



# Study and Comparison of Naturally-Aged and As-Received Silvered-Glass Reflectors

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# Background

## Silvered-Glass reflectors at the Masdar Institute Solar platform (MISP)



Heliostat field and Beam-Down Tower at the Masdar Institute Solar Platform (MISP)

Installed July 2009

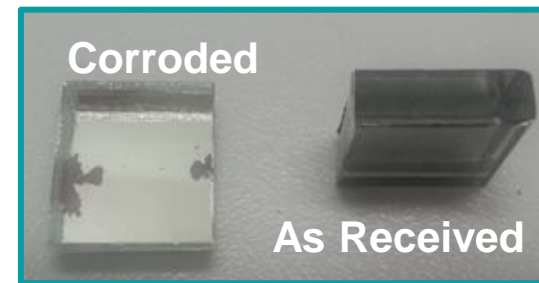


Corroded facet selected for analysis

> 10 pp specular reflectance drop  
after 7 years exposure

# Motivation

- Availability of heavily degraded mirrors after 7 years of outdoor exposure and as-received ones from the same original manufacturing batch:
  - ⇒ Possibility of comparison of as-received and outdoor-exposed samples :  
Understanding corrosion causes



- ⇒ Possibility of comparison of samples under accelerated aging and “naturally”-aged ones:  
Allows evaluation of accelerated-aging protocols

\* Presentation by Johannes Wette @ 5:50 , same room

# Objectives

**Understanding the cause of rapid corrosion observed in  
back-silvered glass mirrors exposed for 7 years at the MISIP, Abu Dhabi**

- ⇒ Characterization of exposed and corroded mirrors, and as-received ones using Scanning Electron Microscopy (SEM) with Energy Dispersive X-ray Spectroscopy (EDS)
- ⇒ Same analysis performed on as-received samples of a state-of-the-art commercial mirror as a reference for comparison

# Back-silvered Glass Mirrors Corrosion Mechanisms

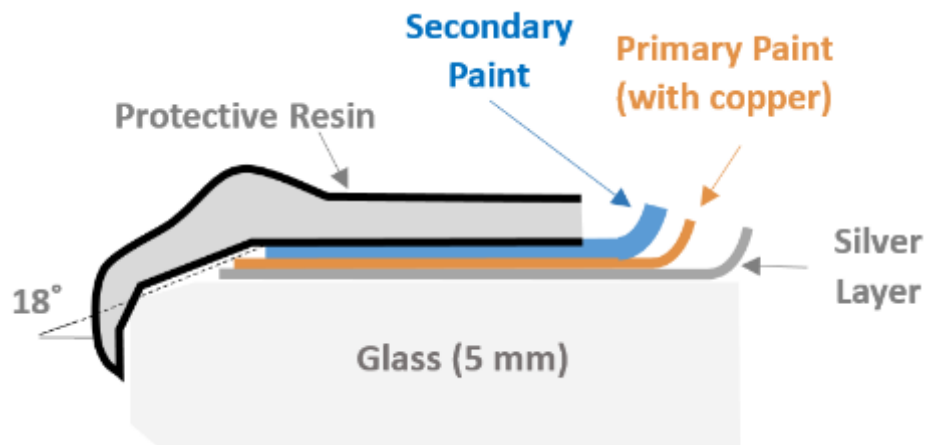
- Silver corrosion (tarnishing) can happen in presence of **salts** (mainly chloride), **ozone** and **sulfur**
- Silver vulnerable to air pollutants: acids, ammonia, hydrogen sulfide
- Corrosion influenced by environmental factors:  
pH, humidity, radiation, wind velocity, temperature
- Main corrosion products: **AgCl** , **AgS**.       $\text{Ag}_2\text{O}$  and  $\text{Ag}_2\text{SO}_4$  can also be found

⇒ Backside protective coatings necessary to prevent corrosion:

