Phase-field study of the pattern formation in Al-Ag-Cu under the influence of the melt concentration

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Abstract

In single transverse sections of directional solidified ternary eutectics, various microstructure patterns can be observed. These patterns influence the mechanical properties and it is therefore of interest to gain a better understanding of the microstructure formation process. It is assumed that local variations in the concentration of the melt lead to different patterns. To investigate this effect, large-scale three-dimensional phase-field simulations of directional solidification in the vicinity of the ternary eutectic point of Al-Ag-Cu are applied. The different arising patterns from the simulations are compared and analyzed with statistical methods. The simulations show different stable patterns within a range of \(\pm 1\%\) of the melt concentration around the ternary eutectic point. The same tendencies are observed in experimental micrographs.

Keywords: directional solidification, phase-field, ternary eutectic alloy, Al-Ag-Cu, pattern selection, melt concentration

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1. Introduction

Different microstructures evolve, during the solidification of multicomponent systems. These microstructures influence the mechanical properties of the macroscopic component [1–3]. Especially in the vicinity of a ternary eutectic point a wide range of patterns in the microstructure form during directional solidification.

Five theoretical arrangements of the three solid phases in ternary eutectic systems are predicted from geometrical assumptions in [4] and are graphically depicted in [5]. In the experimental study of [6], the pattern formation in various ternary eutectic systems is presented. Several of the theoretically predicted arrangements are found in the evolving patterns of the ternary eutectic system Al-Ag-Cu. The influence of the growth rate on the evolving patterns is investigated in [7–9] and the crystallographic orientation of the patterns is studied with EBSD in [10]. Statistical analyses to classify the phase arrangements are conducted with nearest neighbor statistics [11], with shape factors [12], polar plots [13] as well as with Fourier analyses [14]. Six different, experimentally observed patterns in the system Al-Ag-Cu are characterized and investigated in [15]. First three-dimensional representations of the microstructure in Al-Ag-Cu, obtained from synchrotron tomography, are reported in [16]. A thermodynamic assessment is applied in [17, 18].

Simulative research of the pattern formation in three-phase ternary eutectics with the phase-field method is conducted in 3D for idealized systems in [19–23] and for Al-Ag-Cu in [23–26]. In idealized systems of directionally solidified ternary eutectics, various patterns are found in [20] and the influence of the concentration of the melt and the solid-liquid interface energies on the pattern formation is identified in [23]. With large-scale simulations, visual [25] and quantitative [26] accordance between phase-field simulations and experiments can be achieved. The necessity of large-scale simulations to obtain statistical volume elements (SVE) is presented in [26] with principal component analysis based on two-point correlations. In [22] different alignments of a hexagonal pat-
tern in single transverse sections of an ideal system are reported for large-scale simulations.

Beside the different alignments, also various patterns are found within a single micrograph parallel to the solidification front as depicted in fig. 1 for the system Al-Ag-Cu. In the upper right corner, paw structures and in the lower left part, chain-like structures evolve in a concurrent manner. It is assumed, that these structures are separated by a grain boundary. The sample solidified under uniformly imprinted process conditions with a growth velocity of 0.32 $\mu m/s$ and a gradient of 2.8 $K/mm$.

![Figure 1: Experimental transverse section of a directionally solidified microstructure in the ternary eutectic system Al-Ag-Cu, with a velocity of 0.32 $\mu m/s$ and a gradient of 2.8 $K/mm$. Three solid phases $Ag_2Al$ (white), $Al_2Cu$ (gray) and $Al$ (black) can be distinguished. In the upper right corner, paw structures and in the lower left part, chain-like structures are observed and exemplary extracted as magnified images. It is assumed, that these structures are separated by a grain boundary.]

To investigate the simultaneous formation of different patterns under uniformly imprinted process conditions, the effect of various melt concentrations is studied. Therefore, we apply large-scale three-dimensional phase-field simulations based on the Grand potential approach and systematically vary the concentration of the melt in the vicinity of the ternary eutectic point. To quantify the evolving patterns, the phase fractions, nearest neighbor statistics and principal component analysis, based on two-point correlations are applied.
2. Model

