460 NO is intermediately converted in HCN before being converted back to N₂.

461

462 On the CO side, further analysis on the CO reburning phenomenon should
463 be realized to focus on how EGR can reduce CO by recirculating a fraction
464 on the incomplete products present in the exhaust gases through the com-
465 bustion chamber.

466

467 While EGR was so far applied in steady-state, no experiment has yet
468 tried a cold-start with the direct application of EGR. Some comparisons
469 could then be done on the emission levels and performances achieved when
470 different levels of EGR are directly applied at the cold start of the turbine.
471 Furthermore, transient aspects related to the recirculation of the exhaust
472 gases have also never been investigated.

473 **5. Conclusions**

474 The benefits of EGR have clearly been identified on both NO_x reduction
475 aspects, as well as for the carbon capture perspective due to the increase in
476 concentration of CO₂, the O₂ depletion and the mass flow reduction. How-
477 ever, the main limiting factors of EGR are the combustion instabilities and
478 the CO produced. While the numerical impact of EGR on the performance
479 of the mGT has been simulated in some papers, a clear lack of experimental
480 investigations was present.

481

482 In this paper, an innovative setup is presented where the MTT Ener-
483 Twin® has been modified with an external EGR loop recirculating gases
484 up to flameout. The experimental setup is a micro gas turbine designed for
485 combined heat and power generation purposes and fed with natural gas.

486

487 At the maximum EGR level (84%), the dry CO₂ concentration in the
488 exhaust gases reached 7.9%. Over the entire recirculation spectrum, CO₂
489 concentration increased according to the analytical expression. While the
490 NO_x concentration reaches a maximum of 20 ppm at around 50% EGR (due
491 to the fictive higher residence time), the specific emission (mg/kWh) keeps
492 decreasing with EGR due to a reduced total flue gas flow rate. On the CO
493 side, its constant augmentation in concentration is counteracted by the re-
494 duction of the mass flow rate up to 70% EGR, after which its emissions rise
495 sharply. Regarding the combined heat and power production, the applica-
496 tion of recirculation, with a lower than ambient recirculation temperature
497 increases both the power production and the heat recovered in the water
498 heater. This tendency can be explained by the higher mass heat capacity of
499 the flue gases and the control strategy ensuring a constant TOT, resulting
500 in a higher heat duty.

501

502 While the realized experiments address the need of experimental data on
503 the cycle performance when EGR is applied, some questions remain unan-

504 swered regarding to the NO_x and CO production at high pressure as well as
505 their reburning in the combustion chamber. These considerations will be the
506 subject of future work.

507

508 Even if the results of EGR observed on an mGT (i.e. the higher CO_2
509 content and lower exhaust mass flow rate) have not yet be extrapolated to
510 an industrial size, the application of EGR already appears as a clear pathway
511 to reduce the penalty of amine-based carbon capture unit. However, CO has
512 already been identified as the main limiting factor to higher recirculation
513 rate, requiring further development in combustor technologies.

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519 **7. Declaration of interests**

520 The authors declare that they have no known competing financial inter-
521 ests or personal relationships that could have appeared to influence the work
522 reported in this paper.

523 **8. Declaration of Generative AI and AI-assisted technologies in the**
524 **writing process**

525 During the preparation of this work the author(s) used chatGPT 4o in
526 order to improve readability and language of the text. After using this
527 tool/service, the author(s) reviewed and edited the content as needed and
528 take(s) full responsibility for the content of the publication.

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