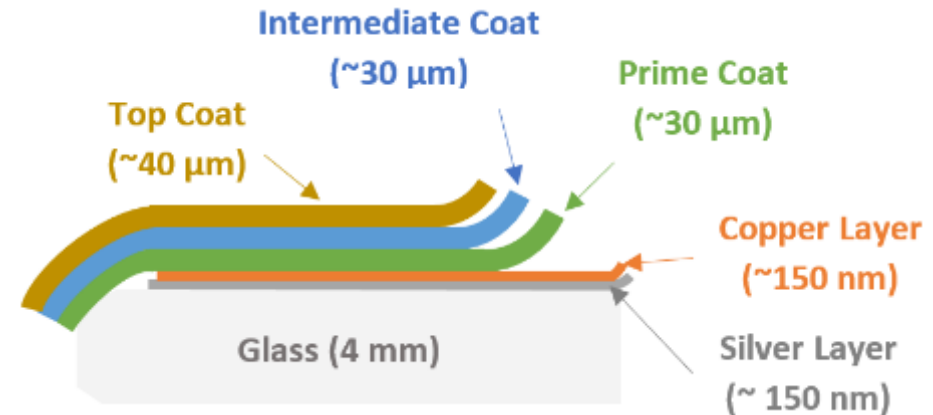
- Diffusion barrier to aggressive ions and humidity
- Protection from UV-radiation
- Protection from erosion

# Materials Description

- **Nishio Glass Mirror from MISP:** as-received and exposed, *Nishio\_AR* and *Nisho\_Exp*
- **Reference mirror:** State-of-the-art commercial back-silvered mirror, *Ref\_AR*



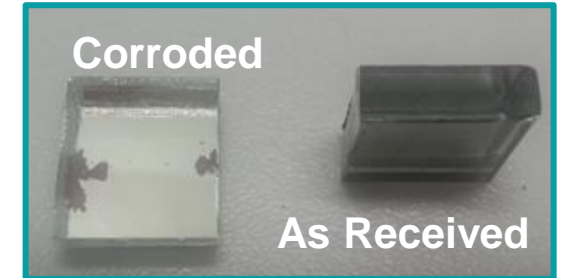
Reflector at the MISP by *Nishio Glass Mirror*  
(*Nishio\_AR*)



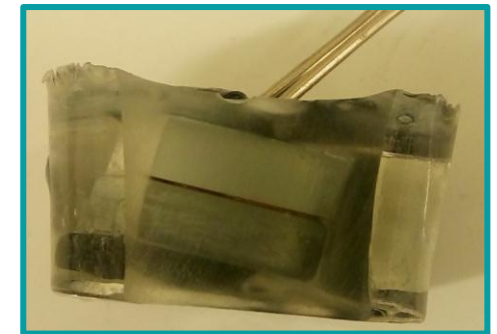
Commercial mirror for reference  
(*Ref\_AR*)

# Sample preparation

- Samples of 1 x 1 cm<sup>2</sup> cut and cross-section polished
- Polishing in multiple steps with decreasing abrasive sizes on a rotating disc (min 0.25 µm)
- 2 polishing trials:
  - 1- Direct polishing  
=> coatings detachment for the Nishio mirror even on the as-received mirror
  - 2- "Face-to-face" preparation with resin embedding



↓ Embedding



↓ Polishing





# Instrumentation

- FEI Quanta 250 SEM in combination with X-ray detector for EDS elemental analysis
- Backscattered Electrons Detector (BSED) used with SEM to provide contrast between different elements ( Z contrast)
- Surveyed elements: Silver **Ag**, Copper **Cu**, Chlorine **Cl**, Sulfur **S**, Oxygen **O**

