Modeling SEI Formation and Morphology

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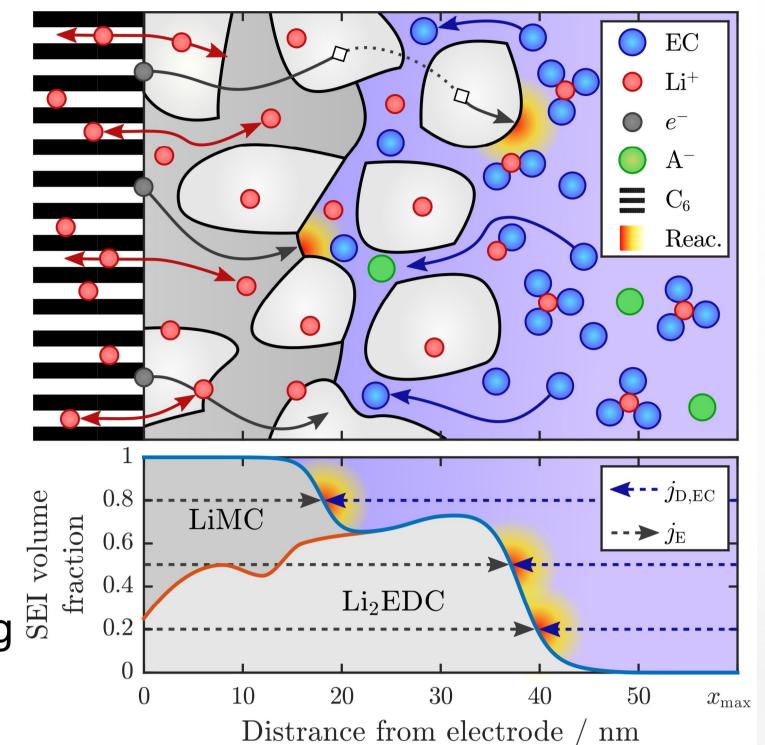
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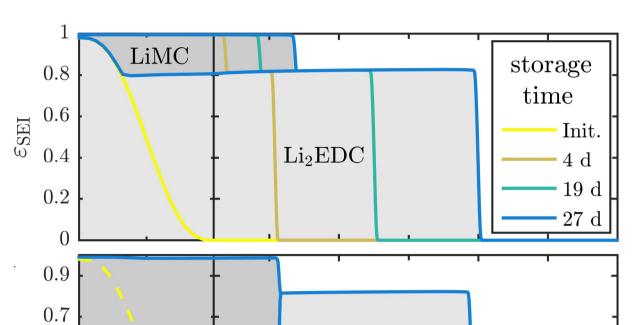
Introduction

- Modeling growth of nanoporous SEI
- 1D model, model axis perpendicular to electrode
- Transport of electrons and