For the simulations, a thermodynamically consistent phase-field model is used, derived from a Grand potential functional and Allen-Cahn type variation \cite{25, 27, 28}. For a three-phase ternary eutectic system, the $N = 4$ order parameters $\phi_\alpha$, describe the local phase fractions. From the mass balance of the concentrations and Fick’s law, the $K = 3$ chemical potentials in $\mu$ are derived. Coupling the $N$ phase-fields, the $K$ chemical potentials and the imprinted temperature $T$, results in the following set of evolution equations:

\begin{equation}
\tau \epsilon \frac{\partial \phi_\alpha}{\partial t} = -\epsilon T \left( \frac{\partial a(\phi, \nabla \phi)}{\partial \phi_\alpha} - \nabla \cdot \left( \frac{\partial a(\phi, \nabla \phi)}{\partial \phi_\alpha} \right) \right) - \frac{1}{\epsilon} \frac{\partial \omega(\phi)}{\partial \phi_\alpha} - \sum_{\beta=1}^{N} \psi_\beta(\mu, T) \frac{\partial h_\beta(\phi)}{\partial \phi_\alpha} - \frac{1}{N} \sum_{\beta=1}^{N} \rho s_\beta, \tag{1}
\end{equation}

\begin{equation}
\frac{\partial \mu}{\partial t} = \left[ \sum_{\alpha=1}^{N} h_\alpha(\phi) \left( \frac{\partial c_\alpha(\mu, T)}{\partial \mu} \right) \right]^{-1} \left( \nabla \cdot \left( M(\phi, \mu, T) \nabla \mu - J_{\text{kt}}(\phi, \mu, T) \right) \right) - \sum_{\alpha=1}^{N} c_\alpha(\mu, T) \frac{\partial h_\alpha(\phi)}{\partial t} - \sum_{\alpha=1}^{N} h_\alpha(\phi) \left( \frac{\partial c_\alpha(\mu, T)}{\partial T} \right) \frac{\partial T}{\partial t}, \tag{2}
\end{equation}

\begin{equation}
\frac{\partial T}{\partial t} = \frac{\partial}{\partial t} (T_0 + G(z - vt)) = -Gv. \tag{3}
\end{equation}

The relaxation parameter $\tau$ is introduced, to couple the different timescales of the evolution equations. The shape of the diffuse interface between the phases is modeled by the gradient energy $a$, by the obstacle potential $\omega$ and $\epsilon$, which controls the interface width. The driving force for the phase transitions is described by the differences of the Grand potentials $\psi_\beta$, which are calculated from the Gibbs energies of the different phases. The Gibbs energies are incorporated from thermodynamic CALPHAD databases \cite{17, 18} and, to optimize the computational effort, fitted by a parabolic approach of the form:

\begin{equation}
G_\alpha(c, T) = \langle c, \Xi_\alpha(T) c \rangle + \langle c, \xi_\alpha(T) \rangle + X_\alpha(T), \tag{4}
\end{equation}

with the matrix $\Xi_\alpha(T)$, the vector $\xi_\alpha(T)$, the scalar $X_\alpha(T)$ and the scalar product $\langle \cdot, \cdot \rangle$ \cite{29}. The evolution equation of the chemical potentials is derived in
including the mobility term $M$, the anti-trapping current $J_{at}$ \cite{28,30,31} and the interpolation function $h_\alpha$. Starting from an initially imprinted temperature field with the base temperature $T_0$, the temperature $T$ evolves with the gradient $G$ and the velocity $v$ in the growth direction $z$. The partial differential equations (PDEs) \cite{1,2}, are spatially discretized with finite differences and the temporal evolution is calculated with a forward Euler scheme \cite{32}. These PDEs are implemented in the massive parallel framework \textsc{walBerla} \cite{33}. A detailed description of the model is presented in \cite{25}.

3. Setup

In the following, the simulation setup and the applied parameters for the concentration variations are introduced.

3.1. Simulation setting

The setup for the simulations of the ternary eutectic directional solidification is schematically depicted in fig. 2. The simulation domain has a base size of $800 \times 800$ voxel cells. To obtain statistical volume elements these large-scale simulations are required to minimize the influence of the periodic boundary conditions, as shown in \cite{22,26}. Starting from an initial Voronoi tesselation, to model the nucleation, three solid phases cooperatively grow in $z$-direction with a defined velocity, controlled by the temperature gradient.