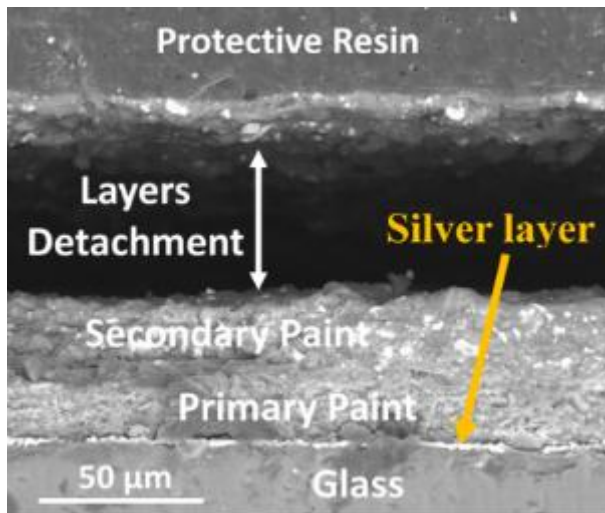




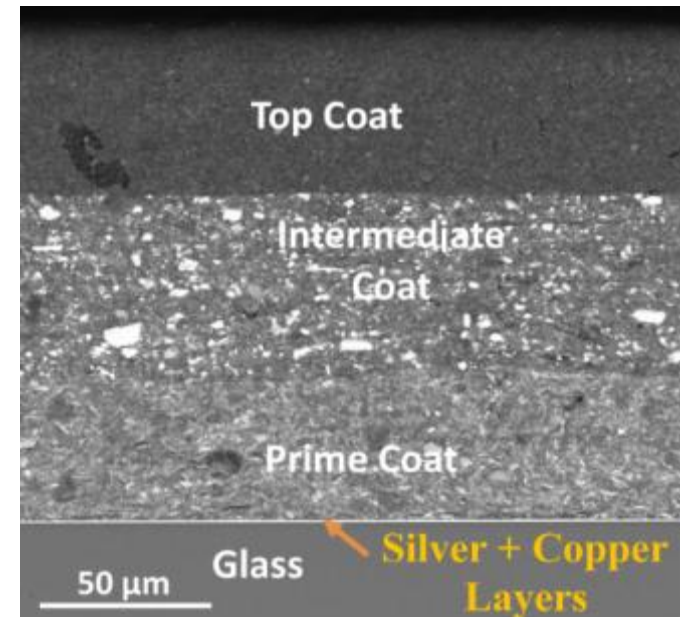
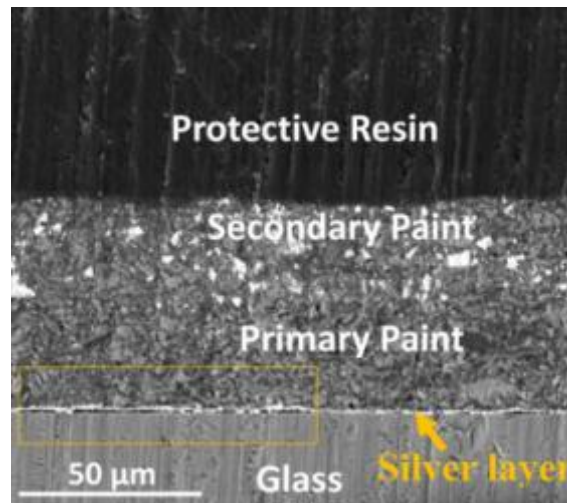
# Results – AR samples

