

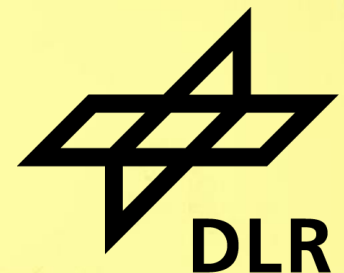
# STANDARDIZED TECHNO-ECONOMIC ANALYSIS OF PTX-FUELS

## SNG- and HSNG-production in Germany

DECHEMA Jahrestreffen der Fachsektion Energie, Chemie und Klima  
Frankfurt am Main, 12.03.2024

N. Heimann<sup>a</sup>, F. Moser Rossel<sup>a</sup>, R.-U. Dietrich<sup>a</sup>

<sup>a</sup> Deutsches Zentrum für Luft- und Raumfahrt e.V. (DLR), Stuttgart



# Outline

- Motivation
- Methodology
- SNG
- HSNG
- Comparison
- Conclusion and outlook



# Motivation

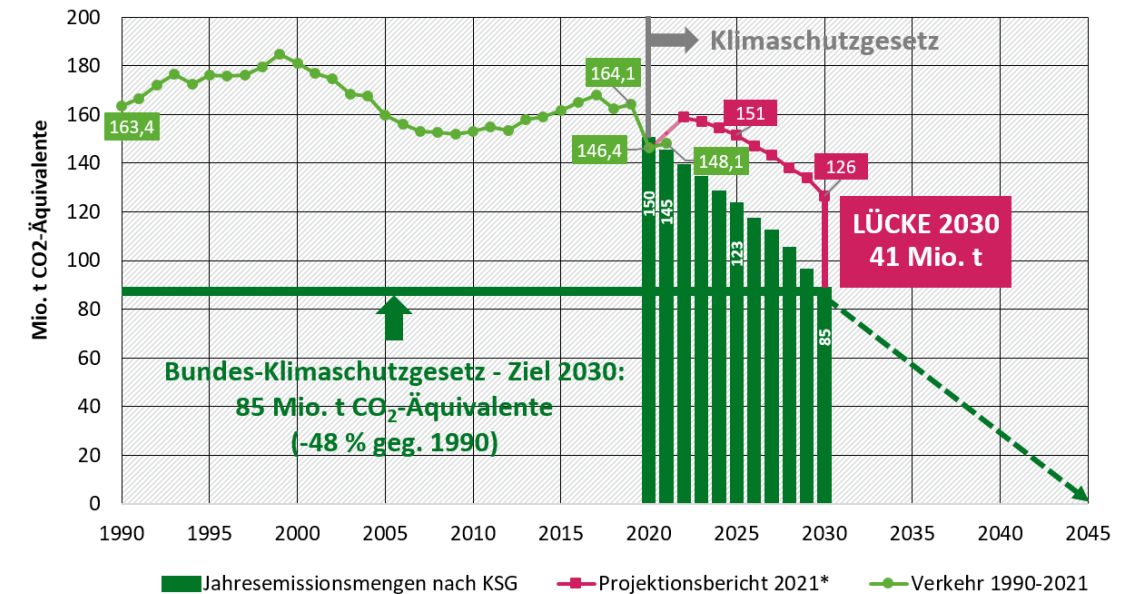
- EU Climate Neutral Goal by 2050 (Germany already by 2045)<sup>[1][2]</sup>
- Renewable transport lacks behind
- Hard to abate: chemical energy carriers



- But which?

Octanol  
 Butanol  
 Methane  
 FT-products  
 MeFo  
 OME  
 DME  
 DMC  
 Methanol  
 Hythane  
 Hydrogen  
 Ammonia  
 ???

Entwicklung und Zielerreichung der Treibhausgasemissionen in Deutschland im Sektor Verkehr des Klimaschutzgesetzes (KSG)



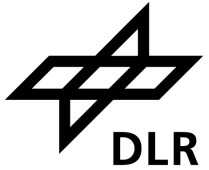
\* Berechnete Werte des „Projektionsbericht 2021“ (rote Linie, basierend auf Daten mit Stand August 2020) weichen für die Jahre 2020 und 2021 von den später veröffentlichten offiziellen IST-Werten (grüne Linie) ab.

Quelle: UBA 22.03.2022

[1] [European Climate Law \(europa.eu\)](http://europa.eu)

[2] Jedamzik et al. (2020) [Energiewende in Deutschland: Definition, Kosten & Ziele | co2online](https://www.co2online.de)

# Motivation



## Energy transition in the transport sector (EiV) - BEniVer: Scientific supervision



Begleitforschung Energiewende im Verkehr

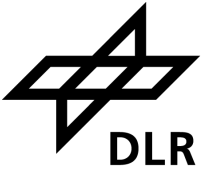
- EiV: funding 99 Mio. € | 16 projects | 100+ partner
- Renewable electricity based fuels

Cluster	Fuels in focus	Application
C3-Mobility	synth. Gasoline, DME, OME <sub>3-5</sub> , Methanol, Butanol, Octanol	
CombiFuel	Hythan (Hydrogen + Methane)	
E2Fuels	Methanol, OME <sub>3-5</sub> , Methan, Hythan	
FlexDME	Dimethylether (DME)	
ISystem4EFuel	synth. Diesel, OME <sub>3-5</sub>	
KEROSyN100	synth. Jet fuel	
LeanStoichH2	Hythan (Hydrogen+ Methane)	
MEEMO	Methanol	
MENA-Fuels	(Import strategies from MENA region)	
MethQuest	Methan, Methanol, Hydrogen	
NAMOSYN*	OME, Methylformiat (MeFo), Dimethylcarbonat (DMC)	
PlasmaFuel	synth. Diesel	
PowerFuel	synth. Jet fuel	
SHARC	(Smart energy management in harbors)	
SolareKraftstoffe	synth. Gasoline	
SynLink	synth. Diesel, synth. Jet fuel, Methanol	

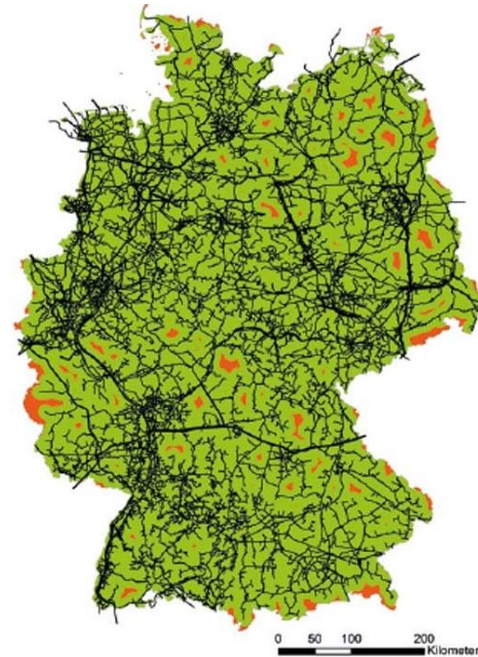
- Comparable assessment: BEniVer – Scientific supervision of „Energy transition in the transport sector (EiV)“



# Motivation: SNG



- Potential of SNG and HSNG:
  - Application proven
  - Little CO<sub>2</sub> needed per molecule
  - Existing infrastructure