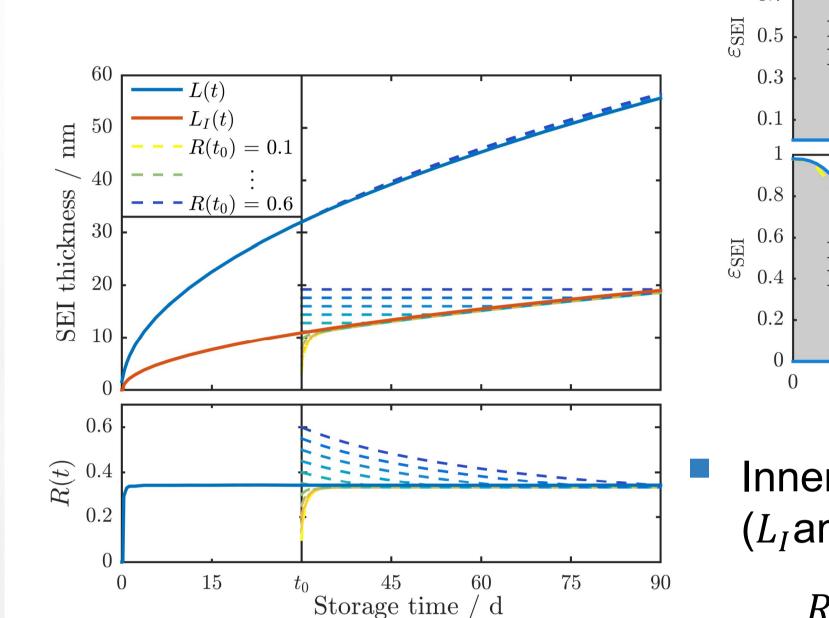


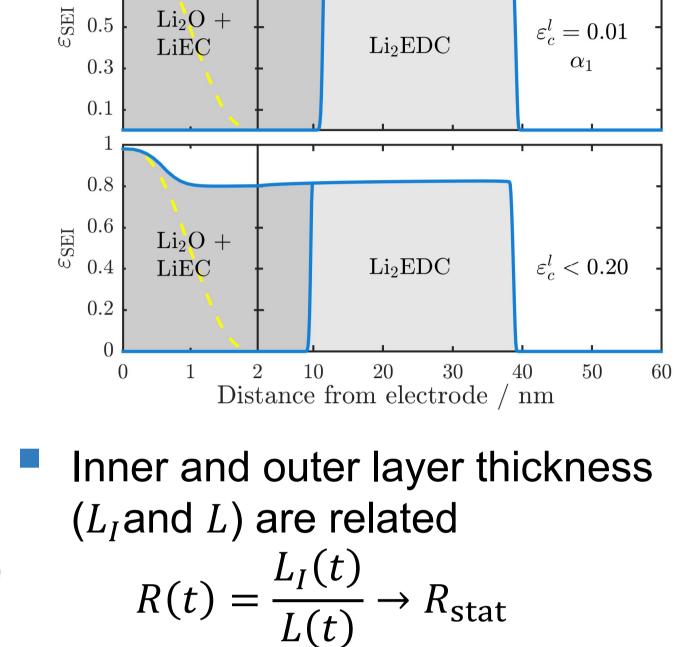
Simulations: Dual-Layer SEI

- Second reduction reaction
- Active co-solvent
- Conversion of Li₂EDC
- Dual-layer SEI



- solvent [1-3]
- **Electrons: solid SEI**
- Binary electrolyte in pores, EC/DMC 3:7
- Up to two electrolyte reduction reactions forming different SEI compounds





SEI Growth Model

SEI volume fraction evolution along an axis perpendicular to the electrode surface

- Solvent diffusion and convection $\partial_t(\varepsilon c_i) = -\operatorname{div}(j_{\mathrm{D},i} + j_{\mathrm{C},i})$ in the electrolyte phase (binary mixture)
- Electron conduction within the solid SEI phase

Convection velocity from incompressibility $(V_i^{solv}c_i = 1)$ $\partial_t \varepsilon_i = V_i \dot{s}_i - \operatorname{div}(\varepsilon_i \tilde{v})$