3.2. Parameters

The system Al-Ag-Cu exhibits a ternary eutectic point $E_{cal}$ at the mol-fractions 0.181, 0.691 and 0.128 ($\text{Ag,Al,Cu}$) as depicted in the liquidus projection in fig. 3 based on the CALPHAD database of \cite{17,18}. At a temperature 20 K below the eutectic point $E_{cal}$, the solubility of Ag in the Al-phase has changed from 16.9\% to 8\%, leading to different phase fractions at room temperature. This solubility shift is described in \cite{11,17,18}. To reproduce the phase fractions reported from micrographs, the ternary eutectic point is shifted to $E_{Exp}$ with the mol-fractions 0.237, 0.622, 0.141 (Ag,Al,Cu) analogues to \cite{23,25}. This
Figure 2: Schematically depiction of the setting applied in the simulations.

allows to reduce the computational effort significantly.

According to [17, 18], the parameters for the shifted and adjusted parabolic Gibbs energies $\Xi_\alpha$, $\xi_\alpha$, $X_\alpha$ as well as the interface energies $\gamma$, the model-intrinsic higher order term $\gamma_{\alpha\beta\delta}$ to suppress the unphysical evolving of third-phases [19, 34, 35], the diffusion coefficient $D$ and the process parameters $\nabla T$ and $v$ are listed in table 1. The numerical parameters are given in table 2.

In the simulations, the concentration of the melt $c_\ell$ is systematically varied in the vicinity of the ternary eutectic composition at $E_{Exp}$, following the three axis parallel to the edges of the ternary simplex. One concentration is kept constant and the other two are changed by $\pm 0.5\%$ and $\pm 1\%$ respectively. The changes of $\pm 0.5\%$ are indicated by small labels a-f and of $\pm 1\%$ by capital letters A-F, as highlighted by the enlargement in fig. 3.

4. Results

All simulations are conducted with $800 \times 800$ voxels base size and 3 million time steps, corresponding to a growth height of approximately 6300 cells. They
Table 1: Summary of the physical parameters used in the simulations. The concentrations of the melt are varied by ±0.5% and ±1% respectively, according to fig. 3.

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Simulation value</th>
<th>Physical value</th>
</tr>
</thead>
<tbody>
<tr>
<td>( \gamma )</td>
<td></td>
<td></td>
</tr>
<tr>
<td>phase</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Al</td>
<td>- 0.5 0.4 0.2</td>
<td></td>
</tr>
<tr>
<td>( A_{2}Al )</td>
<td>0.5 - 0.4 0.4</td>
<td>scaling factor for physical values: 0.2 J/m(^2) based on [36, 39]</td>
</tr>
<tr>
<td>( A_{2}Cu )</td>
<td>0.4 0.4 - 0.4</td>
<td></td>
</tr>
<tr>
<td>( \text{liquid} )</td>
<td>0.3 0.4 0.4 -</td>
<td></td>
</tr>
<tr>
<td>( \gamma_{\alpha\beta\delta} )</td>
<td>6.5</td>
<td>2.6 J/m(^2)</td>
</tr>
<tr>
<td>( D )</td>
<td>5.0</td>
<td>7.5 ( \times ) 10(^{-10}) m(^2)/s</td>
</tr>
<tr>
<td>( c_{\ell} ) at ( E_{Exp} )</td>
<td>0.237, 0.622, 0.141</td>
<td>mol-fraction (Ag, Al, Cu)</td>
</tr>
<tr>
<td>( \Xi_{Al,\xi_{Al},X_{Al}} )</td>
<td>[2 1 -2.25 1.561866]</td>
<td>based on [17, 18]</td>
</tr>
<tr>
<td>( \Xi_{A_{2}Al,\xi_{A_{2}Al},X_{A_{2}Al}} )</td>
<td>[2 1 -3.104 1.427846]</td>
<td>based on [17, 18]</td>
</tr>
<tr>
<td>( \Xi_{A_{2}Cu,\xi_{A_{2}Cu},X_{A_{2}Cu}} )</td>
<td>[2 1 -1.395 0.9176]</td>
<td>based on [17, 18]</td>
</tr>
<tr>
<td>( \Xi_{\ell,\xi_{\ell},X_{\ell}} )</td>
<td>[2 1 -2.220 3.8655 - 2.5T]</td>
<td>based on [17, 18]</td>
</tr>
<tr>
<td>( T ) at ( E_{Exp} )</td>
<td>1.0</td>
<td>773.6 K</td>
</tr>
<tr>
<td>( \nabla T )</td>
<td>( 10^{-4} )</td>
<td>200 K/mm</td>
</tr>
<tr>
<td>( v ) of ( \nabla T )</td>
<td>2.1 ( \times ) 10(^{-3}) cells per time step</td>
<td>0.25 ( \mu )m/s</td>
</tr>
</tbody>
</table>

Table 2: Summary of the numerical parameters of the simulations.