Gas-grid >4 bar  
Image: DVGW



Image: Volvo Trucks

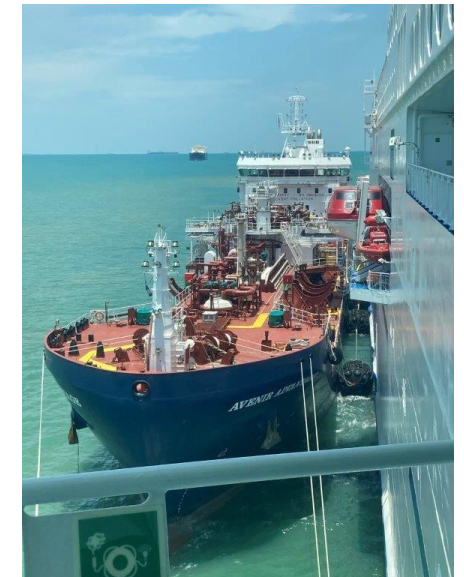


Image: Avenir LNG



Source: Wolfgang Jargstorff / Fotolia.com



Image: Daimler

# Motivation: HSNG

## Combifuel project of Graforce GmbH, Berlin

- Addition of H<sub>2</sub> to SNG → less CO<sub>2</sub> needed
- Up to 30 % H<sub>2</sub> tested



BEniVer

Begleitforschung Energiewende im Verkehr



# Motivation: HSNG

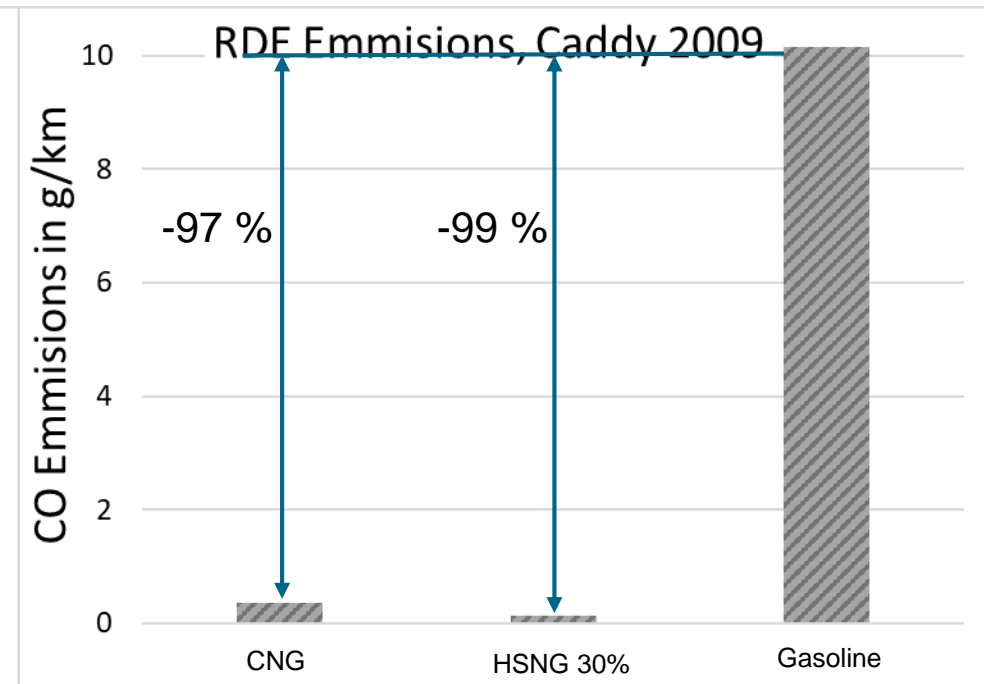
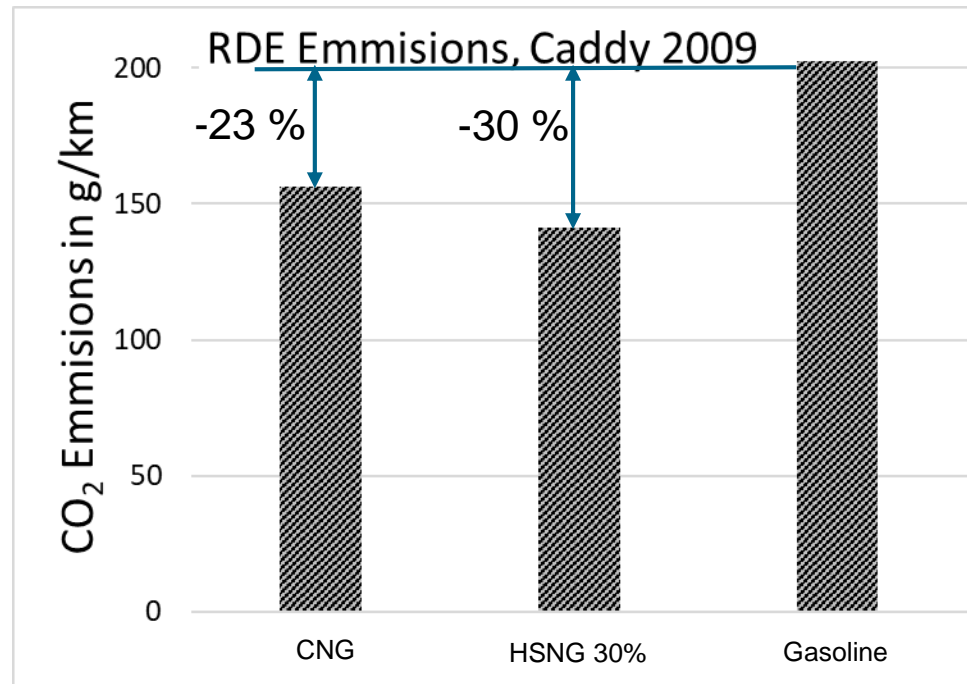
BEniVer

Begleitforschung Energiewende im Verkehr



## Combifuel project of Graforce GmbH, Berlin

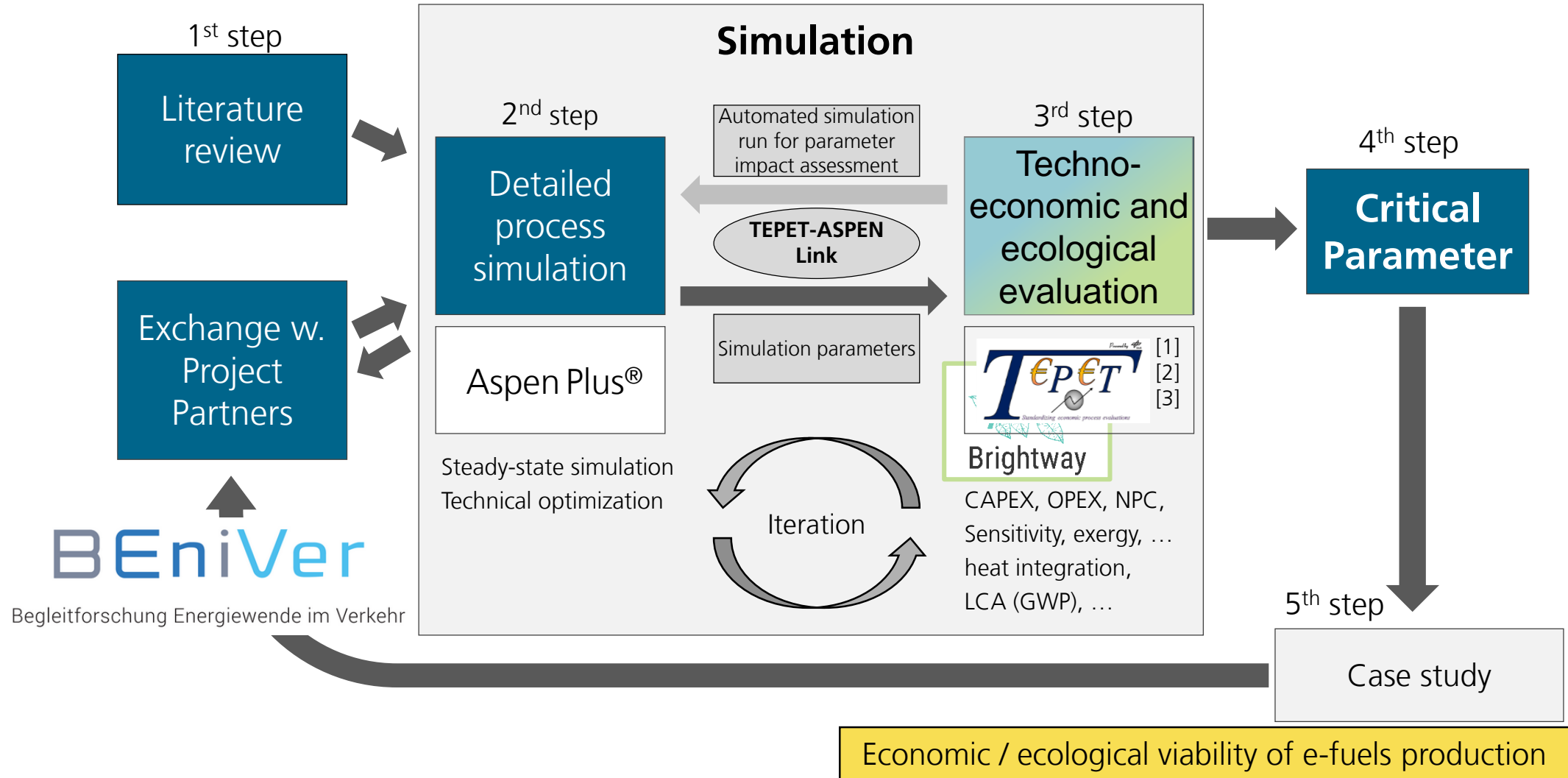
- Addition of H<sub>2</sub> to SNG → less CO<sub>2</sub> needed
- Up to 30 % H<sub>2</sub> tested
- Emission reduction: CO<sub>2</sub>, CO, HC, increase: NO<sub>x</sub>