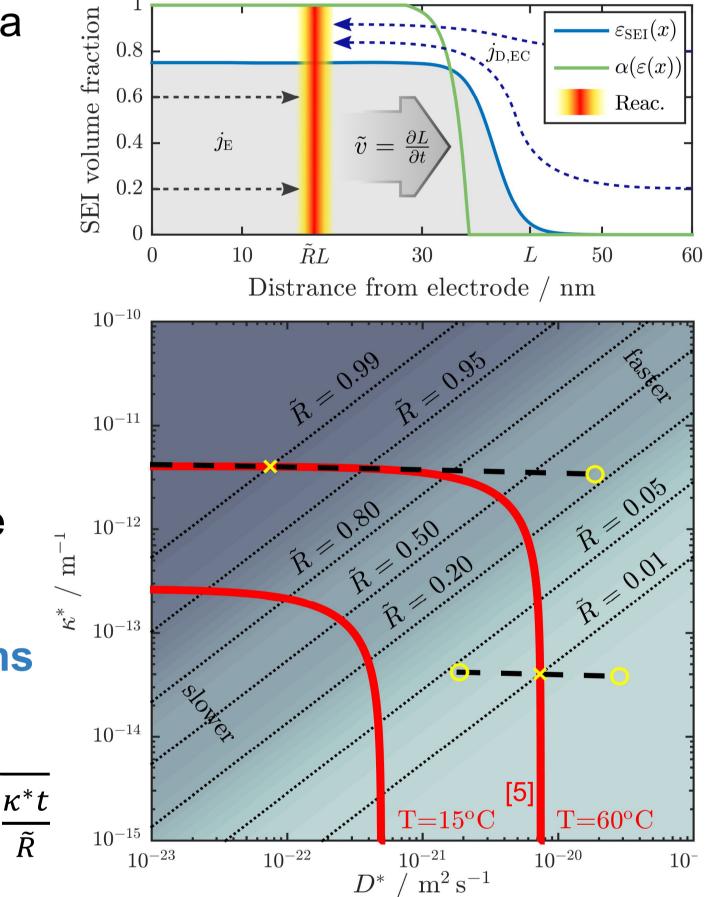
 $-v_i \dot{s}_i$

 $0 = -\operatorname{div}(j_{\rm E}) - F\dot{s}_i$

div $v = \sum (2V_i - v_i V_i^{\text{solv}}) \dot{s}_i$

Simulations: Self-Shaping SEI

- Assumption: SEI porosity is a material property
- Set porosity with $\alpha(\varepsilon)$ 0.8 0.6

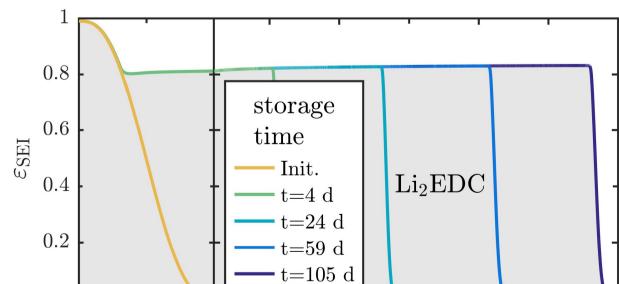


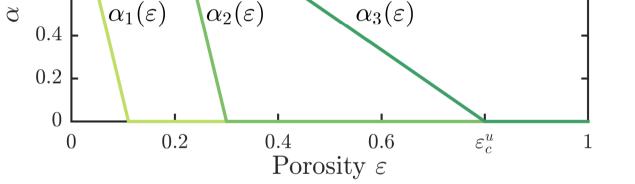
$\frac{1}{1} = \frac{1}{2}$	
Solid convection	div $\tilde{v} = \alpha(\varepsilon) \dot{\varepsilon}_{\text{SEI,C}} / \varepsilon_{\text{SEI}}$
Bruggeman relation	Flux densities
$D = \varepsilon^{\beta} D_{\text{Bulk}}$ $\kappa = (1 - \varepsilon)^{1.5} \kappa_{\text{Bulk}}$	$j_{\mathbf{C},i} = c_i v$ $j_{\mathbf{D},i} = -\mathbf{D} \cdot \text{grad } c_i$ $j_{\mathbf{E}} = -\mathbf{\kappa} \cdot \text{grad } \Phi$

Butler-Volmer:
$$\dot{s}_i = A(\varepsilon)\Gamma \frac{k_B T}{h} \exp\left(\frac{-E_A}{k_B T}\right) \left(c_i/c_i^0\right)^{\frac{\nu_i}{2}} 2\sinh\left(\frac{RT}{F}\eta_i\right)$$
 $\eta_i = -\left(\Phi - \Phi_i^0\right) + \nu_i \ln\left(\frac{c_i}{c_i^0}\right)$
 Specific surface area: $A(\varepsilon) = \frac{1}{a_0}\varepsilon\left(\varepsilon_{\text{SEI}} + \frac{a_0^2}{6}\frac{\partial^2\varepsilon}{\partial x^2}\right)$

Simulations: Inert Co-Solvent

- Formation of porous SEI
- constant SEI porosity $\varepsilon^* = 1 - \varepsilon^*_{\text{SEI}}$
- SEI thickness growth with \sqrt{t}