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Simulation value</th>
<th>Physical value</th>
</tr>
</thead>
<tbody>
<tr>
<td>dx</td>
<td>1.0</td>
<td>3.89 ( \times ) 10(^{-7}) m</td>
</tr>
<tr>
<td>dt</td>
<td>3.2</td>
<td>3.2 ( \times ) 10(^{-3}) s</td>
</tr>
<tr>
<td>( \varepsilon )</td>
<td>4.0</td>
<td>1.56 ( \times ) 10(^{-6}) m</td>
</tr>
</tbody>
</table>
Figure 3: Liquidus projection of the ternary eutectic system Al-Ag-Cu, based on [17, 18]. In red the adjusted ternary eutectic point and the adjusted tie lines are depicted to reproduce the experimentally observed phase fractions, analogously to [23, 25]. The right image magnifies the ternary eutectic region and marks the simulated concentrations a-f, A-F and $E_{Exp}$.

are each executed with 11 200 cores for approximately 12 h on the SuperMUC cluster at the Leibniz supercomputing centre Munich. For all simulations, the parameter sets in tables 1 and 2 are applied. In the simulation results, the Al phase is depicted in red, the Ag$_2$Al phase in green and the Al$_2$Cu phase in blue.

4.1. Variation of the melt concentration

In fig. 4 the solidification fronts at the end of the simulations for the varied concentrations of the melt are shown in their arrangement around the ternary eutectic point $E_{Exp}$. Depending on the variations, different patterns of the three solid phases evolve. An increasing distance of the concentration $c_l$ from $E_{Exp}$ leads to a more pronounced deviation of the arising patterns. A further change of the melt composition results in two-phase growth. To quantify the visual observation of the pattern modification, the phase fractions and their relative deviation from $E_{Exp}$ (table 3) as well as nearest neighbor statistics for Al$_2$Cu around Ag$_2$Al (fig. 5(a)) and vice versa (fig. 5(b)) are collected.
Figure 4: The solidification fronts after 3 million time steps with different concentrations of the melt in their arrangement around the ternary eutectic point $E_{EFP}$. The $Al$ phase is depicted in red, the $Ag_2Al$ phase in green and the $Al_2Cu$ phase in blue. One concentration is kept constant and the other two are changed by \( \pm 0.5\% \) (a-f) and \( \pm 1\% \) (A-F) respectively.
For the quantitative analysis of the simulation results, we focus on the simulations with ±1% deviation from $E_{\text{Exp}}$, indicated by the capital letters A-F, to improve the clarity of the results. However, all observed outcomes are also qualitatively valid for the corresponding simulations a-f. At the ternary eutectic concentration at $E_{\text{Exp}}$, the phase fractions of Al, $Ag_2Al$ and $Al_2Cu$ are 35%, 34% and 31%. Chain-like structures of $Al_2Cu$ and $Ag_2Al$ phase arrangements, embedded in an Al matrix evolve, similar to the structures in the left and lower part of the micrograph in fig. 1. Chains appear as two neighboring rods in the nearest neighbor statistics in fig. 5 for $Ag_2Al$ around $Al_2Cu$ and vice versa.

The simulations at the compositions A, E and F lead to an increase of the Al phase fraction, incorporated by decreasing fractions of the $Al_2Cu$ and $Ag_2Al$ phases. A more pronounced matrix phase with shorter embedded chains, so-called island structures [25], and paw structures [15] evolve. In the simulations, the paw structures occur as $Ag_2Al$ rods attached to $Al_2Cu$ lamellae and lead to three and more $Ag_2Al$ neighbors for each $Al_2Cu$ lamella as shown in fig. 5(b). Also the number of three and more neighbors in fig. 5(b), respectively the number of one neighbor in fig. 5(a) reflects the chain ends. In the simulations A and F the observed pattern changes are caused by an increase of the Al amount in the melt. Simulation E is characterized by a decrease of Cu and an increase of Ag, which leads to an enrichment of Al in the melt, due to the different amount of Al in the phases $Ag_2Al$ and $Al_2Cu$.

The shifts of the concentrations towards the $Al_2Cu$ phase in B and C, result in a larger fraction of this phase and branched chains, as well as so-called ring-like structures. The junctions, consisting of $Al_2Cu$ rods, are reflected as three neighbors in fig. 5(b). Analogue to the simulation at $E_{\text{Exp}}$, the chains are expressed by the percentage of 77% in B and 80% in C for two $Al_2Cu$ neighbors around $Ag_2Al$ in the nearest neighbor statistic of fig. 5(a).