# METHODOLOGY



# Assessment Workflow



**BEniVer**  
Begleitforschung Energiewende im Verkehr

[1] Albrecht et al. (2017). A Standardized Methodology for the Techno-Economic Evaluation of Alternative Fuels  
 [2] Maier et al. (2021) Techno-economically-driven identification of ideal plant configurations for a new biomass-to-liquid process  
 [3] Weyand et al. (2023) Process design analysis of a hybrid Power-and-Biomass-to-Liquid process

# Methodology: technical analysis

## Key performance indicator (KPI)

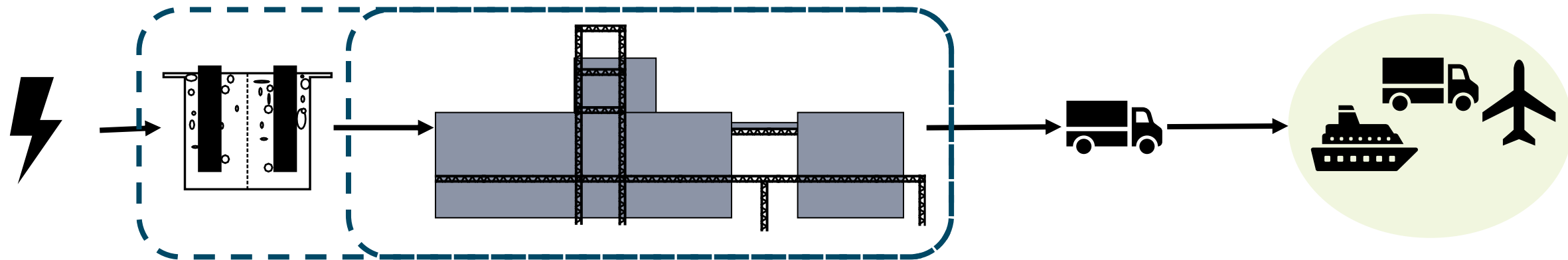
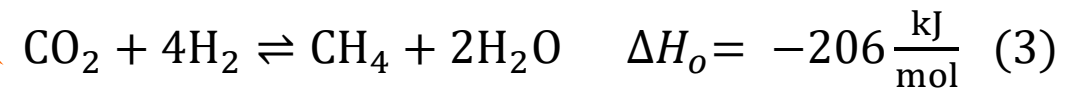
### KPI / efficiencies<sup>[1]</sup>

$$\eta_{PtF} = \frac{LHV_{Fuel} \cdot \dot{n}_{Fuel}}{\dot{P}_{el}}$$

$$\eta_{H_2tF} = \frac{LHV_{Fuel} \cdot \dot{n}_{Fuel}}{LHV_{H_2} \cdot \dot{n}_{H_2}}$$

4 moles H<sub>2</sub> converted into  
1 mole Methane  
 $\eta_{H_2tF, ideal} = 83.3\%$

### Methane synthesis reactions<sup>[2]</sup>



[1] Rahmat et al. (2023) Techno-economic and exergy analysis of e-MeOH production <https://doi.org/10.1016/j.apenergy.2023.121738>

[2] Rönsch, S., et al., Review on methanation – From fundamentals to current projects. Fuel, 2015. 166: p. 276-296.

# Methodology: Economical Analysis

Peters et.al. [1]

$$NPC \left[ \frac{\text{€}}{L} \right] = \frac{ACC + \sum OPEX_{ind} + \sum OPEX_{dir} + t_{labor} c_{labor}}{\frac{\dot{m}_{fuel}}{\rho_{fuel}} \cdot \frac{\omega_{fuel}}{\omega_{reference}}}$$

$$ACC \left[ \frac{\text{€}}{a} \right] = FCI \cdot \left( \frac{IR \cdot (1+IR)^y}{(1+IR)^y - 1} + \frac{IR \cdot y}{9} \right)$$

$$FCI = \sum_{i=1}^m EC_i \cdot \left( 1 + \sum_{j=1}^{10} F_{ind,i,j} \right) \cdot \left( 1 + \sum_{j=11}^{12} F_{ind,i,j} \right) \quad i, j \in \mathbb{N}$$

$$EC_i = f_i(S_{i,1}; S_{i,2}; \dots S_{i,k}) \cdot \left( \frac{CEPCI}{CEPCI_{ref}} \right) \cdot F_{pre,i} \cdot F_{mat,i} \cdot \left( 1 - L_i^{\log_2(n)} \right), \quad n, i, k \in \mathbb{N}$$

$$f_i(S_{i,1}; S_{i,2}; \dots S_{i,k}) = f_i(S_i) = EC_{ref,i} \cdot \left( \frac{S_i}{S_{ref,i}} \right)^{d_i}$$

Basic conditions	[3]
Base Year	2018
Location	Germany
Currency	€ <sub>2018</sub>
Electricity input (plant + electrolysis)	300 MW <sub>el</sub>
Full-load Hours	8,000 h/a
Electricity	56 € <sub>2018</sub> /MWh
H <sub>2</sub> cost	4,742 € <sub>2018</sub> /t
H <sub>2</sub> feed conditions	50 bar; 50 °C
CO <sub>2</sub> cost	69 € <sub>2018</sub> /t
CO <sub>2</sub> feed conditions	3 bar; 25 °C
Interest Rate	5 %
Labor cost	41 € <sub>2018</sub>
Plant lifetime	20 a

[1] M. Peters, K. Timmerhaus, R. West, 2004.