## Backside coatings comparison: Nishio and Ref

*SEM images with BSED => Z contrast*



Nishio\_AR

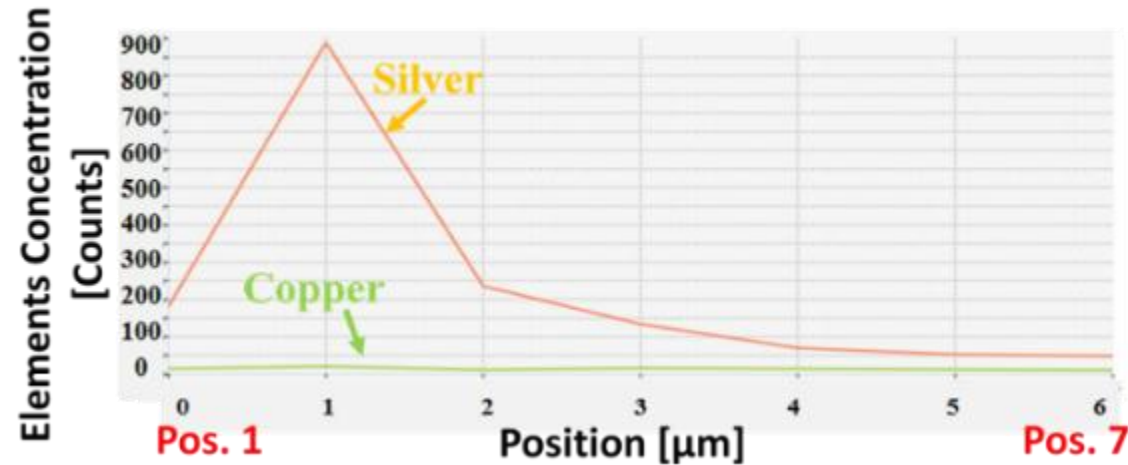
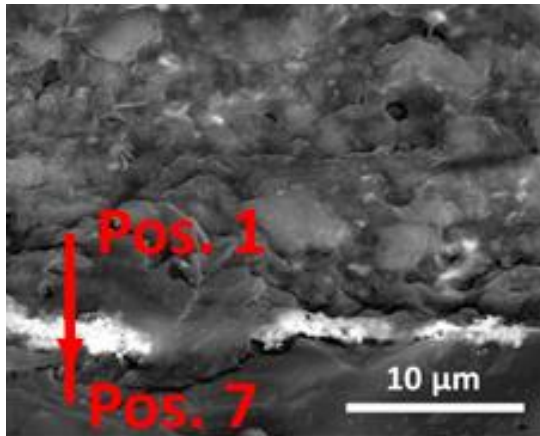


Ref\_AR

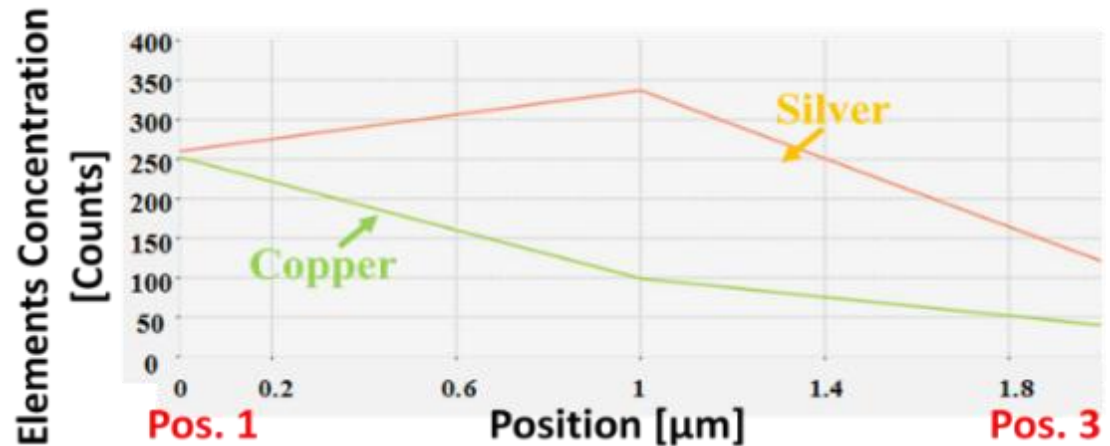
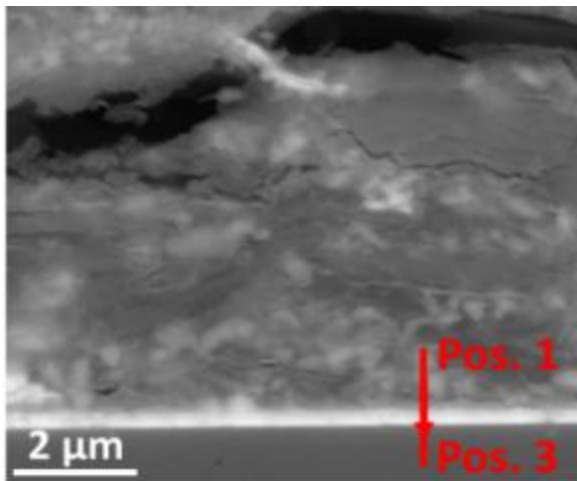
# Results – AR samples