In D a larger amount of Ag and a constant concentration of Cu is simulated and results in a larger phase fraction of $Ag_2Al$. The nearest neighbor statistics as well as the evolving chain-like structures are similar to the simulation at $E_{\text{Exp}}$. Despite the constant amount of Al in the melt of the simulations B and E, the
phase fractions of the Al phase change. Due to the different compositions of the solid phases and their interactions, it is not possible to predict the arising patterns just from the overall concentration of the melt. The observed results for the simulations A-F also occur less pronounced in the corresponding simulations a-f.

Table 3: Phase fractions of the simulations A-F and \( E_{Exp} \).

<table>
<thead>
<tr>
<th>Simulation</th>
<th>Al rel. deviation</th>
<th>( Ag_2Al ) rel. deviation</th>
<th>( Al_2Cu ) rel. deviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>( E_{Exp} )</td>
<td>0.350</td>
<td>0.309</td>
<td>0.341</td>
</tr>
<tr>
<td>A</td>
<td>0.407</td>
<td>0.264</td>
<td>0.329</td>
</tr>
<tr>
<td>B</td>
<td>0.324</td>
<td>0.292</td>
<td>0.384</td>
</tr>
<tr>
<td>C</td>
<td>0.286</td>
<td>0.330</td>
<td>0.384</td>
</tr>
<tr>
<td>D</td>
<td>0.325</td>
<td>0.343</td>
<td>0.332</td>
</tr>
<tr>
<td>E</td>
<td>0.405</td>
<td>0.316</td>
<td>0.279</td>
</tr>
<tr>
<td>F</td>
<td>0.445</td>
<td>0.277</td>
<td>0.278</td>
</tr>
</tbody>
</table>

Figure 5: Nearest neighbour statistics of the phase \( Al_2Cu \) adjacent to \( Ag_2Al \) (a) and vice versa (b).

The study indicates, that a variation of the melt concentration of \( \pm 0.5\% \) and \( \pm 1\% \) from the ternary eutectic composition results in visual and measurable pat-
tern change. In fig. 6 the three-dimensional microstructures for the simulations A-F are depicted. The lower 2D transverse sections exempt the solidification fronts after one and two millions time steps. Between one and two million time steps only small changes of the microstructure occur. In the following one million iterations, till the end of the simulations, the phase arrangements and the form of the rods grow in stationary states. We conclude that a variation of the concentration in the melt leads to the growth of several, stable patterns.

4.2. Approach to an explanation of complex pattern formation in experiments

During directional solidification experiments of ternary eutectic systems, like Al-Ag-Cu, various patterns evolve [6, 11, 15]. Even in a single micrograph, different patterns can be observed as exemplary shown in fig. 1 and in fig. 7. In the experimental micrographs of Al-Ag-Cu in fig. 7, three regions I-III can be distinguished, which probably consist of different grains. In each of these regions, different phase fractions and patterns occur. On the left side, a general view of the directional solidified sample and on the right side, a detailed micrograph of the region with three different patterns is depicted.

In table 4, the phase fractions and their relative deviation from $E_{Exp}$ in table 3 are listed. The phase fractions in region I of table 4 differ less than 1% for all phases from the simulated results at $E_{Exp}$. Both nearest neighbor statistics in fig. 8 show a peak at two neighbors for $E_{Exp}$ as well as for region I. Based on these statistics as well as the occurring microstructure, the experimentally observed results in region I are in accordance with the simulation at $E_{Exp}$. The results of region II show a similar tendency as in simulation D, which corresponds to an increase of Ag in the melt concentration. The deviation from the simulation at $E_{Exp}$ is more pronounced in the experimental region II, than in the simulation D. In region III, the phase fractions deviation show a trend similar to simulation B. In all regions I-III, variations of chain-like structures can be observed visually as well as with nearest neighbor statistics.

To further quantify the observed variations of the microstructure, a principal component analysis (PCA), based on two-point correlations is conducted [40].
Figure 6: The three-dimensional microstructure of the three solid phases for the simulations A-F are shown. Below, the solidification fronts after 1 and 2 million time steps are extracted.
Figure 7: Directional solidified experimental transverse sections of Al-Ag-Cu parallel to the growth front. The sample solidified with a velocity of 0.11 μm/s and a gradient of 2.2 K/mm. The composition of the melt was measured to be at the ternary eutectic point. Left, a general view of the directional solidified sample and right, a detailed micrograph of the region with three different patterns is depicted.

