HYDROGEN PRODUCTION VIA THERMOCHEMICAL CYCLES USING SOLAR PARTICLE REACTORS

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Helmholtz Energy Conference 2023 | 12 - 13 June 2023 | Koblenz





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Solar Thermochemical Redox Cycle



Solar Thermochemical Redox Cycle



image adapted from [Richter2017]

Solar Thermochemical Redox Cycle





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Reactor and System Design Concepts



Redox material must be cycled between

- High (1450°C) and low (900°C) temperatures \rightarrow heat recovery required 1.
- Atmospheres with low oxygen partial pressures and steam atmospheres 2.
 - either purge gas or vacuum for low p_{02}
 - heat/power demand for steam / purge gas / vacuum generation important

Concentrated Solar Radiation

Move the material or change its surrounding conditions?





Fixed Redox Monoliths

Solar Thermochemical Hydrogen Production State of the Art: Batch Reactors with Fixed, Porous Monoliths of Redox Material





Source: Plataforma Solar de Almeria (Owned by the Spanish research centre CIEMAT)

Source: DLR



Source: IMDEA / SUN-to-LIQUID project



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Advantages of Moving Particle Reactor Systems



- Continuous and more flexible operation
- Less sensible heat losses, less thermal stresses
- Easy replacement of redox material
- Better heat transfer & better heat recovery



DOE-STCH Project (2015-2017) Vacuum Reactors with Moving Ceria Particles

initial concept







image: Grobbel2017 Ceria (CeO₂) Particles

- Sauter diameter 277 µm
- Density 6.6 g/cm³

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Ongoing Work of Sandia and DLR

Main goals:

- Improve vacuum receiver-reactors
- Prove pressure separation by packed bed between two receivers
- Collect more operational data in DLR Synlight[®]





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Modeling Solar Particle Receivers





Discrete Element Method (DEM) coupled to CFD

- Particle motion
- Heat transfer
- Reactions



Challenges (some of them):

- Moving particles at very high temperatures (1500°C)
- Wear on particles and components
- Gas flow control through particle beds (uniformity, pressure drop, ...)

Outlook:

- Prototype testing of vacuum reactor system in DLR Synlight
- Modeling of new SOMPIHR concept (i.e. CFD-DEM models)
- Prototype design and construction



- Concentrated solar energy can drive high-temperature chemical redox reactions to produce hydrogen or syngas (w/o producing electricity first)
- Reactor efficiencies can be increased significantly by moving the redox material instead of changing its surrounding conditions (temp. / p_{O2})

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