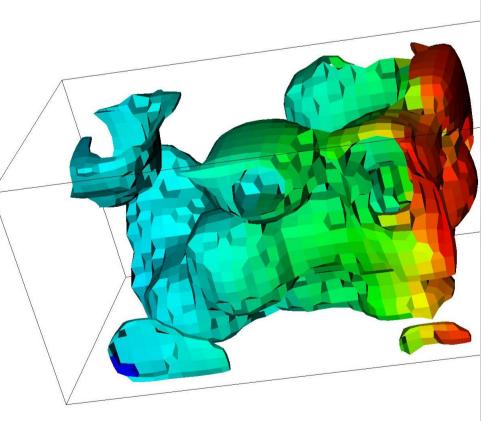


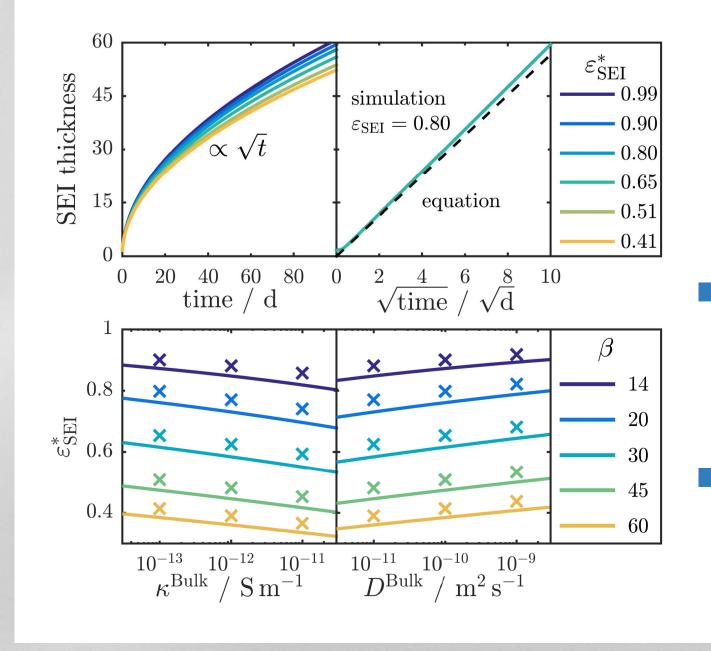
- Reaction interface inside the SEI, located at $x = \tilde{R} \cdot L$
- **Both transport mechanisms** contribute

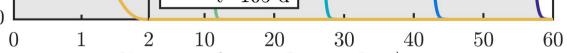
$$ilde{R} = \left(1 + \frac{D^*}{\kappa^*} \frac{Fc}{\Delta \Phi}\right)^{-1}, \ L(t) \propto \sqrt{\frac{P}{2}}$$

3D Electrode Simulation

- BEST: 3D transport simulation with porous electrodes [6]
- Simple SEI growth model on the graphite surface
 - Prediction of inhomogeneous SEI thickness
 - Understanding intercalation through SEI and lithium loss during cycling







 $\begin{array}{ccccccccc} 2 & 10 & 20 & 30 & 40 \\ \text{distance from electrode} & / & \text{nm} \end{array}$

- $L(t) \approx \sqrt{V_1 \kappa^* \Delta \Phi_1 / 2F \varepsilon_{\text{SEI}}^* \sqrt{t}}$
- Electronic conductivity from fit to experiment [1,2]

 $\sigma_{\rm Bulk} = 0.3 - 4.5 \, {\rm pS/m}$

Analytic expression for porosity $=\frac{F^2c_1^0}{PT}\left(\frac{1}{2}+\beta\frac{1-\varepsilon^*}{\varepsilon^*}\right)$

References

[1] Pinson, M.B., Bazant, M.Z. J. Electrochem. Soc. 160, A243-A250 (2012). [2] Christensen, J., Newman, J. Electrochem. Soc. 151 (11), A1977 (2004). [3] Jorn, R et al. J. Phys. Chem. C, **117** (8), 3747-3761. (2013) [4] Shi, S. et al. J. Am. Chem. Soc. **134** (37), 15476-15487 (2012). [5] Liu, P., Wang et al. *J. Electrochem. Soc.*, **156**, A499, (2010). [6] ITWM, BEST - Battery and Electrochemistry Simulation Tool, http://itwm.fraunhofer.de/BEST

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