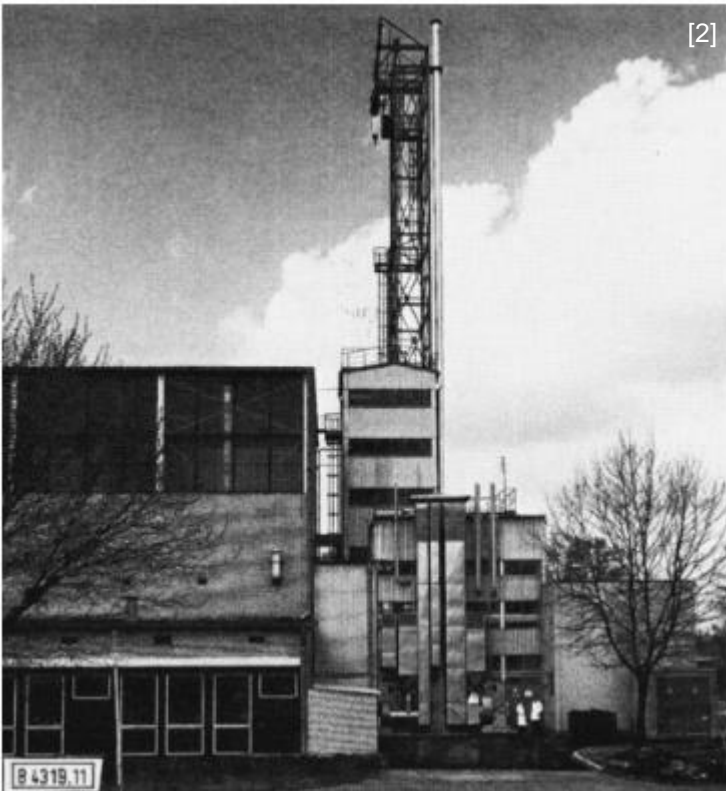
[2] F. G. Albrecht, D. H. König, N. Baucks, R.-U. Dietrich, 2017

[3] Heimann, N. et al (2023), submitted.

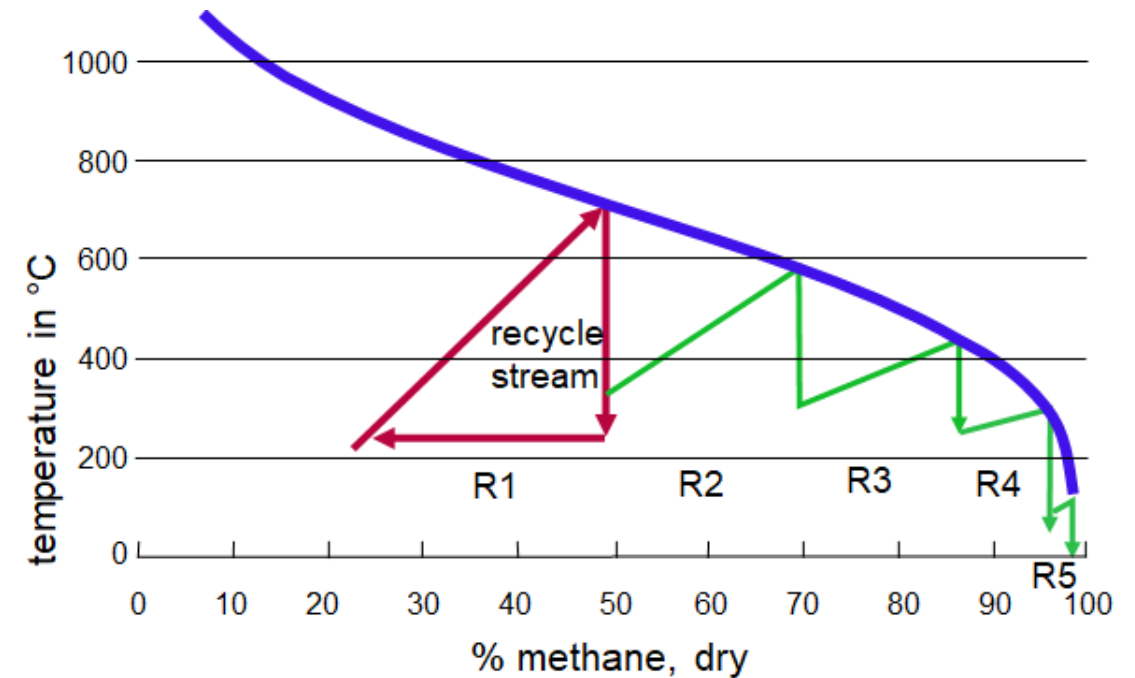
# SNG

# Large scale e-Methane production (SNG w. 98 vol.% CH<sub>4</sub>)

## Advanced TREMP™-process [1]



- Composition adjustment  
Transport: DIN EN 16723-2:2017-10  
Gas grid: DVGW G260
  - Polishing reactor & water removal



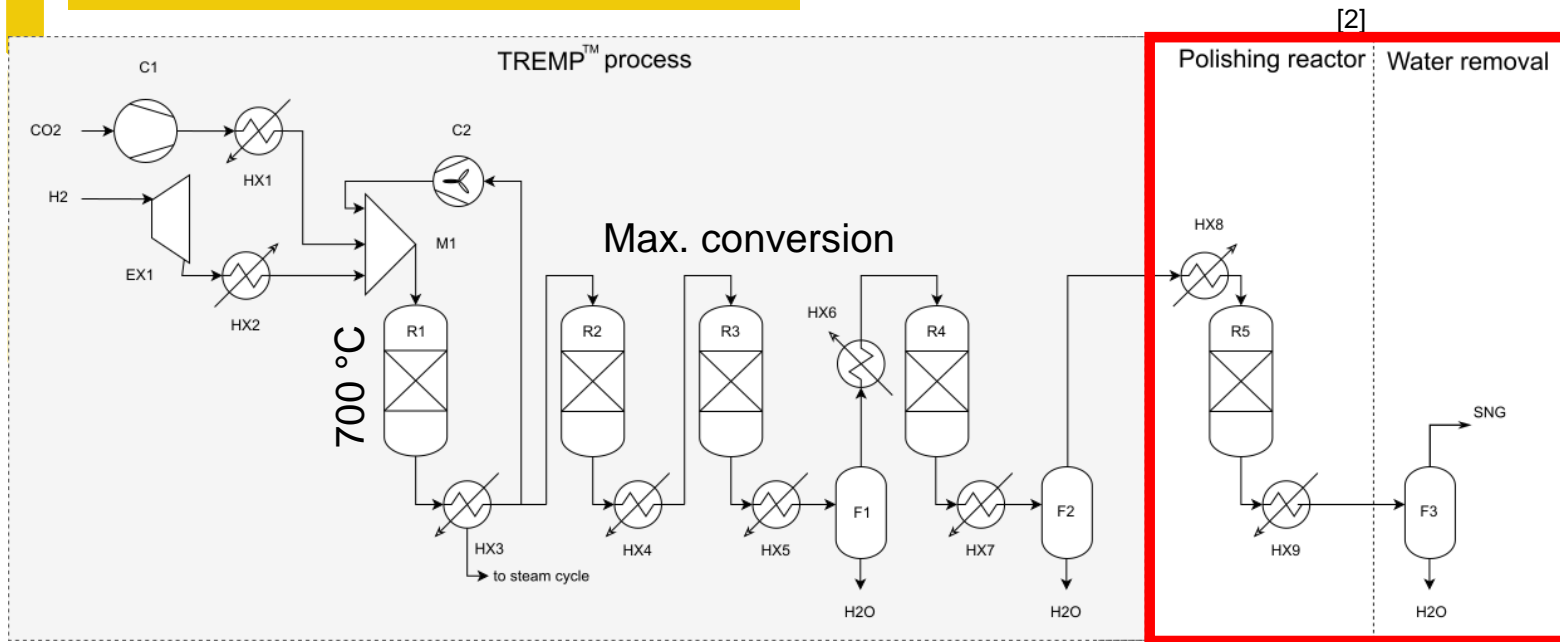
[1] Topsøe, H., From coal to clean energy. 2011