The two-point correlations describe the probability that two points on a plane have defined characteristics, depending on their relative positions. The set of all two-point correlations is projected in the direction of the highest variances, which generates the highly non-linear principal component (PC) space. The capability of this method to analyze microstructure evolutions is demonstrated in [26]. The same procedure as in [26] is applied to the simulations a-f, A-F and $E_{Exp}$ as well as to the different regions I-III of the experimental micrograph. In fig. 9 the projection of the results in the three PC directions with the highest variance are plotted.

In both projections, the arrangement and the order of the simulations A-F around $E_{Exp}$ is reflected. Similar to the visually observation and the measured phase fractions, the PCA shows the accordance between the simulation $E_{Exp}$.
Table 4: Phase fractions at the different regions of fig. I-III and their relative deviation from the simulation at $E_{Exp}$.

<table>
<thead>
<tr>
<th></th>
<th>Al rel. deviation</th>
<th>$Ag_2Al$ rel. deviation</th>
<th>$Al_2Cu$ rel. deviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>I</td>
<td>0.347 -0.01</td>
<td>0.303 -0.02</td>
<td>0.35 0.03</td>
</tr>
<tr>
<td>II</td>
<td>0.306 -0.13</td>
<td>0.479 0.55</td>
<td>0.215 -0.37</td>
</tr>
<tr>
<td>III</td>
<td>0.348 -0.01</td>
<td>0.273 -0.12</td>
<td>0.379 0.11</td>
</tr>
</tbody>
</table>

Figure 8: Nearest neighbour statistics of the phase $Al_2Cu$ adjacent to $Ag_2Al$ (a) and vice versa (b) for the different regions of fig. I-III and selected simulations.

and region I, due their relative location. Like $E_{Exp}$, the projection of region I is framed by the representations of the simulations A-F.

The PCA results for region II and III are in accordance with the tendencies of the previous statistics. Based on $E_{Exp}$, in both plots the projection of region II is located in the direction of the simulations D and E, whereas the projection of region III lies in the direction of the simulations B and C. It is shown that the pattern change as well as the shift of the phase fraction is reflected with the principal component analysis.

Due to the mentioned tendencies from various analyses of the regions I-III and of the simulations with different melt concentrations, we suppose, that the variations of the chain-like patterns depend on local concentration deviations. One
Figure 9: Projection of the simulation at $E_{Exp}$, the simulations A-F and three different regions I-III from the experimental micrograph in the PC1 and PC2 plane as well as the PC1 and PC3 plane.

Explanation of these local concentration deviations in fig. 7 is gravity induced segregation due to the different densities of the chemical elements in the melt [49, 50]. It is assumed that this density differences lead to convection cells in the melt. To study the effect of gravity induced segregation on the microstructure evolution, either low gravity experiments or the incorporation of convection in the phase-field model is necessary. Experimental studies on the international space station are planned as part of the SETA project [51]. The coupled simulation of ternary directional solidification and convection requires a high computational effort [52, 53] and is focus of further work.

5. Conclusion

In this work, the pattern formation during the directional solidification of the ternary eutectic system Al-Ag-Cu is investigated. The concentration of the melt is systematically varied around the ternary eutectic point $E_{Exp}$ in the range of ± 1%. For this, large-scale phase-field simulations are conducted to gain statistical volume elements.
To quantify the evolving patterns, nearest neighbor statistics, the phase fractions and principal component analysis (PCA) based on two-point correlations are applied. Based on these we draw our five main conclusions: (i) Variations of the melt concentration of $\pm 1\%$ lead to different patterns. (ii) All simulations evolve in a stationary state for the different arising patterns. (iii) In the investigated range, the arising patterns at $\pm 1\%$ show similar, but enhanced tendencies as the corresponding simulations at $\pm 0.5\%$. (iv) The results from the PCA reflect the arrangement of the concentration variation around the ternary eutectic point and allow to determine a quantitative relation between different micrographs. (v) In the presented experimental micrograph, the same tendencies of the patterns and phase fractions like in the simulative study with different melt concentrations are observed. Following these results, we conclude, that local deviations in the concentration of the melt can result in different patterns within the same micrograph as experimentally observed.

6. Acknowledgements

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References