## Copper protective layer comparison – EDS line scan

Nishio\_AR

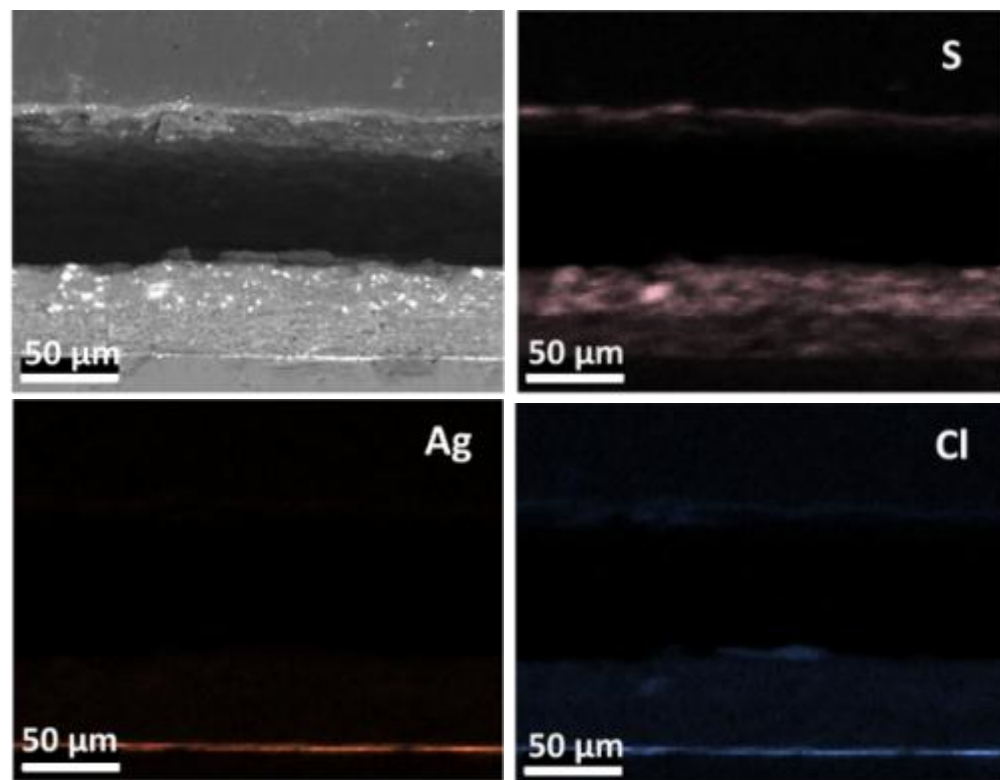


Ref\_AR

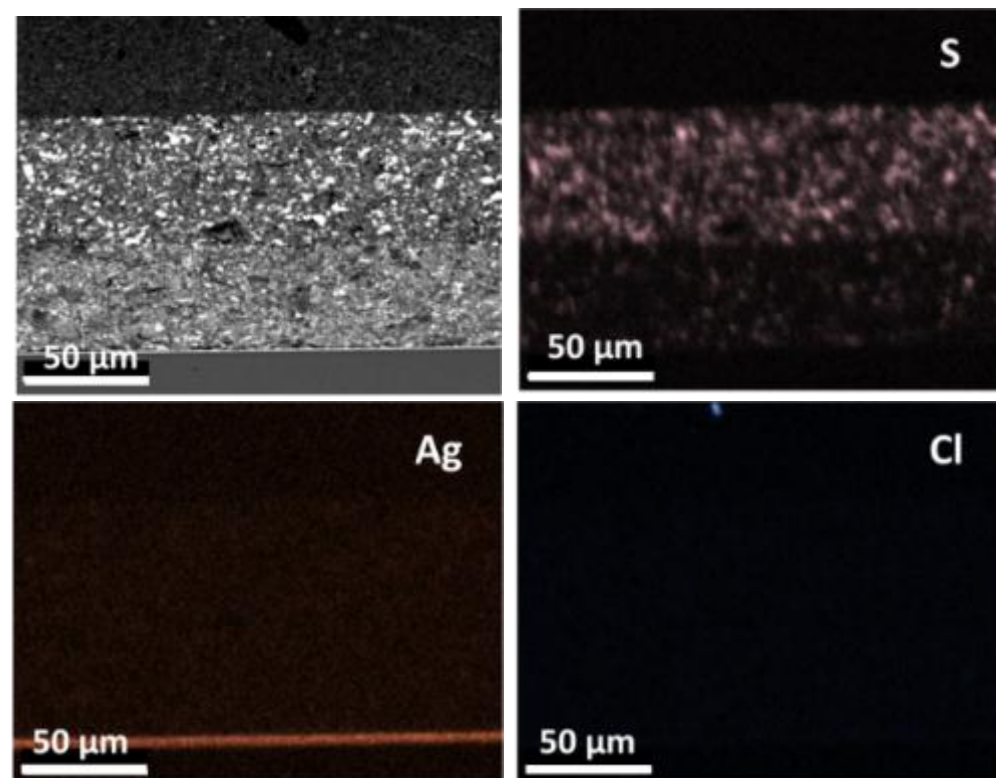


# Results – AR samples

## Initial Presence of contaminants – EDS mapping



Nishio\_AR