[2] Harms, H., B. Höhle, and A. Skov, 1980, Methanisierung kohlenmonoxidreicher Gase beim Energie-Transport. Heimann,

[3] N. et al., 2023, Standardized TEA of sCNG and HCNG, to be submitted

# Large scale e-Methane production (SNG w. 98 vol.% CH<sub>4</sub>)

## Advanced TREMP™-process [1]



- Composition adjustment  
Transport: DIN EN 16723-2:2017-10  
Gas grid: DVGW G260
  - Polishing reactor & water removal

	R1	R2	R3	R4	R5
t [°C]	700	545	400	428	427
GHSV	8900	8800	4200 <sup>[2]</sup>	4200 <sup>[2]</sup>	4200 <sup>[2]</sup>

[1] Topsøe, H., From coal to clean energy. 2011

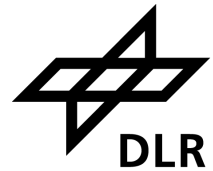
[2] Harms, H., B. Höhle, and A. Skov, 1980, Methanisierung kohlenmonoxidreicher Gase beim Energie-Transport. Heimann,

[3] N. et al., 2023, Standardized TEA of sCNG and HCNG, to be submitted

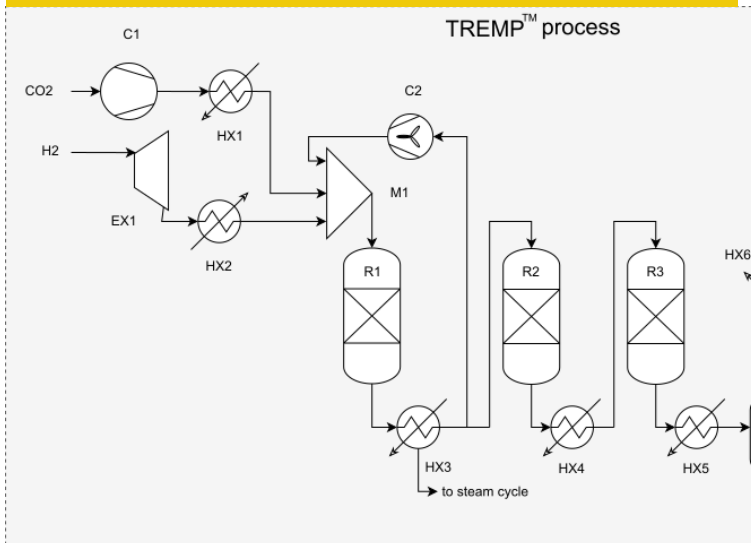
# Process description (SNG w. 98 vol.% CH<sub>4</sub>) Catalysis

BEniVer

Begleitforschung Energiewende im Verkehr



## Advanced TREMP™-process [1]



**Reaction kinetic:** Rönsch et.al. [3-5]  
Combination of WGS and CO-Methanation

AspenPlus® model: RPlug

$$T_{\max,R1}^{[3]} = 700 \text{ °C}$$

$$T_{\text{in},R1} = 250 \text{ °C}$$

$$p^{[2]} = 20\text{-}30 \text{ bar}$$

$$L_{\text{reactor}} = 2 \text{ m}$$

$$D_{\text{reactor}} = f(\text{GHSV})$$

$$\text{GHSV}^{[2]} = 4200\text{-}8900$$

Pressure drop: Ergun's equation

Catalyst<sup>[2]</sup> MCR-2X (22 % Ni)

Bulk density<sup>[2]</sup> = 1000 kg m<sup>-3</sup>

Bed voidage<sup>[2]</sup> = 0.3

Lifespan<sup>[5]</sup> = 1 year

**Color coding process parameters:**

Blue → taken from literature

Green → own assumption/calculation

[1] Topsøe, H., From coal to clean energy. 2011

[2] in range of: Harms, H., B. Höhle, and A. Skov, 1980, Methanisierung kohlenmonoxidreicher Gase beim Energie-Transport.

[3] Rönsch et al., 2016, Review on methanation – From fundamentals to current projects.

[4] Klose, J., 1984, Kinetics of the methanation of carbon monoxide on an alumina-supported nickel catalyst. Journal of Catalysis

[5] Zhang, J., et al., 2013, Kinetic investigation of carbon monoxide hydrogenation under realistic conditions of methanation of biomass derived syngas

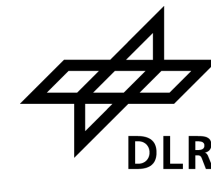
[6] Meylan et al., 2016, Material constraints related to storage of future European renewable electricity surpluses with CO2 methanation

# Results KPI

SNG w. 98 vol.% CH<sub>4</sub>

BEniVer

Begleitforschung Energiewende im Verkehr

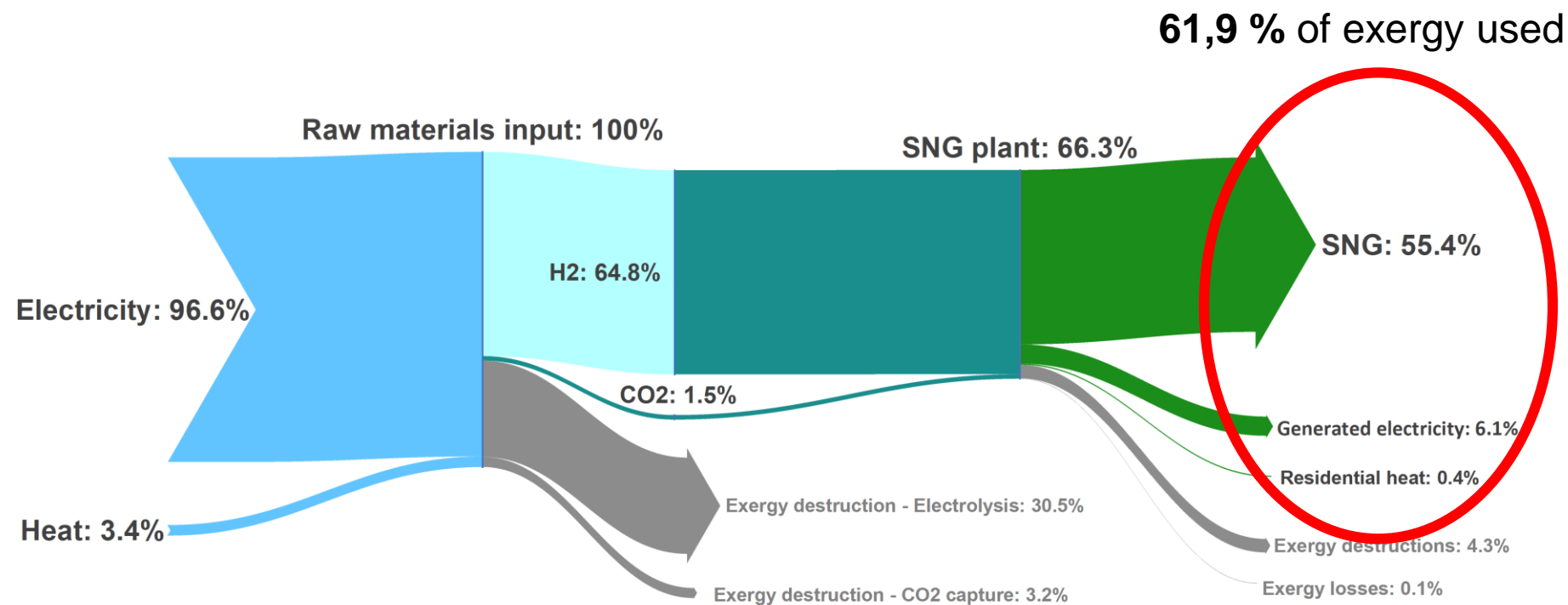


## KPI / efficiencies

$$\eta_{H_2tF} = 82 \%$$

$$\eta_{PtF} = 57 \%^*$$

## SNG production exergy flow



\*Electrolysis combined efficiency 69.2 % P<sub>el</sub>/LHV:

assumed: 1/3 PEM, AEL, SOEC each<sup>[1-3]</sup>

- Exergy reuse: steam-cycle and residential heat
- Highly exergy efficiency optimized

[1] Noack et al., 2015  
 [2] Jansen et al., 2018, p. 36  
 [3] BEniVer assumptions

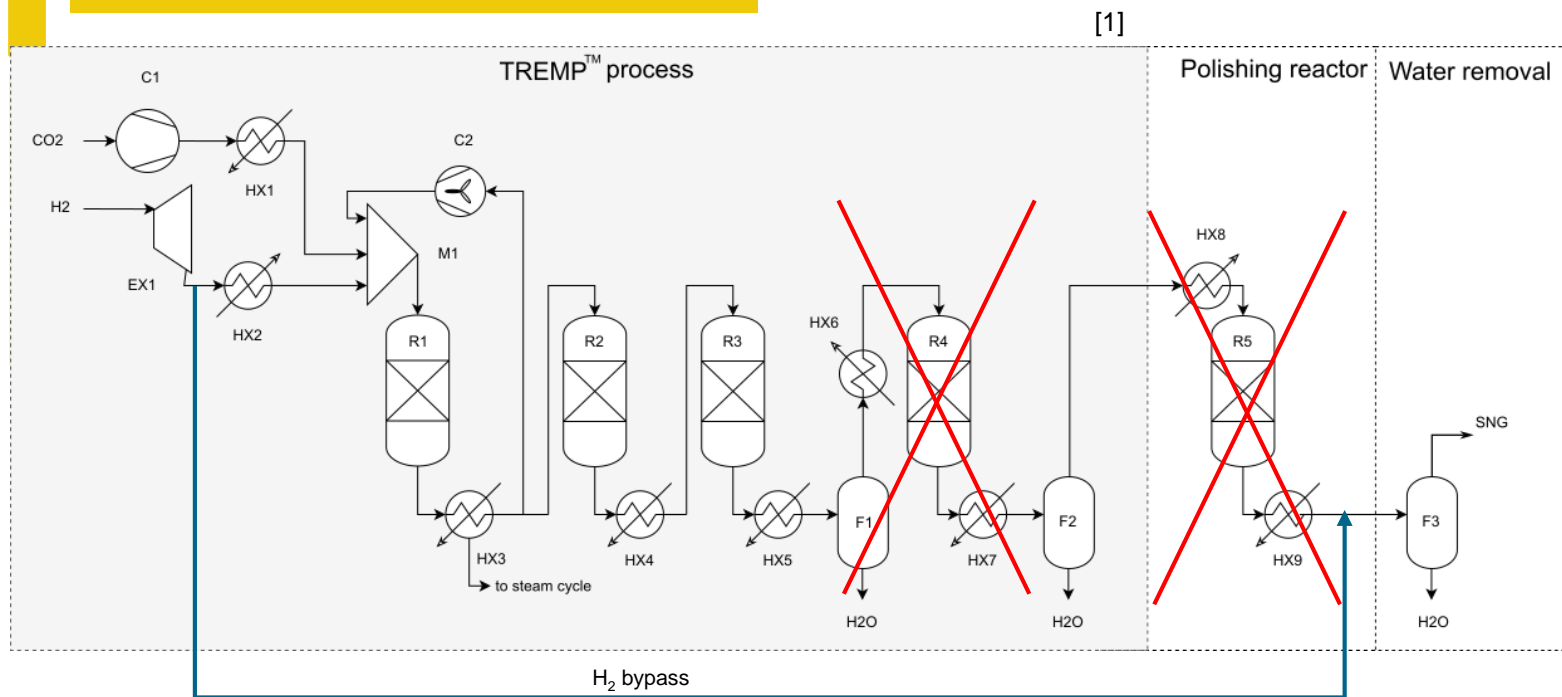


# HSNG

# Large scale HSNG production

## HSNG w. 30 vol.% H<sub>2</sub>

### Adapted TREMP™ process [2]



- High temperature in R1
  - Steam cycle
- Composition adjustment  
30 vol.% H<sub>2</sub> content → HSNG-30
  - Number of reactors reduced
  - Partial H<sub>2</sub> bypass
  - Smaller reactors for same output
  - less H<sub>2</sub>O production

Assumptions in the simulation:

- No impurities
- No side reactions

[1] Rönsch, S., et al., 2016

[2] Heimann, et al 2023, to be submitted

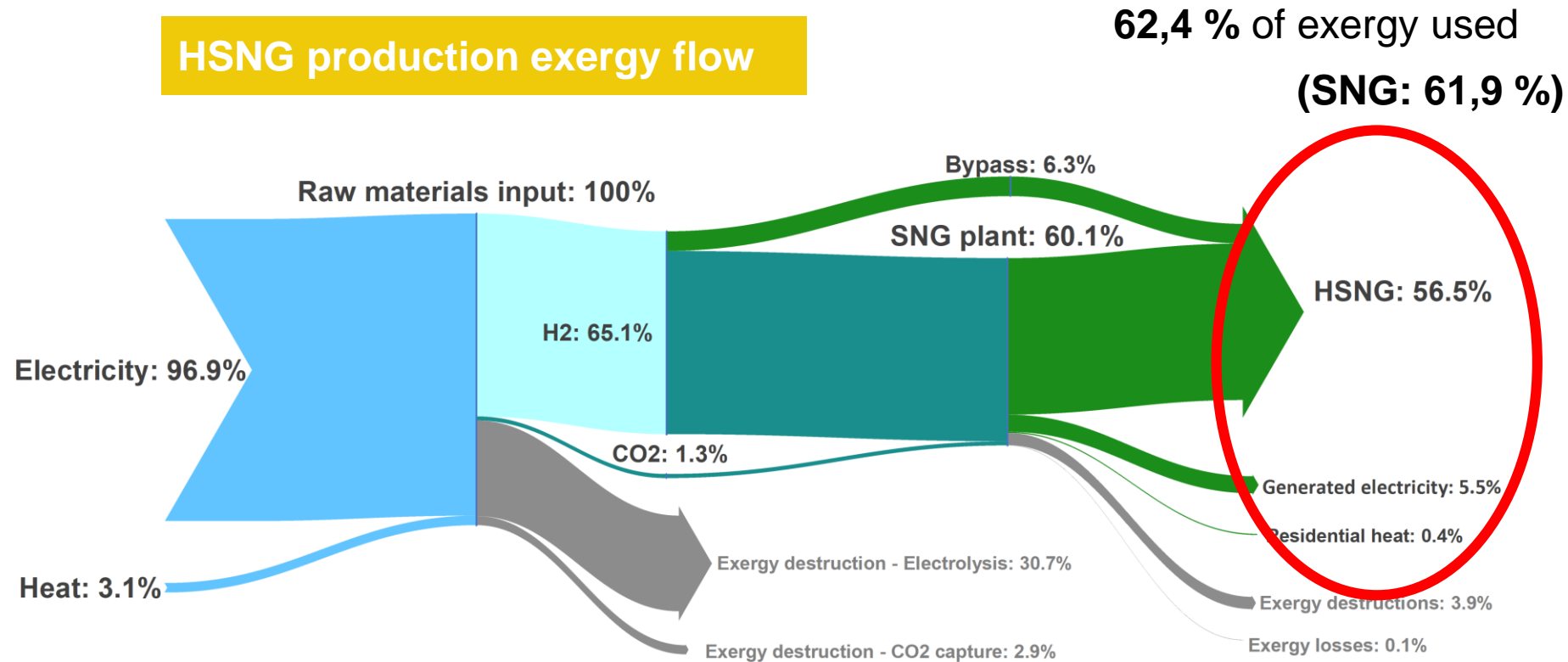


# Large scale e-Hythane production (HSNG-30)

## KPI / efficiencies

$$\eta_{H_2tF} = 89 \%$$

$$\eta_{PtF} = 58 \%^*$$



\*Electrolysis combined efficiency 69.2 %  $P_{el}/LHV$ :