Ref\_AR

# Results – Exposed samples

## Visual inspection of newly exposed mirror (6 months)

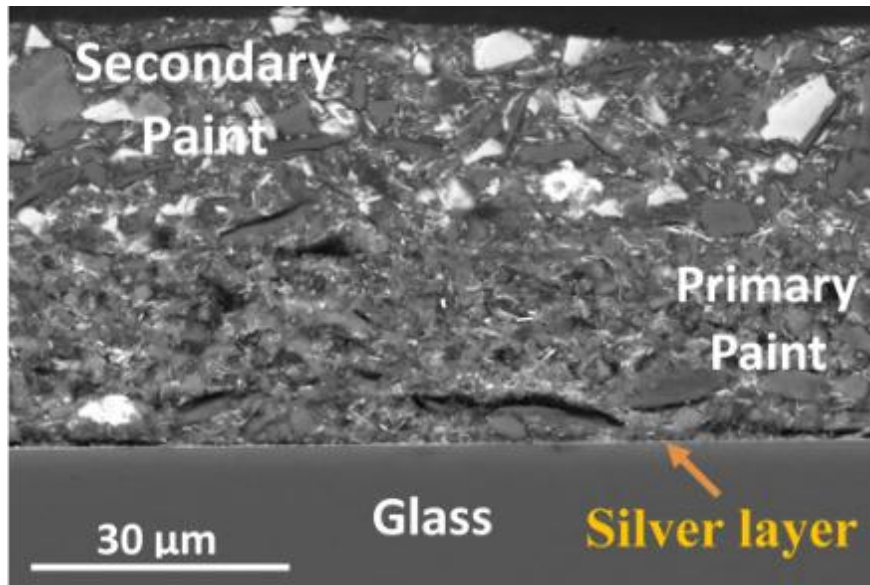
Backside protective resin detachment at the edges