assumed: 1/3 PEM, AEL, SOEC each<sup>[1-3]</sup>

- 1.1 % more power to fuel than SNG
- 5.5 % reused in steam-cycle (compared to 6.1 %)

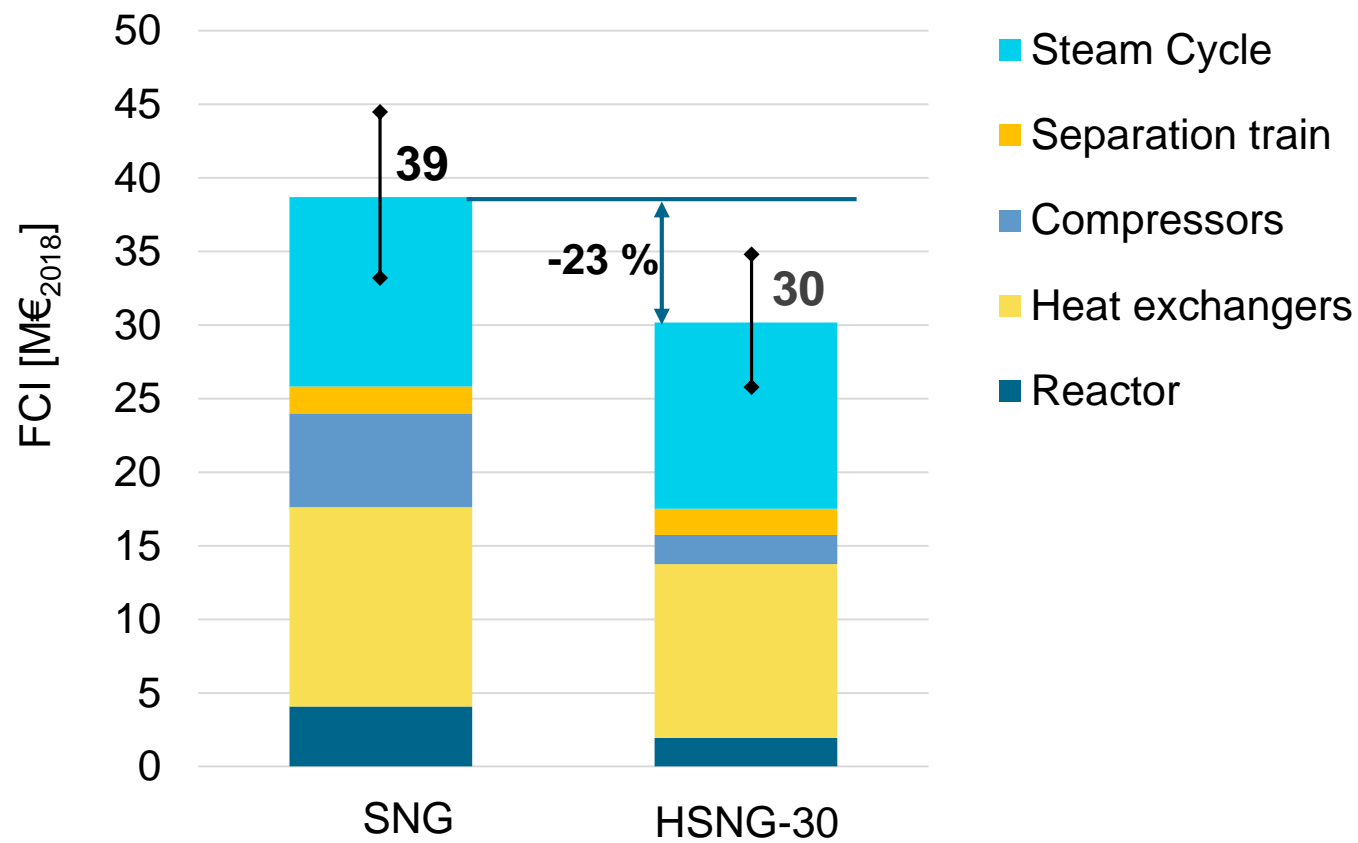
[1] Noack et al., 2015  
 [2] Jansen et al., 2018, p. 36  
 [3] BEniVer assumptions