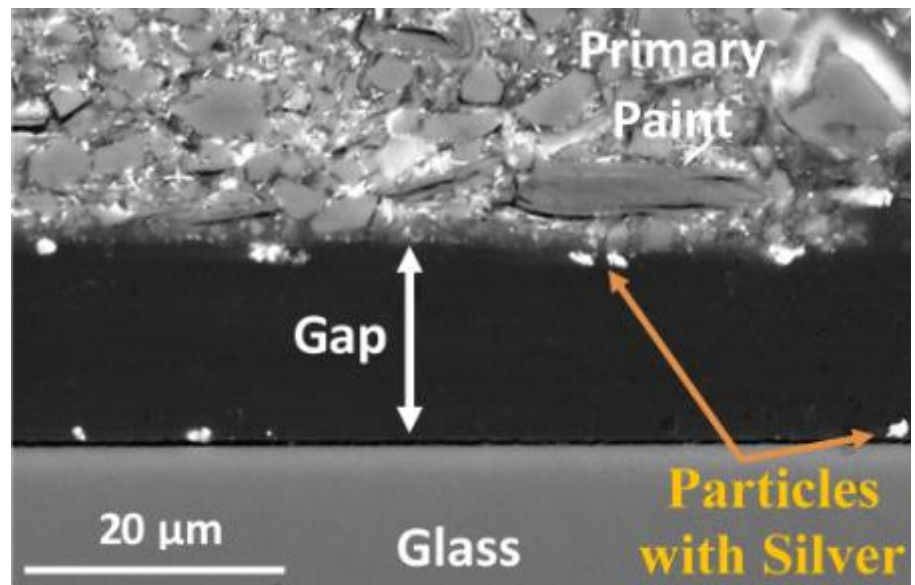
Edge corrosion of the silver layer

# Results – Exposed samples

## Comparison of Nishio\_Exp and with Nishio\_AR



Nishio\_AR

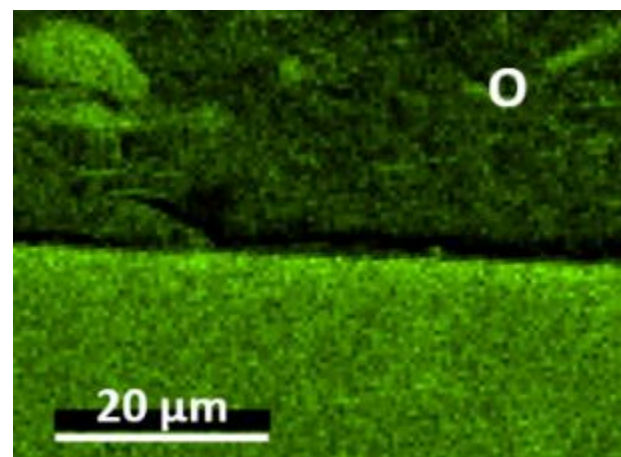
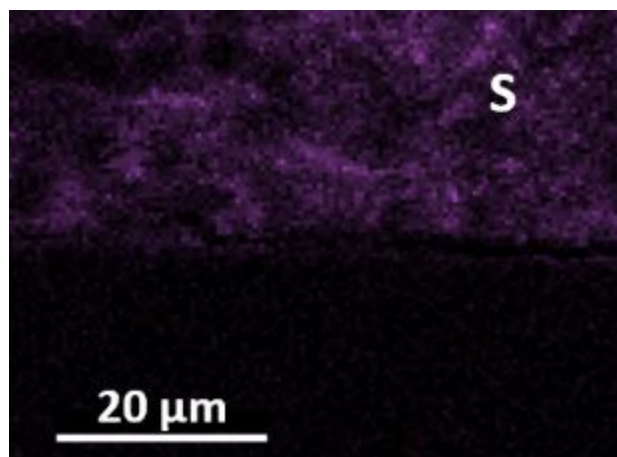
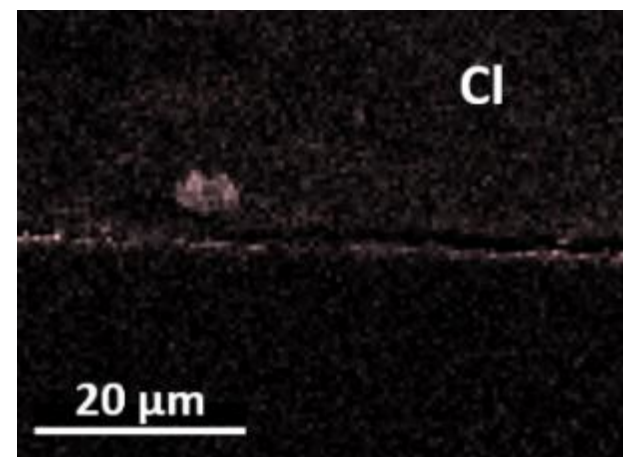
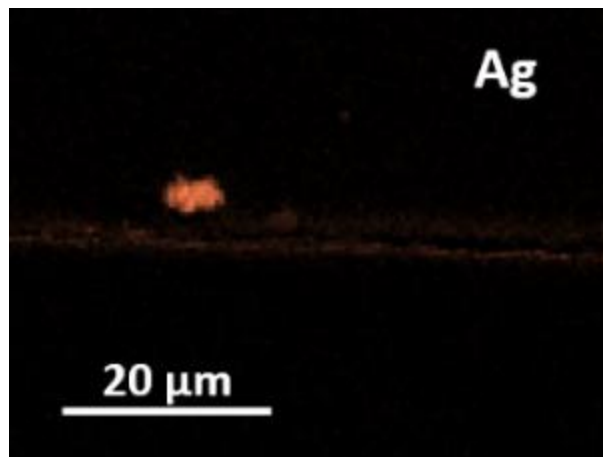
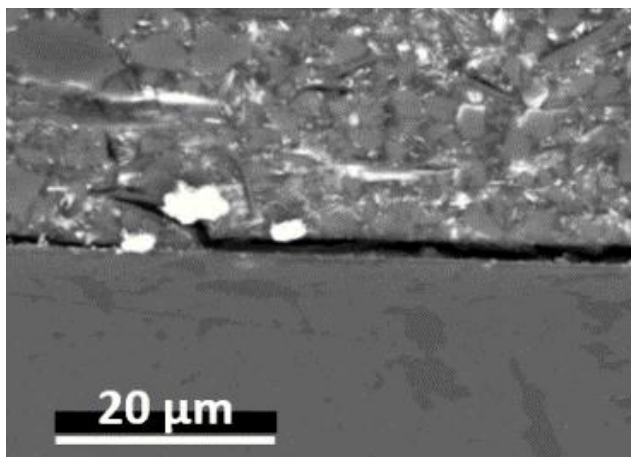


Nishio\_Exp



# Results – Exposed samples

## Analysis of Nishio\_Exp – EDS mapping



# Conclusions and future work

## Analysis of corroded and as-received samples

5 main possible causes of the rapid corrosion identified:

- 1) weak adhesion and manufacturing defects in reflecting layers
- 2) weak protective top coating, especially against UV
- 3) absence of copper layer
- 4) high initial presence of corrosive contaminant, chlorine, remaining from the manufacturing process
- 5) low mechanical stability of top coating especially at the edges.

## Future work:

- Evaluation of corrosivity of the exposure site: standard tests ISO 9223-9226
- Further combined accelerated aging tests



## Acknowledgments

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**Thank You**  
**Questions?**