# ECONOMIC ANALYSIS

# Results: Fixed Capital Investment

BEniVer

Begleitforschung Energiewende im Verkehr

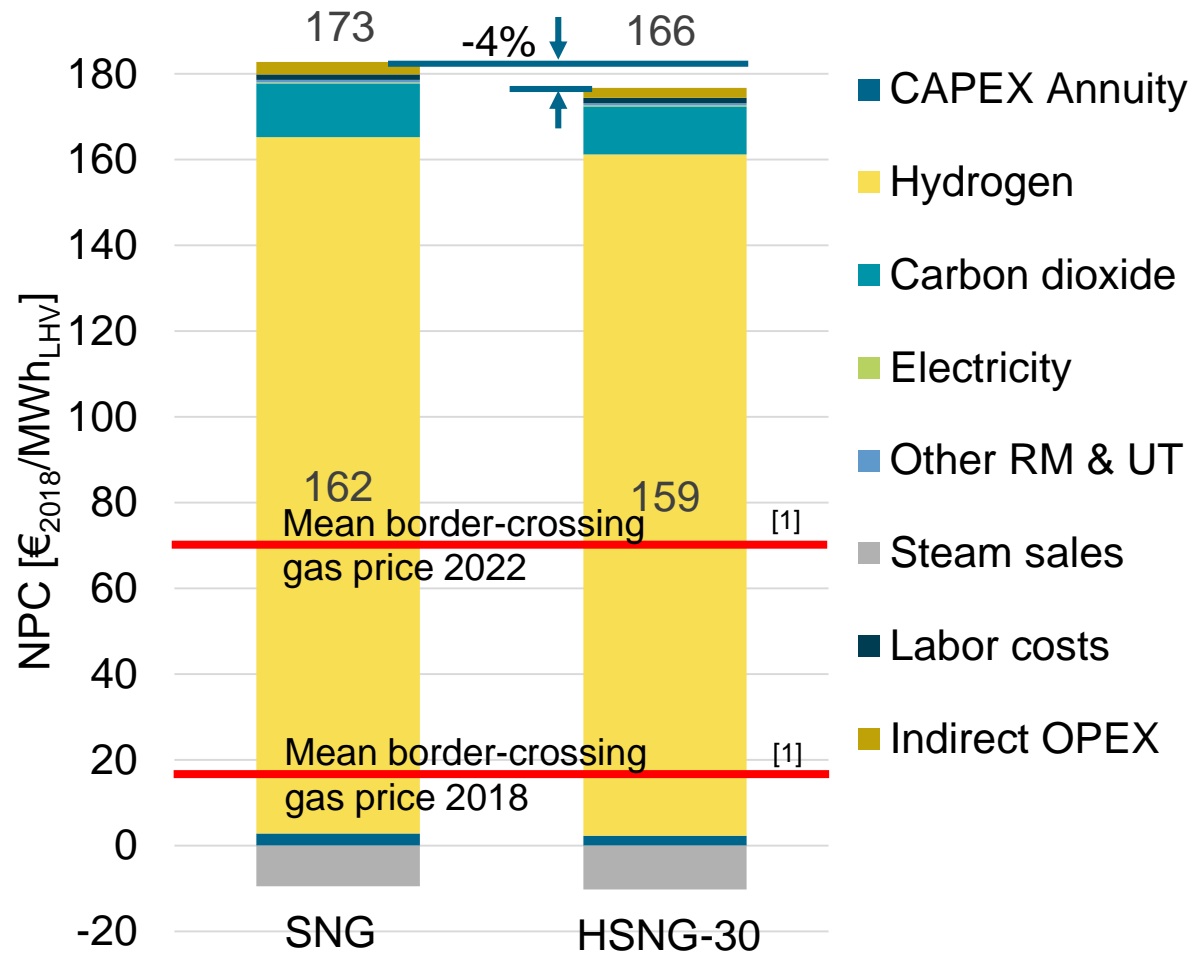


FCI reduction for Hythan30 compared to SNG → 23 %

Significant reduction in compressors and reactors

Steam cycle, heat-exchangers remain significant FCI

# Results: Net Production Costs electrolyzer excluded

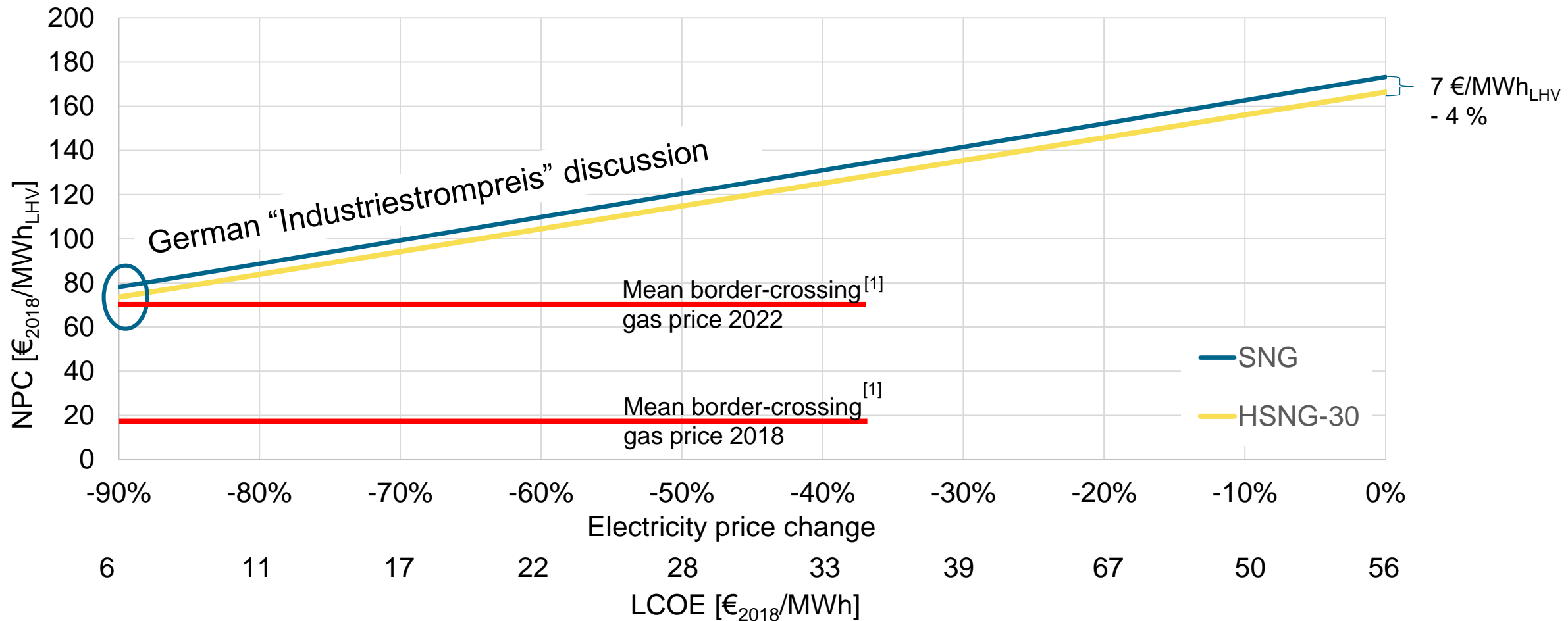


- NPC reduction for Hythan30 compared to SNG: ➔ 4 %
- Less H<sub>2</sub> needed
- >8 fold natural gas price

# Results: Sensitivity of NPC

BEniVer

Begleitforschung Energiewende im Verkehr



# Results: Comparison of e-fuels

## Comparing generic fuels / designer fuels

	SNG	HSNG-30	MeOH	FT	OME <sub>3-5</sub>	DMC	MeFo
$\eta_{PtF}$ [%]	57	58	53	40	42	47	52
NPC [€ <sub>2018</sub> /MWh <sub>LHV</sub> ]	173	166	204	321	360	329	298
Application parameter examples	<ul style="list-style-type: none"> <li>• Heavy truck</li> <li>• Drivetrain retrofit</li> <li>• ...</li> </ul>	<ul style="list-style-type: none"> <li>• Combifuel</li> <li>• Heavy truck</li> <li>• Drivetrain retrofit</li> <li>• ...</li> </ul>	<ul style="list-style-type: none"> <li>• Used in China</li> <li>• Low vapor pressure</li> <li>• Further conversion in Europe?</li> </ul>	<ul style="list-style-type: none"> <li>• Certified sustainable jet fuel</li> <li>• ...</li> </ul>	<ul style="list-style-type: none"> <li>• Better combustion</li> <li>• Blending ratio?</li> <li>• ...</li> </ul>	<ul style="list-style-type: none"> <li>• Better combustion</li> <li>• Blending ratio?</li> <li>• ...</li> </ul>	<ul style="list-style-type: none"> <li>• Better combustion</li> <li>• Blending ratio?</li> <li>• ...</li> </ul>

HSNG-30: highest efficiency cheapest e-fuel of EiV  
 Ecological assessment still pending  
 Application assessment started



## Conclusion

- Renewable non-electrical transport : SNG and HSNG preferable
- 300 MW production plant to small for German demand
- German grid electricity not 100 % renewable

### + SNG:

- High efficiency
- Existing technology
- Existing infrastructure

### + Hythan

- Higher efficiency
- Better performance in combustion



## Outlook

- Cheap renewable electricity needed
  - Political will needed
- ➔ Outlook: Identical HSNG spec. for both heat and transport applications  
**Transparent, standardized DLR assessment methodology available**

# THANK YOU FOR YOUR ATTENTION!



Contact:  
**Nathanael Heimann, M.Sc.**  
Research fellow Techno-economic Analysis  
DLR Institute of Engineering Thermodynamics  
[Nathanael.Heimann@dlr.de](mailto:Nathanael.Heimann@dlr.de)



**Dr.-Ing. Ralph-Uwe Dietrich**  
Group leader Techno-Economic Analysis  
DLR Institute of Engineering Thermodynamics  
[Ralph-Uwe.Dietrich@dlr.de](mailto:Ralph-Uwe.Dietrich@dlr.de)

